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Development and evaluation of a fluorescence emission ratio-based fiber optic $\mathbf{p H}$ measurement system for use in monitoring changes in tumor pH during clinical hyperthermia

McCarthy, John Francis, Ph.D.

University of Ilinois at Urbana-Champaign, 1989

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BY<br>JOHN FRANCIS MCCARTHY<br>B. A., Boston University,<br>1976<br>M. S., University of Connecticut, 1978

## THESIS

Submitted in partial fulfillment of the requirements for the degree of Doctor of Philosophy in Biophysics in the Graduate College of the
University of Illinois at Urbana-Champaign, 1989

Urbana, Illinois

# UNIVERSITY OF ILLINOIS AT URBANA-CHAMPAIGN <br> THE GRADUATE COLLEGE 

WE HEREBY RECOMMEND THAT THE THESIS BY

JOHN FRANCIS MCCARTHY

ENTITLED DEVELOPMENT AND EVALUATION OF A FLUORESCENCE EMISSION RATIO BASED FIBER OPTIC pH MEASUREMENT SYSTEM FOR USE IN MONITORING CHANGES IN TUMOR pH DURING CLINICAL HYPERTHERMIA

BE ACCEPTED IN PARTIAL FULFILLMENT OF THE REQUIREMENTS FOR THE DEGREE OF_ DOCTOR OF PHILOSOPHY


Committee on Final Examination $\dagger$

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#### Abstract

The pH of the tumor microenvironment may be important in assessment of response to hyperthermic therapy. Little clinical in vivo data is available during such therapy due to the inherent limitations of the microelectrode technique in the presence of a microwave field. A fiber optic pH measurement system, due to its dielectric nature, provides a method for overcoming such limitations. Optrodes sensitive to pH have been studied based on absorption and fluorescence. Absorption based optrodes are difficult to fabricate due to the complex nature of the required optical geometry. Fluorescence based optrodes have been developed based upon either a single emission intensity measurement at a specific wavelength or the ratio of two emission intensities, at the same wavelength, following sequential excitation. Single intensity measurements are prone to substantial errors introduced by differences in the ionic strength and temperature of the samples as well as by fluorophore leakage, photobleaching, and fluctuations in the intensity of the excitation source. The ratio technique minimizes the above sources of measurement error, but the need for a sequential excitation results in instrumentation that may be too complex and expensive for routine use. Using the dual emission pH sensitive fluorophore 1,4-dihydroxyphalonitrile (1,4-DHPN), a simple ratio based optical pH measurement system was constructed. Optrodes were fabricated using glass capillary tubes $<1 \mathrm{~mm}$ in diameter with a Cuprophan membrane fixed at one end. The 1,4-DHPN was encapsulated in 4:1 DPPC/DPPG containing LUV in order to limit fluorophore loss and extend the sensor lifetime. A 2 mole percent quantity of gramicidin was added to the lipid phase, during preparation of the LUV, in order to insure rapid equilibration of hydrogen ions across the lipid bilayer. A flashlamp excitation source was used in conjunction with a single optical fiber to excite the fluorophore and to collect its fluorescence. Emission wavelengths of 488 and


434 nm were detected using narrowband interference filters in the optical subsystem. The electronics subsystem was used to electronically process the resultant signals before digitization. Ratios were computed digitally in real time using an Apple 2E microcomputer. This ratio based fiber optic pH measurement was able to to measure pH values in the 6.5-7.5 range with a standard deviation of better than 0.1 pH unit. Over this range a maximum standard deviation of 0.007 pH units $/{ }^{\circ} \mathrm{C}$ was measured. The time constant of these optrodes was determined to be 3.2 minutes when measured in 305 mOsm phosphate buffer. The time constant in whole blood increased to 10.0 minutes due to a decrease in the hydrogen ion permeability of the LUV membrane. This is most likely due to the blockage of gramicidin channels by divalent cations in the blood plasma.

## ACKNOWLEDGEMENTS

I am deeply grateful for the help of the many people who have assisted me throughout the course of this research project. Without their help an undertaking of this magnitude would not have been possible.

First, I would like to thank my main thesis advisor, Dr. Richard Magin, for his many helpful suggestions during the course of this project. His friendship, patience and encouragement throughout this endeavor helped in easing the burden of an otherwise formidable task. I would also like to thank Dr. Floyd Dunn for his guidance and help in overcoming many of the problems associated with the completion of this thesis.

I am grateful to Dr. Enrico Gratton for the use of his fluorescence laboratory as well as his many helpful technical discussions. Several of the experiments and ideas discussed in this thesis have come about as consequence of these discussions.

The help of several of my fellow students Tom White, Kevin Ehlert, Jay Alameda, and Francis Jatico was indispensable in completing this research project. Tom's theoretical noise study and Kevin's prototype hardware and software helped establish the foundation upon which the current system is built. Jay's expertise and help in preparing liposomes played a significant role in the successful fabrication of a practical sensor. Francis was directly responsible for acquiring much of the sensor data presented in this thesis.

I would like to extend a sincere note of thanks to several members of the Bioacoustics Research Laboratory for their contributions to this research project. Without the expert technical assistance provided by Joe Cobb and Bob Cicone much of this thesis would not have been possible. I am especially grateful to Billy McNeill for his help in turning many crude ideas and sketches into workable systems. His mechanical expertise was instrumental in making the optical subsystem a reality. A
special thanks is extended to Wanda for her help in getting this manuscript into final form. Her dedication and experience made this task far easier than it otherwise might have been.

Finally, I would like to thank the countless other faculty, staff, students and friends who have assisted me in some fashion during either the research or preparation phase of this thesis.

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## CHAPTER 1

## INTRODUCTION

The effects of elevated temperatures on physiological functions has fascinated researchers for many years [55]. The history of mammalian hyperthermia (temperatures $\geq 42^{\circ} \mathrm{C}$ ) as a means of treating malignant disease can be traced back several centuries [78]. Recently, there has been a revival of interest in the the use of hyperthermia as a clinical modality in the treatment of certain types of malignant tumors [41]. This has been primarily due to advent of new heating methods which make the induction of elevated temperatures, either locally or over the whole body, clinical feasible.

Today, the three physical modalities that are most often employed for power deposition in local and regional clinical hyperthermia are ultrasound at frequencies of about $\cdot 3-3 \mathrm{MHz}$, electromagnetic fields at radio frequencies of less than 300 MHz , and electromagnetic radiation at microwave frequencies of 300 $2,450 \mathrm{MHz}$ [70]. Use of these physical heating modalities, however, introduce difficulties in the monitoring of tissue parameters during the course of treatment.

Tissue temperature is usually monitored through the use of conventional thermistor or thermocouple probes. While these probes permit accurate data acquisition under normal conditions, interaction between the metallic leads of the probe and the applied electromagnetic field present complications during therapy. These complications result from a current density in the probe leads, as a consequence of an induced electric field along their length [70]. This current density can cause perturbations in different ways.

The electromagnetic field in the surrounding tissue can be perturbed by the reradiated fields from the probe leads. This would lead to nonuniform power deposition in the vicinity of the probe. In addition, probe heating and noise caused
by electromagnetic interference could cause errors in the accuracy and precision of the measurements obtained.

In order to circumvent these type of problems, fiber optic temperature measurement techniques were developed. Fiber optic probes, made of glass or plastic, are dielectric in nature and as such, their interaction with electromagnetic fields is considerably less than that seen with probes of conventional design. All optical temperature probes measure changes in some temperature-dependent interaction with light [70]. In a fluorescence based sensor, light is absorbed and reemitted at longer wavelengths. Many chemical and physical processes can effect the efficiency of optical energy transfer, as well as the observed emission wavelengths. One of the most widely used optical temperature sensors in hyperthermia is a fluorescence sensor made from a mixture of two phosphors [99]. The two phosphors employed have fluorescence intensities that are quenched at widely different temperatures with emission wavelengths that can easily be spectrally separated. A calibration curve, based on the ratio of these two wavelengths as a function of temperature, can be constructed and used to determine unknown temperatures from ratio data. Luxtron Corporation (Mountain View, CA) markets an optical temperature measurement system based on this concept.

As a result of the growth in the use of hyperthermia in cancer therapy, increased interest was generated in defining physiological parameters that could be used as prognostic indicators of response [91]. Such information is necessary in order to develop effective treatment protocols which could be used to optimize the therapeutic potential of hyperthermia.

Many in vitro studies and a limited number of in vivo studies (Chapter 2), suggest that the tumor microenvironmental values of $\mathrm{pH}, \mathrm{pO}_{2}, \mathrm{pCO}_{2}$ and perfusion may play a significant role in the response tumors to hyperthermia. The pH of the tumor microenvironment has been suggested as a good candidate for a
prognostic factor in predicting the response of tumors to hyperthermia. A decrease in extracellular pH within tumors has been shown to result in hyperthermia sensitization relative to normal tissue [92]. Furthermore, pH not only affects hyperthermic sensitivity but also may be expected to influence the cellular transport, metabolism, and cytotoxicity of chemotherapeutic drugs [31].

To date, limited data has been obtained on the in vivo correlation between tumor pH and hyperthermic response. Furthermore, no data exists on the dynamic pH response of tumors to induced clinical hyperthermia using current heating techniques. This situation is principally due to the lack of suitable pH measurement techniques and instrumentation.

Nearly all in vivo measurements of tumor pH have been made with microelectrodes. Due to the electrical nature of this measurement technique, it suffers from the same difficulties as conventional thermometry when used in an applied electromagnetic field. Fo: this reason, clinical measurements of tumor pH are performed, before and after therapy only, by removal and reinsertion of the microelectrode using an in-dwelling catheter. This technique precludes the acquisition of dynamic pH data. Such data has only been obtained during water bath heating experiments [82]. Furthermore, such techniques make exact relocation of the microelectrode difficult, so the two measurements may not be obtained from the same location. Given the heterogeneous pH distribution found within tumors (Chapter 2), large errors in pH measurement and subsequent data interpretation could result.

In order to help overcome the above difficulties, the development of a fiber optic, fluorescence emission ratio based pH measurement system was undertaken. The pH sensor was developed and evaluated using solutions of fluorophore dissolved in buffers at various values of pH . Suitability of both the measurement system and fluorophore, for use within the temperature range employed in clinical
hyperthermia, was demonstrated. In addition, miniature optical sensors, suitable for in vivo use with minimum tissue destruction, were developed and tested. Finally, the ability of both the sensors and measurement system to make accurate and precise pH measurements, in a physiologically relevant medium, was demonstrated.

Through the use of optical measurement techniques, for monitoring important hyperthermic parameters during therapy, planning and control of an optimal treatment strategy for a given tumor should one day be possible.

## CHAPTER 2

## BACKGROUND

### 2.1. Hyperthermia

### 2.1.1. Introduction

Many studies, some of which go back at least 100 years, have suggested that intrinsic differences in survival occur between normal and neoplastic cells when treated at hyperthermic temperatures [15, 88, 90, 91]. However, Hahn [38] concludes that although studies demonstrating increased heat sensitivity of malignant cells clearly exist, there are enough contradictory reports to suggest that this sensitivity is not a general characteristic of all malignant cells. In some in vitro studies, either no difference in heat sensitivity between normal and neoplastic cells was found, or neoplastic cells were found to be more heat resistant. In the older literature the data obtained were from experiments in which the assay procedures used, such as dye exclusion or morphological criteria, are no longer considered reliable. Thus, it is not surprising that earlier studies produced results that were extremely variable. In more recent experiments, the diversity of results obtained are more difficult to explain. Using in vitro studies, Chen and Heidelberger [16] clearly showed that mouse prostate cells, on being transformed by carcinogenic hydrocarbons, acquired a pronounced heat sensitivity. On the other hand, Harisiadis [42] compared survival of "normal" liver cells in vitro with those from a closely associated hepatoma. The hepatoma cells were found to be slightly more resistant to heat than normal liver cells.

With respect to in vivo studies, a slightly different picture appears. In early in vivo studies Overgaard [68] observed that hyperthermic exposures between 41.5 and $43.5^{\circ} \mathrm{C}$ caused little histological damage in normal mouse mammary tissue, but
resulted in severe damage to mammary carcinoma tissue. However, Sapareto [78] has pointed out that temperature differences between tumor and adjacent normal tissue complicate the interpretation of this study.

Kang [49] has studied the response of SCK mammary adenocarcinoma, growing subcutaneously in the leg of $\mathrm{A} / \mathrm{J}$ mice, to both in vivo and in vitro hyperthermia. He found that the number of clonogenic cells in tumors excised immediately after heating was significantly less than that in the in vitro culture treated with the same heat doses. This result suggests that factors in the tumor microenvironment may play an important role in modulating the effects of in vivo heat treatment.

Despite the lack of convincing evidence that neoplastic cells are intrinsically more sensitive to heat than normal cells in vitro, Song [85] suggests that clinical experience as well as studies with animal tumors would indicate that tumors are more heat sensitive than normal tissue in vivo.

To understand why in vivo results do not always agree with in vitro observations, it is necessary to study the mechanisms of heat injury, and their potential for modulation by physiological and biochemical factors present in the tumor microenvironment [9].

### 2.1.2. Cellular Mechanisms of Heat Injury

Due to the ubiquitous nature of heating effects on cellular components, as well as on physiological and biochemical variables, its is impossible to establish a single mechanism for all heated induced cell injury. According to Sapareto [78], three major mechanisms of heat injury at the cellular level have been proposed.

The first mechanism involves a direct effect on the cell membrane, changing its permeability, composition or fluidity, and ultimately leading to the death of the cell. In Chinese hamster ovary (CHO) cells, hyperthermia inhibits thymidine uptake
by facilitated diffusion [7]. Also, heat causes the plasma membrane to become permeable to polyamines, such as putrescine, spermidine, and spermine [32, 59]. Heat effects on membrane fluidity have been implicated by the observed interaction of heat with membrane modifying drugs. Both alcohols [57] and local anesthetics [102] have been shown to cause increased sensitivity to heat. As far as membrane composition is concerned, Cress [19] has shown an inverse relationship between cholesterol to phospholipid ratio and heat sensitivity. Evidence against the membrane hypothesis is that activation energies for most types of membrane damage are low. For example, the activation energy required to induce permeability changes is $20 \mathrm{kcal} / \mathrm{mol}$ for ascites tumor cell membranes, and the activation energy for loss of adenyl cyclase activity is $27 \mathrm{kcal} / \mathrm{mol}$ [81]. These values are far lower than the observed activation energies for cell killing of 150 and $300 \mathrm{kcal} /$ mol above and below $43^{\circ} \mathrm{C}$, respectively. Another argument against membrane damage being solely responsible for heat induced cell death is that membrane turnover and replacement occurs to a greater extent in plateau phase cells, which have been found to be more heat sensitive than in exponentially growing cells.

The second mechanism of cellular heat injury was suggested by the histological observations of Overgaard [67]. These observations show an increase in lysosomes in the cellular cytoplasm after heat exposure. It has been suggested that disintegration or damage of these lysosome vesicles may release digestive enzymes leading to cellular death. Biochemical evidence of increased lysosomal enzyme activity during hyperthermia by several investigators $[45,69]$ lends support to this theory. However, because lysosomes are involved in the destruction of dead cells, this evidence may reflect tissue response to other physiological changes caused by heat, and thus be a secondary effect of cell death. Evidence against the lysosome
theory is the observation that agents which modify lysosomal membranes (trypan blue, retinol, and hydrocortisone) did not affect heat induced cell killing [44].

The third mechanism for cellular heat injury involves thermal damage to proteins. The evidence for the idea that protein denaturation is involved in cell killing is that the activation enthalpy for cell killing above $45^{\circ} \mathrm{C}$ is $150 \mathrm{kcal} / \mathrm{mol}$, which is similar to that observed for protein denaturation. Several investigators implicate heat in affecting a number of protein functions such as DNA synthesis, RNA synthesis, protein synthesis, and respiration. However, Roti-Roti [76] maintains that the protein denaturation hypothesis is untestable unless the proteins responsible for cell killing are specified.

A possible fourth mechanism of heat injury, at the nuclear level, has also been suggested. Tomasovic [93] and Roti-Roti [76] both have reported an increased, nonspecific attachment of nonhistone nuclear proteins to DNA following hyperthermia. They have subsequently demonstrated that this increased chromatin protein mass impairs chromatin function and is also highly correlated with heat induced cell killing.

### 2.1.3. Biochemical and Physiological Modulating Factors

Regardless of the actual mechanisms involved in heat induced cell killing, from the differences observed between in vitro and in vivo studies, it is reasonable to postulate the existence of physiological and biochemical factors in the tumor microenvironment which modulate the underlying mechanisms in ways that result in increase thermal sensitivity. Current research has found at least four significant factors which modify cellular heat response. These factors are perfusion, pH , oxygen consumption, and nutrient or metabolic levels. However, in spite of the importance of these factors for clinical hyperthermia, little in vivo information is
available about either their temporal or spatial variation, their relationship to one another, or their relationship to any of the previously proposed mechanisms.

### 2.1.3.1. Perfusion

Tumor blood flow, or perfusion, is emerging as the major mediator in tumor response to heat, since it governs not only the local tumor environment (nutrient supply, oxygen level, pH ) but is also the key link in the host-tumor relationship [23]. While tumor blood flow is influenced by heat, it is also affected by the local tumor microenvironment. Constituents of the microenvironment that are influenced by blood flow have in turn been found to exert some degree of reciprocal control on the flow itself. Thus, it could well be that a tumor's blood flow response to heating could be a secondary manifestation of a more direct effect on a primary microenvironmental variable.

There is considerable controversy about whether or not the blood perfusion of tumors is greater than that of normal tissue under normothermic conditions. The flow response to hyperthermic conditions in both tissues is even more controversial. LeVeen [56] states that tumor blood flow from surgically excised material was $2-15 \%$ of normal tissue. On the other hand, Bierman [13] found that in 12 patients with metastatic, neoplastic lesions, the blood flow through the tumors was greater than that through normal tissue. Song [86] concludes that tumor blood flow varies significantly depending on the type, age, and size of the tumors. He also concludes that due to the heterogeneous distribution of perfusion in tumors, blood flow may or may not be greater than the surrounding tissues at normothermic temperatures. At hyperthermic temperatures, Song [86] found that tumor blood flow either remains unchanged or increases less than a factor of two, when heated at $41-43^{\circ} \mathrm{C}$. In contrast he found that the blood flow to normal tissue increase by a factor of 3-20 on heating at $42-45^{\circ} \mathrm{C}$. Meanwhile, Bicher [10] found that a rise in
temperature up to $41^{\circ} \mathrm{C}$ leads to a significant increase in tumor blood flow, while a further rise in temperature up to $42^{\circ} \mathrm{C}$ results in a marked breakdown of this flow to below the initial value. He also found similar results for normal tissue, however, the break point occurs at a much higher temperature (approximately $46^{\circ} \mathrm{C}$ ).

The actual mechanisms responsible for the variations in tumor blood flow observed during hyperthermia remain the subject of much controversy. Virtually all the blood flow measurements that have been done to date either measure blood flow at a single point or measure average blood flow rate throughout the tumor. Hahn [39] has pointed out that the inherent biological variability and heterogeneity of tumors make the description of blood flow rates by one number not very meaningful. To characterize the role of blood flow during hyperthermia, a means of determining its temporal and spatial variation must be found. Since blood flow is responsible for almost all convective heat transfer in tissue, its importance in the development of successful hyperthermia treatment planning is clear.

### 2.1.3.2. $\mathbf{p H}$

There is now a large amount of evidence [92] indicating that exposure of cells in vitro to a low pH environment sensitizes them to hyperthermia. It has also been established that the intratumor environment is acidic relative to normal tissue and that its pH further decreases during hyperthermic treatment [83].

Most of the in vitro studies in this area were carried out by Gerweck and Overgaard. Gerweck [29] showed that the pH sensitizing effect took place over a temperature range of $41-44^{\circ} \mathrm{C}$ and increased with decreasing pH ; the effect was particularly pronounced at $42^{\circ} \mathrm{C}$ and became less evident at 43 and $44^{\circ} \mathrm{C}$. Gerweck [30] also found that maintaining tissue culture cells at low pH after heating increased the cytocidal effects of hyperthermia and inhibited the onset of thermal tolerance. Overgaard [65, 66] studied the ability of L182 ascities tumor cells
suspensions, heated in vitro at $42.5^{\circ} \mathrm{C}$ for 60 minutes at either pH 7.2 or 6.4 , to form tumors when injected into mice immediately after the incubation period. He found that cells that had been heated at pH 7.2 were able to form tumors in $100 \%$ of the hosts, while those heated at pH 6.4 were incapable of initiating tumors. The major change occurred between pH values of 7.2 and 7.0 where the percentage of successful tumor growths was reduced from 100 to $33 \%$. Overgaard also found on ultrastructural examination that the number of cells having lesions in their plasma membranes, as well as those showing increased lysosomal activity, increased when heating and incubation took place at low values of pH .

Dickson and Calderwood [23] measured the extracellular pH values both in tumors and in normal tissue. They found the extracellular pH range for tumors (7.19-6.99) to be slightly lower than that for normal liver (7.32) or normal muscle (7.21). Several other studies have also shown that the extracellular pH of tumors is consistently lower than that of normal tissue. However, Bicher [10] has shown that, as in blood flow, extracellular pH values differed considerably between different parts of tumors. The data of Eden et al. [25] indicate that hydrogen ion concentration in some areas of tumors may be more than 10 times those in areas of neutral pH .

The effective of hyperthermia on pH has been studied by several investigators. Song [82] observed that the pH in control SCK tumors of mice was 7.05. On heating at $43.5^{\circ} \mathrm{C}$, it temporarily increased and then rapidly decreased reaching 6.67 at the end of 30 minutes of heating. When the heating was terminated, the pH rose to 6.78 , but it decreased to $6.5-6.6$ when the tumors were heated again. He also found similar decreases in pH in Walker tumor 256 heated at 43 or $46^{\circ} \mathrm{C}$. However contrary to the behavior of tumor pH , the pH in the muscle tissue of rats increased when heated at temperatures up to $46^{\circ} \mathrm{C}$, but decreased at temperatures above $46^{\circ} \mathrm{C}$. Bicher [10] reported extracellular pH decreases in human tumors of
0.5 to 1 unit at temperatures above $42^{\circ} \mathrm{C}$. It seems that the extracellular pH change observed after heating varies according to tumor type and heating conditions. In general, the higher the temperature and the longer the heating the greater the decrease in tumor pH .

The cause of the extracellular decrease in tumor pH during hyperthermia is not clear. Changes in blood flow and oxygen level have been suggested. Also, changes in nutrient level and a metabolic shift to a higher level of glycolytic activity may cause lactic acid accumulation resulting in decreased extracellular pH . Von Ardenne [95] found that hyperglycemia by itself selectively reduces tumor pH to near 6.0 while heating further reduces the pH . The influence of blood flow and other microenvironmental changes on pH has yet to be unravelled. Song [82], while measuring temporal variation in pH and temperature, did not measure simultaneous temporal variations in blood flow. His use of average blood flow measurements did not allow him to resolve the issue of whether changes in blood flow caused the observed changes in pH or whether the change in pH was the dominant factor in modifying blood flow. Bicher [10], while able to measure blood flow, pH , and temperature simultaneously with some degree of temporal resolution, did not use a hyperthermic range of temperatures for those measurements ( $<40^{\circ} \mathrm{C}$ ). Also, his failure to measure the simultaneous spatial variation of these variables made it impossible to determine their relationship to each other, given the known heterogeneity of tumor microenvironments.

From a clinical perspective a knowledge of extracellular pH and the ability to modulate it are extremely important. Hahn et al. [40] have demonstrated the enhanced cytotoxicity of some chemotherapeutic drugs in regions of decreased extracellular pH . Knowledge of the relationship between blood flow and pH may allow for selective modulation of the appropriate factors, during hyperthermia, in order to achieve maximal therapeutic effect.

### 2.1.3.3. Oxygen Level

Most reports in the literature indicate that in vitro hypoxic cells are as sensitive or more sensitive to heat when compared to oxic cells. The early reports of Hahn [37] on the response of CHO cells to heating at $43^{\circ} \mathrm{C}$ found survival to be independent of the presence or absence of oxygen during heating. Bass [8] found a slight protective effect of hypoxia against the killing of HeLa cells exposed to $43^{\circ} \mathrm{C}$. Kim [50] found that oxygen depleted HeLa cells were appreciably more heat sensitive than their well oxygenated counterparts. Hahn [38] points out that since the roles of pH and nutritional factors were not appreciated at the time of these experiments, the response of the cells may not be simply the result of their hypoxia but may be due to a combination of factors. In fact, Adams et al. [2] found that cells cultured in suspension at pH 7.4 show substantial heat resistance in air after chronic exposure to hypoxia. These results cast some doubt on the concept of hypoxia enhanced heat sensitivity in vitro.

There is no data directly linking hypoxia with thermal sensitivity in vivo, although there is a considerable body of indirect evidence. Crile [20] and Suit [89] found that clamping the tumor blood supply increases thermosensitivity, and this sensitizing effect increases with the duration of clamping before heating. This seems to indicate that chronic hypoxia is more important than acute hypoxia in gaining increased thermosensitivity. However, Dickson et al. [23] point out that increased heat sensitivity with duration of clamping indicates that the effect involves more than just increased uniformity of heating or decreased oxygen levels. Nutrient level, pH , catabolite level, and other biochemical parameters may be altered. During hyperthermia, Bicher [10] has shown that tumor $\mathrm{pO}_{2}$ closely follows changes in tissue temperature. The response is very fast, with tumor $\mathrm{pO}_{2}$ increasing shortly after the rise in temperature and then decreasing as the tumor
cools off. This effect was always seen when heating took place below $41^{\circ} \mathrm{C}$. At higher temperatures, there was an initial increase in tumor $\mathrm{pO}_{2}$ which was followed by a decrease to low levels as the temperature was held constant at $46^{\circ} \mathrm{C}$. The same pattern of $\mathrm{pO}_{2}$ variations was found when normal tissue was heated, only the temperature of $\mathrm{pO}_{2}$ fall off was always significantly lower in tumors. A strong correlation between decreases in tumor $\mathrm{pO}_{2}$ and blood flow was found as the temperature was increased to $45^{\circ} \mathrm{C}$. Furthermore, Bicher noted widely differing values of $\mathrm{pO}_{2}$ even within a single tumor [11].

When hyperthermia is used as an adjuvant to radiotherapy, the tumor $\mathrm{pO}_{2}$ could be a major factor in determining the outcome of this treatment. It is well known that oxygen sensitizes cells to radiation. If hypoxic cells are truly more sensitive to heat, then the ability to modulate tumor $\mathrm{pO}_{2}$ may have significant impact in treatment planning when using these combined modalities. To resolve this question, however, the relationship of $\mathrm{pO}_{2}$ to blood flow, pH , and nutrient level must first be established.

### 2.1.3.4. Nutrient Levels and Metabolism

Several investigators have postulated that low levels of nutrients found in the microenvironment of many tumors, as a result of their low levels of perfusion, aid in increasing their thermosensitivity. Warburg [46] demonstrated that unlike normal cells, malignant cells are generally capable of aerobic glycolysis and thus use glucose at a rapid rate. He also demonstrated that lactic acid production by malignant cells, contrary to normal cells, increases as a function of extracellular glucose concentration. These facts led Von Ardenne et al. [38] to speculate that glucose infusions could lead to hyperacidification of tumors with a resultant increased thermosensitivity. Other researchers, such as Song [84] demonstrated by use of 5-thio-D-glucose, an inhibitor of glycolysis, that since hypoxic cells depend
on glycolytic pathways for energy metabolism, the increased cytotoxicity seen in hypoxic versus oxic cells in this experiment was not due to glucose deprivation alone but due to a complex interrelationship between the availability of both oxygen and glucose. Kim [51] showed that the selective effect of this glucose analogue on hypoxic cells was greatly magnified at elevated temperatures. In these studies, however, no data on extracellular pH were presented.

Thus, again evidence is seen of the complex interrelationship of one microenvironmental variable with others. It should also be noted that Song (unpublished results) suggests that insufficient nutrient levels, along with decreased pH , may also be a contributing factor to the reduced thermotolerance seen in heated tumors in vivo, but he is unable to separate their combined influences.

There is considerable controversy regarding the effect of hyperthermia on metabolism. This is partly due to the fact that metabolism is once again regulated by two microenvironmental variables, pH and $\mathrm{pO}_{2}$. Mondovi [59] reported a decrease of the respiration rate when Novikoff hepatoma cells were incubated at $38^{\circ} \mathrm{C}$ with glucose and succinate after a preincubation of 3.5 hours at $43^{\circ} \mathrm{C}$. Glycolysis was only slightly reduced under the same conditions. However, no effect was observed when the assay and preincubation were performed without oxygen. Dickson et al. [24] investigated the metabolism of Yoshida sarcoma in rats. They found that heating of the tumor for 1 hour at $40^{\circ} \mathrm{C}$ had no influence on the respiration or on anaerobic glycolysis, both measured in vitro at $38^{\circ} \mathrm{C}$. However, after preincubation of the tumors in situ at $42^{\circ} \mathrm{C}$, both parameters were depressed, with a greater depression for aerobic glycolysis. Furthermore, if the rats were made hyperglycemic by glucose injections given before hyperthermia, than even a temperature of $40^{\circ} \mathrm{C}$ caused a marked inhibition especially on aerobic glycolysis. Extracellular pH decreased after glucose loading but the correlation between lactate accumulation and decrease of extracellular pH was poor. These results are at odds
with the theory of Song. Streffer [87] measured glycolytic metabolites in liver and in transplanted adenocarcinoma E0771 in mice after heat treatment. He found that both respiration and glycolysis were enhanced during hyperthermia but decreased immediately afterwards. Furthermore, he found that lactate was only slightly increased whereas two acidic metabolites, acetoacetate and B-hydroxybutyrate were increased considerably. However, if the hyperthermic treatment was combined with a glucose load given before hyperthermic exposure, an increase in lactate accumulation was observed. These findings do not agree with those of the similar glucose loading experiments of Dickson et al. [24] previously mentioned.

Again many widely varying and confusing findings have been reported. As in the past, much of this variability can probably be attributed to significant differences in the state of the experimental tissues used. Only when the complete state of the cellular system is known, can any valid conclusions be drawn regarding the possibility of metabolically or nutritionally modulating cellular heat response for more effective therapy. One step in this direction lies in the development of fiber optic sensors to measure the microenvironmental variables of interest during actual clinical hyperthermic treatments.

### 2.2. Fiber Optic Chemical Sensors

### 2.2.1. Introduction

Early fiber optic sensors relied on the physical property of the medium being sampled to cause a change in the light transmitting properties of the optical fiber. By using such methods, acceleration, strain, position, magnetic field, and other physical properties could be monitored [4]. Later, optical transducers were used at the distal end of an optical fiber to provide enhanced sensitivity or chemical specificity. These optical transducers (optrodes) were classified as physical
(responding to such parameters as temperature and pressure) or chemical (responding to chemical concentration), depending on what was being measured [4]. All chemical optrodes involve a reagent phase coupled to an optical fiber, and measure concentrations through a change in the optical properties of the reagent [63]. Light from a suitable source travels along an excitation fiber and is returned from the reagent phase by either scattering or luminescence stimulated by the excitation source. Either the same or different fibers can be used to collect and transmit the returned optical signal to an appropriate photodetector.

According to Angel [4], optical sensors have many advantages over the use of potentiometric electrodes. These include electromagnetic immunity, low cost, small physical size, physical separation of the sample and the instrument, and the ability to easily multiplex many sensors to a central instrument. In addition, optrodes are not as sensitive to surface contamination because they respond to actual concentrations rather than concentration gradients.

Several disadvantages of fiber optic chemical sensors have also been noted [63]. Since ambient light will interfere with the sensors, they must be used in the dark or the optical signal must be modulated. In addition, the response time of such sensors is limited by the necessity of a mass-transfer step before a constant response can be reached. The limited dynamic range of most chemical sensors also tends to limit their applicability.

### 2.2.2. pH Sensors

Fiber optic pH sensors have principally been developed for biomedical or biological applications. Consequently, the pH range of such sensors always includes the physiological range (7.0-7.5) [79]. All fiber optic pH sensors developed to date have used a pH sensitive indicator dye immobilized on a solid support matrix which is then affixed in some fashion maner to the end of an optical fiber. In most cases a
semipermeable membrane is used to prevent dye leakage and allow for easy diffusion of hydrogen ions into the dye matrix. Both absorption and fluorescence techniques have been used to monitor the pH dependent behavior of these dyes.

Recently, techniques have been developed which allow the indicator dye to be directly fixed to the fiber end surface rather than to a solid support [61]. This improvement has resulted in further probe miniaturization, reduction in response time, since the membrane envelope is no longer necessary, and high mechanical stability.

Since pH is defined in terms of activity, while optical techniques measure the number of molecules (i.e. concentration), significant errors can occur in the optical measurement of pH if the matrix of the sample differs substantially from that of the calibration solution [48]. The solute-solvent and solute-solute interactions, which determine the value of activity, show up as second-order effects which are frequently ignored [47]. Unfortunately, this omission can only be tolerated for very dilute aqueous solutions where the activity coefficients tend to unity. In real measurement situations, the effect of ionic strength differences on the indicator dye must be taken into account in arriving at the correct value of pH for an unknown sample.

The first fiber optic pH sensor was developed by Peterson [73]. This sensor was based on the spectral changes of the indicator phenol red with pH . This indicator dye was immobilized on polyacrylamide microspheres (5-10 $\mu \mathrm{m}$ in diameter) which contain small light scattering polystyrene microspheres (approximately $1 \mu \mathrm{~m}$ in diameter). These microspheres are confined to the tip of the optical fiber by using hydrogen ion permeable cellulose dialysis tubing. Light of wavelengths 558 and 600 nm is used. The 558 nm light signal is a function of pH , while the 600 nm light serves as a pH independent reference signal. This probe can measure pH in the physiological 6.8-7.4 range. It had an accuracy and precision of
$\pm 0.01 \mathrm{pH}$ unit and a temperature coefficient of 0.017 pH units $/{ }^{\circ} \mathrm{C}$. This probe had a reported diameter of about 0.4 mm , a length of 3 mm and a time constant of approximately 0.7 minutes. A significant dependence of the pH response on the ionic strength of the sample was noted for this sensor.

The first fluorescence based fiber optic pH sensor was developed by Saari and Seitz [77]. It was based on fluoresceinamine immobililized on cellulose or porous glass. Excitation was at 480 nm with a single wavelength fluorescent emission measured at 520 nm . The working pH range of this sensor was 3-6. A bifurcated optical fiber was used for making the optical measurements. Steady state fluorescent response was achieved by this sensor in about 15-30 seconds. The accuracy and precision of this sensor are not good. This is mainly due to a poor SNR resulting from a reduction in the fluorescence signal intensity as a consequence of the dye immobilization procedure employed. In addition, high levels of background signal, as a result of light scattering by the substrate, caused further degradation of the SNR.

More recently the development of a fluorescence sensor for quantifying pH values in the 6.5-8.5 range has been based upon the electrostatic immobilization of the trisodium salt of 8-hydroxyl-1,3,6-pyrene trisulfonic acid (HOPSA) on an anion exchange membrane [103]. A bifurcated optical fiber was again employed. The pH values were obtained from the ratio of two fluorescent emissions measured at 520 nm following a sequential excitation at wavelengths of 405 and 470 nm . For a pH change from 6 to 8 , the measured response time was 2 minutes and the standard deviation of the final pH was $\pm 0.03$ units. A temperature coefficient of $1.1 \% /{ }^{\circ} \mathrm{C}$ was measured. No measurable influence on the calculated pH by the ionic strength of the sample could be observed while using this sensor.

Lastly, a fluorescence sensor for pH in the 6.4-7.7 range was recently developed based upon the glass-immobilized fluorescent pH indicators

1-hydroxypyren-3,6,8-trisulphonate (HPTS) and 7-hydroxycoumarin-3carboxylic acid (HCC) [64]. Again a bifurcated optical fiber was used. However in this study, only single excitation and emission wavelengths were employed due to the complexity of the instrumentation required for the sequential wavelength excitation method. Analytical excitation and emission wavelengths were, respectively 410 and 455 nm for the HCC-based sensor, and 465 and 520 nm for the HPTS-based sensor. These sensors were reported to have a precision of $\pm 0.01$ unit and a response time in the order of 1 minute. Effects of ionic strengths on the optically determined pH values were small.

With the introduction of the pH sensitive dual emission fluorophore 1,4dihydroxyphalonitrile (1,4-DHPN), an optical pH measurement system can be built in order to take advantage of the benefits of the dual wavelength ratio technique, without the additional complexity of the instrumentation usually associated with this method. Furthermore, since only a single broad band excitation is required, simultaneous acquisition of the fluorescence emission intensities adds to the short term stability and precision of this technique.

The research effort described in the following chapters focuses on the development and evaluation of a fluorescence emission ratio based fiber optic pH measurement system utilizing the above technique. This system was designed specifically for use in monitoring changes in tumor pH during clinical hyperthermia.

## CHAPTER 3

## PHYSICAL CHEMICAL STUDIES OF 1,4-DHPN

### 3.1. Introduction

In fabricating a simple pH sensitive optrode, it is desirable to use the same optical fiber for both exciting the fluorophore and collecting its fluorescence emissions. This single fiber design simplifies instrument development by avoiding the difficulties associated with multiple fiber alignment at the sensor interface. Reduction in overall size and bulk of the resulting optrode is also achieved by using this type of design. With such an optrode, fluorescence emission spectroscopy following a single excitation offers many of the same advantages for optical sensors that it does for flow cytometry.

Due to the similarity in requirements of fluorophores for both optical sensors and flow cytometry, 1,4-DHPN was chosen as the pH sensitive component of the optrode. Further studies of 1,4-DHPN were undertaken to both confirm and extend the existing base of knowledge with regard to the physical chemical properties of this compound. In particular, studies designed to measure the pK values of 1,4 -DHPN were carried out and the sensitivity of several physical chemical properties of this compound to shifts in both pH and temperature were examined. Finally, consideration was given to the variety of ways in which this compound might be used for measuring pH . The advantages and disadvantages of each of these techniques, with regard to optical sensor design, is discussed.

The pH sensitive fluorescent probe 1,4-dihydroxyphalonitrile (1,4-DHPN), also known as 2,3-dicyano-1,4-hydroquinone, was purchased from Molecular Probes, Inc. (Eugene, OR), and forms the core of the proposed pH sensitive optrode.

The chemical forms of this compound (Fig. 3.1) in solution are dependent upon the pH of the solution with respect to the pK values of the fluorophore. Since the optical properties of each form are distinct, several spectroscopic techniques may be used to discern the fractional contribution of each form to the chemical characteristic being measured, and thus uniquely determine the pH of the solution under a standard set of conditions.

Until recently, the acid-base properties of hydroquinones, especially cyanosubstituted hydroquinones, have received relatively little attention. Since the purpose of any spectroscopic probe is to be able to discover information about its surroundings from a study of the properties of its absorption or emission, it is essential that a thorough knowledge of the properties of the isolated probe be acquired. Only then can alterations in its spectroscopic behavior, caused by the properties of a new environment, be analyzed with certainty.

The fluorescent probe 1,4 -DHPN can be prepared by the addition of two moles of hydrogen cyanide to one mole of benzoquinone, followed by recrystallization from distilled water [18]. This results in a yellow leaf shaped crystal with a molecular weight of 160.16 Daltons and a melting point of $230^{\circ} \mathrm{C}$. These crystals are very soluble in ethyl alcohol or diethyl ether and only slightly soluble in water, benzene, and chloroform [96]. Since hydroquinones are highly conjugated structures, they are rather closely balanced energetically against the corresponding quinones [60]. This results in a ready interconversion of 2,3-dicyano-1,4-hydroquinone to 2,3 -dicyano-1,4-benzoquinone in the presence of molecular oxygen (Fig. 3.2). This process is made apparent by a change in color of the solution of 1,4-DHPN from pale yellow to rust brown and occurs over the course of several days, at room temperature. Changes in both absorbance and fluorescence have also been noted during the course of this interconversion.

The acid-base and spectroscopic properties of 1,4-DHPN were first measured in 1977 by Brown et al.[14]. Absorption spectra of 1,4-DHPN solutions between pH 10.0 and pH 3.5 were analyzed as a sum of contributions from three species with two equilibria. From this analysis, pK values of $8.0 \pm 0.2$ (dianion/monanion equilibrium) and $5.5 \pm 0.3$ (monoanion/neutral species equilibrium) were obtained. The absorption maximum for the three species involved were found to occur at 405 nm for the dianion, 380 nm for the anion, and 345 nm for the neutral species.

The fluorescence lifetime and emission properties of 1,4 -DHPN were also measured by Brown et al. Lifetime measurements were made using single-photon counting and were evaluated by computer convolution. Lifetimes of 7.7 ns for the dianion, 10.0 ns for the anion, and 14.0 ns for the neutral species were obtained. Computer convolution of the emission spectra at various values of pH yielded emission maxima of 480 nm for the dianion, 450 nm for the anion, and 400 nm for the neutral species.

The first biophysical utilization of 1,4-DHPN did not occur until 1981 when Valet [94] realized its unique potential for determination of the pH of single cells in flow cytometry. Most fluorescent pH indicators, including fluorescein, have maximum emission at a fixed wavelength with an intensity that is pH dependent. This makes a sequential dual-wavelength excitation necessary for reliable pH determination with such indicators. The disadvantage of this sequential method is that it allows only the mean pH value of a cell suspension to be determined by flow cytometry, since the measurement of the pH of any single cell would require two passes of that same cell through the flow cytometer. Valet was able to measure the pH of individual cells in a flow cytometer by taking advantage of the fact that 1,4DHPN shows a pH dependent shift of its peak emission wavelength. By using a 300400 nm broadband excitation, and a simultaneous measurement of the ratio of
fluorescence emission in two distinct bands ( $420-440 \mathrm{~nm}$ and $500-580 \mathrm{~nm}$ ), he was able to determine the pH of a single cell in a one-step measurement.

In a subsequent study by Kurtz et al. [54], 1,4-DHPN was used in conjunction with a microspectofluorometer to measure the topographical variation of intracellular pH within cultured A6 monolayers derived from toad kidney cells. A broad band excitation from $375-407 \mathrm{~nm}$ was used and the $512 \mathrm{~nm} / 455 \mathrm{~nm}$ emission ratio was determined. This ratio was then taken as a gauge of the wavelength of the fluorescence emission maximum, which in turn was a measure of the intracellular pH . Since the two emission intensities were determined simultaneously, Kurtz et al. found that measurements made in this fashion were independent of dye concentration, photobleaching, and intensity fluctuations of the excitation source. All three of these complications introduce measurement errors when sequential excitation spectroscopy was used. Kurtz et al. also established that the value of the ratio obtained was not altered by varying the concentrations of $\mathrm{Na}^{+}$ $(20-130 \mathrm{mM}), \mathrm{K}^{+}(30-130 \mathrm{mM}), \mathrm{Ca}^{++}(0-1 \mathrm{mM}), \mathrm{Mg}^{++}(0-1 \mathrm{mM}), \mathrm{PO}_{4}^{-3}$ (0-10 mM ), and albumin ( $0-10 \mathrm{~g} / \mathrm{l}$ ).

In a recent study [62], the physiological pH sensitive indicators 2,3-dicyanohydroquinone (1,4-DHPN), 4-methyl-umbelliferone (4MU), and 2',7'-bis(carboxyethyl)-5,6-carboxy fluorescein (BCECF) were evaluated in terms of resolution, range, and stability of cellular fluorescence. In each case, the ratio of two emission wavelengths following a single excitation was taken as a measure of pH . It was found that $1,4-\mathrm{DHPN}$ exhibited the best resolution of the three indicators tested over a useful range of greater than 1.5 pH units. The greater pH resolution of 1,4-DHPN, when measured by the ratio technique using fluorescence emission spectroscopy, is probably due to the fact that both of the emission wavelengths used in the ratio are pH sensitive and change in opposite directions as the pH of the sample is varied. This is in sharp contrast to the other two dyes in which only one
emission wavelength is sensitive to pH shifts, while the other remains at a constant value independent of the sample pH being measured. Reports of an accuracy of $\pm 0.02 \mathrm{pH}$ units have been reported where 1,4-DHPN emission spectroscopy has been employed during flow cytometric measurement of pH [33].

### 3.2. Potentiometric Titration

### 3.2.1. Procedure

One liter of 17.5 mM NaOH was prepared and standardized against the primary standard, potassium acid phthalate, according to a modification of a published procedure [27]. This resulted in a mean calculated NaOH molarity of 17.3 mM with a standard deviation of 0.4 mM for three separate determinations. The pH of all titrations was monitored using a Beckman Model 71 pH meter with a combination electrode. The volume of NaOH consumed at the equivalence point was determined graphically from the titration curves.

A 2 mM solution of 1,4 -DHPN was prepared using a $50 / 50$, by volume, ethanol/water solvent. This mixed solvent was necessary due to the limited solubility of 1,4 -DHPN in water at acidic pH values. This solution was then titrated against the previously standardized NaOH . The pH during the entire course of the titration was monitored, using a Beckman Model 71 pH meter, and the total volume of NaOH consumed at each pH was recorded. The pK values of 1,4 -DHPN were then determined from a computer aided analysis of the titration curve.

### 3.2.2. Results and Discussion

The titration curve for 1,4-DHPN (Fig. 3.3) is typical for a weak polyprotic acid titrated with a strong base. The actual shape of any given curve depends on the absolute ionization constants of the acid being titrated, the relative strengths of the ionizable groups, and the concentrations of the solutions used [27]. As the acid
becomes progressively weaker, the distinctness of the inflection at the equivalence points diminishes and the pH at these points shifts to higher values. In addition, for a polyprotic acid, difficulty in locating distinct inflection points occurs when the ratio of the first to the second dissociation constant approaches values of 100 or less. To give sharp inflection points, the ratio of the first to the second dissociation constant must be greater than 105 . Furthermore, it has been determined that for an uncertainty of $0.1 \%$ or less in aqueous solution, the product $\mathrm{K}_{\mathrm{a}}[\mathrm{HA}]$ should exceed $10^{-8}$, assuming the titrant is completely dissociated and 0.1 N in strength [100].

From the above, it is clear that obtaining accurate pK values from a visual inspection of the depicted 1,4-DHPN titration curve shows little promise. At the concentrations of organic acid and titrant used in this experiment, the first pK approaches the limit of detectability and the second pK far exceeds this limit. The situation is further complicated by the fact that a $50 \%$ ethanol solvent was used in this experiment. Thus, both the pH values and ionization constants obtained directly from this curve are "apparent" and may not agree with the values obtained in water alone.

In order to locate more accurately the equivalence points for the experimentally obtained 1,4 -DHPN titration curve, two different techniques were employed. First, titration of a blank solution was performed. This solution was prepared and titrated like the sample itself, but without the addition of the organic acid $1,4-$ DHPN. The volume of titrant used to achieve a specific blank pH was subtracted from the volume of titrant used to achieve the same value of pH in the sample solution. This procedure has been described by Parke et al. [71]. The resultant titration curve (Fig. 3.4) has now been corrected for errors introảuced by solvent impurities, as well as any volume errors that may have been caused by the acid-base properties of the solvent utilized. The equivalence points for this corrected titration curve are then obtained by numerical differentiation, in order to
locate the volumes of NaOH required to make the second derivative of the titration curve equal to zero as the value of the ordinate rapidly changes from a positive to a negative number. The pH values corresponding to these volumes can be read directly from the corrected titration curve and correspond to the pK values of the acid being titrated. Using this procedure in the case of $1,4-\mathrm{DHPN}$, the pK values of $5.59 \pm 0.05$ at 2.25 ml of titrant and $7.95 \pm 0.2$ at 9.74 ml of titrant were obtained. The first pK can be read very accurately using this technique, while the second value bears a slightly higher degree of uncertainty. These values agree with the 5.5 $\pm 0.3$ and $8.0 \pm 0.2$ results previously obtained by Brown et al. [14].

### 3.3. Ultraviolet and Visible Absorption Measurements

### 3.3.1. Procedure

Stock solutions of $122 \mathrm{mM} \mathrm{Na}{ }_{2} \mathrm{HPO}_{4} \cdot 7 \mathrm{H}_{2} \mathrm{O}$ (Solution A) and 122 mM $\mathrm{KH}_{2} \mathrm{PO}_{4}$ (Solution B) were prepared in distilled water. Phosphate buffers covering the pH range of $6.0-8.0$, in 0.5 unit pH steps, were prepared by mixing together appropriate volumes of the stock solutions A and B while observing the final pH value of the mixture using a Beckman Model 71 pH meter with a combination electrode. Using a $50 / 50$ mixture of stock solutions $A$ and $B$ resulted in a phosphate buffer of approximately 305 mOsm , with a pH of about 7 . This procedure resulted in a maximum of $\pm 20 \%$ osmotic error, occurring at the extremes of the useable range ( pH 5.0 and 9.0) of this buffer.

A 10 mM stock solution of $1,4-$ DHPN was prepared in distilled water. The pH of this solution was adjusted using 0.1 N NaOH so that a slightly alkaline solution resulted. This led to an increase in dye solubility as a result of a shift in equilibrium favoring the more water soluble basic forms of this molecule. All
subsequent concentrations of 1,4 -DHPN required during the course of this experiment were made by appropriate dilution of this stock solution.

Absorption measurements were preformed using a Perkin-Elmer Lambda 4 $\mathrm{uv} / \mathrm{vis}$ spectrophotometer. Concentrations studies were made using a pH 7.0 phosphate buffer over a concentration range extending form 10 mM to $1 \mu \mathrm{M}$. The pH studies were performed, using the appropriate phosphate buffer in the 6.0-8.0 pH range, at a dye concentration of 0.1 mM .

### 3.3.2. Results and Discussion

Absorbance spectra of 1,4-DHPN at pH 7.32 (Fig. 3.5) show a broad absorption band centered around a maximum at 381 nm with a full width at half maximum (FWHM) of approximately 57 nm , as measured with a dye concentration of 0.1 mM . For concentrations of this compound exceeding 0.5 mM , spectral distortion and a departure form the linear dependence of absorbance upon concentration, as we.ld be anticipated from the application of Beer's law, were observed (Fig. 3.6). The behavior, in these concentrated solutions, may result from both a high degree of light scattering and differential absorption by a variety of polymeric forms of 1,4 -DHPN which may be present. The value of the molar extinction coefficient ( $\varepsilon$ ), at 381 nm and pH 7.32 , can be determined from the slope of the concentration versus absorbance curve for concentrations of 1,4 -DHPN below 0.5 mM . This value was found to be $\varepsilon=5885 \mathrm{M}^{-1} \mathrm{~cm}^{-1}$. The pH sensitivity of the molar extinction coefficient (Fig. 3.7) makes accurate experimental determinations of dye concentration difficult.

As the pH of a solution of 1,4-DHPN is made more alkaline, its absorbance band shifts to longer wavelengths (Fig. 3.8). This can more easily be shown by plotting the peak absorption wavelength against solution pH (Fig. 3.9). This
calibration curve, easily fit by a cubic equation, can be used to determine solution pH if the peak absorption wavelength is known.

### 3.4. Fluorescence Measurements

### 3.4.1. Lifetime

### 3.4.1.1. Procedure

Phosphate buffers extending over the pH range $5.0-9.0$, in 0.5 pH units, were prepared according to the procedure described in Section 3.3.1. Dilutions were made from a 10 mM stock solution of $1,4-$ DHPN so that the final dye concentration in each sample was $1.0 \mu \mathrm{M}$.

Lifetime measurements were preformed in the lab of Dr. Enrico Gratton using an I.S.S. GREG1 multifrequency cross-correlation phase and modulation fluorometer. Excitation was achieved using the 325 nm line of a helium-cadmium laser at a power level of approximately 1 mW . The full emission spectrum of each sample was collected and analyzed for phase shift and demodulation with respect to a known reference. Measurements were taken at temperatures of 31.4, 41.2 and $50.8^{\circ} \mathrm{C}$, and frequencies of up to 160 MHz were employed for an effective resolution of 6.25 ns . Lifetimes were computed from this data using a nonlinear least-squares method for minimizing the value of Chi square in the fitting of a multiexponential decay.

### 3.4.1.2. Results and Discussion

Since the decay time of a fluorophore is often sensitive to its environment, the effect of both pH and temperature on the fluorescence lifetime of 1,4 -DHPN was studied in order to determine the feasibility of using this parameter to quantitatively determine the pH of a given sample.

Lifetime data may be obtained using either pulsed or harmonic methods. In the pulse method, a brief pulse of light is used for excitation and the time-dependent decay of fluorescence intensity is measured. The lifetime of each component can then be directly determined by using a multiple exponential fit to the decay curve.

In the harmonic method, the sample is excited with sinusoidally modulated light. The fluorescence emission will also be sinusoidally modulated at the same frequency but, since the lifetime of the excited state is finite, it will be delayed in phase and less modulated than the excitation. Measurement of the phase delay and the modulation ratio provides independent determinations of the fluorescence lifetime [34]. Furthermore, if a wide range of modulation frequencies are available, the information contained in the phase and modulation values is equivalent to that obtained from the more direct pulse measurements [36]. The primary advantages of the phase-modulation method are the ability to measure short nanosecond lifetimes with good resolution and the speed (seconds) with which the measurement can be carried out [28].

A plot of phase delay, as a function of modulation frequency (Fig. 3.10), shows that this delay is a function of both pH and modulation frequency. The phase delay increases as the modulation frequency is increased, and is greater at more acidic values of pH for all frequencies measured. A similar behavior (Fig. 3.11) is expected and found when the modulation ratio is plotted as a function of modulation frequency and pH . The amount of demodulation increases as the modulation frequency increases and is greater at more acidic values of pH for all frequencies considered. These data suggest that the measured lifetime of this fluorophore is strongly pH dependent and decreases with increasing deprotonation of the sample.This is in sharp contrast to 2 -naphthol, the most widely studied model of a simple fluorophore, and has been attributed to the inductive effect of the cyano groups [14].

After application of nonlinear least-squares curve fitting techniques to the experimentally acquired modulation and phase information, it was found that a double exponential fit was adequate to describe the data over the pH range considered. The largest fraction ( $>90 \%$ at all pH values) had lifetimes that ranged from 8.73 ns at pH 5.12 to 5.48 ns at pH 8.87 (Fig. 3.12). This fraction can best be thought of as a lifetime averaged over a weighted contribution from all species present at any given pH , rather than the lifetime of any individual species. Experiments conducted over a more extensive pH range, as well as use of higher modulation frequencies, would be required in order to determine species specific lifetimes. A second minor fraction ( $<10 \%$ at all pH values) with an average lifetime of 10 ps was found to be required to give an acceptable fit to the experimental data. Due to the small fractional contribution from this component, as well as its short highly variable lifetime, it is reasonable to conclude that this component is most likely an artifact introduced to account for errors in the experimental measurements. Given the above limitations on this study, the lifetimes determined using this technique are in reasonable agreement with those reported in the literature [14].

While the above experimental data does not allow accurate determination of species specific lifetimes, it does provide the basis for a fast and accurate method of pH determination in the physiological range. If the modulation/phase versus frequency curve for pH 8.0 is subtracted from the corresponding curve for pH 6.0 , the maximum change in modulation/phase with pH occurs at a frequency of 30 MHz . By plotting the modulation/phase at this frequency versus pH (Fig. 3.13) an accurate calibration curve can be constructed. An error analysis of the experimental data, acquired with the previously described laboratory based system, suggests that a precision of $\pm 0.04 \mathrm{pH}$ units can be achieved throughout a $6.0-8.0 \mathrm{pH}$
range at a temperature of $31.4^{\circ} \mathrm{C}$, using either modulation ratio or phase information.

The effect of temperature on both the modulation ratio and phase delay at 30 MHz was explored using solutions of 1,4-DHPN dissolved in phosphate buffers of known pH when measured at $25^{\circ} \mathrm{C}$. From this data (Figs. 3.14 and 3.15), it appears that lifetimes of solutions of $1,4-$ DHPN at acidic values of pH are more temperature sensitive then those that are slightly alkaline. Furthermore, temperature sensitivity reaches a minimum at pH 7 and in addition, seems to undergo a sign reversal at this point. A possible explanation for this phenomena is that each species of 1,4-DHPN has a distinct temperature sensitivity with regard to its fluorescence lifetime. As the pH of the environment changes, so does the fractional distribution of the different forms of this dye. The measured temperature sensitivity at a given pH would then be a weighted average of the sensitivities of the particular species involved. If this explanation is correct, it suggests that neutral species interacts with its surroundings in a very different manner from that of the dianion, especially in the excited state.

The change in hydrogen ion activity with temperature has been reported to be -0.0028 pH units $/{ }^{\circ} \mathrm{C}$ for standard phosphate buffers [97]. From the experimental data, an average pH error of approximately +0.05 pH units $/{ }^{\circ} \mathrm{C}$ can be calculated over a pH range of $6-8$. This error is smallest at pH 7 and increases the greater the deviation from this value in either direction. Since this pH error is more than an order of magnitude greater than that introduced by the buffer, as well as being of the opposite sign, it can be seen that a significant error can be introduced when using this technique to measure pH over an extended temperature range. Since acid dissociation constants are known to be temperature dependent, this is the most likely explanation for these temperature dependent pH errors.

### 3.4.2. Excitation and Emission

### 3.4.2.1. Procedure

Phosphate buffer of approximately 305 mOsm was prepared spanning the $6.0-8.0 \mathrm{pH}$ range, in 0.5 unit steps, according to the procedure detailed in Section 3.3.1. Each of these samples was prepared from a 10 mM stock solution of $1,4-$ DHPN, achieving a final dye concentration of $10 \mu \mathrm{M}$. All measurements were preformed using an early prototype of an I.S.S. microprocessor controlled photoncounting spectrofluorometer [35]. An Apple IIE microcomputer was used for both instrument control and data acquisition.

All excitation spectra were obtained by scanning the excitation monochromator form 250 to 450 nm with the emission monochromator fixed at 455 nm . The emission spectra were obtained by scanning the emission monochromator from 400 to 600 nm with the excitation monochromator fixed at 387 nm . The fixed wavelengths were chosen to be near to either emission or excitation maxima, as determined from a prescan of the pH 7.0 sample. Spectra were taken at temperatures of $30.6,39.0$, and $49.4^{\circ} \mathrm{C}$.

### 3.4.2.2. Results and Discussion

Like most pH sensitive fluorophores, 1,4-DHPN can be used to measure pH by calculating the fluorescence emission ratio at a fixed wavelength, after a sequential excitation at two different wavelengths. For maximum sensitivity however, it is best if both excitation wavelengths chosen are sensitive to sample pH . Furthermore, the emission change should be in opposite directions following excitation at each of these wavelengths. Where this is not possible, one wavelength can be chosen at an isosbestic point with some resultant loss in sensitivity.

The overall excitation spectra of 1,4-DHPN is strongly dependent on sample pH (Fig. 3.16). When a difference spectrum ( $\mathrm{pH} 6.0-\mathrm{pH} 8.0$ ) is taken (Fig. 3.17), it
is observed that the maximal change in emission, upon excitation, occurs at wavelengths of 365 and 414 nm . The ratio of these two wavelengths (Fig. 3.18), as a function of pH , gives a calibration curve which, given the limited number of data points, can be fit exactly by a fourth order polynomial.

When the temperature of the sample is increased (Fig. 3.19), the emission decreases slightly for excitation at 365 nm and increases slightly for excitation at 414 nm . The peak of the excitation spectrum, however, is seen to remain fixed across the temperature range considered in this study. This results in an increased ratio ( $414 \mathrm{~nm} / 365 \mathrm{~nm}$ ) with temperature (Fig. 3.20), for any given pH value. The degree of increase in the value of the ratio appears to be a linear function of temperature. The slope of this function is generally found to increase with pH over the pH interval examined in this study. The average amount of error introduced into pH measurements, by this technique, is again found to be on the order of $\pm 0.05$ pH units $/{ }^{\circ} \mathrm{C}$.

By calculating the temperature difference spectrum ( $30.6-49.4^{\circ} \mathrm{C}$ ), it is seen that maximal temperature sensitivity tends to occur at approximately the same wavelengths as maximal pH sensitivity (Figure 3.21). This result precludes use of this dye for simultaneous pH independent temperature measurements by simple selection of another pair of excitation wavelengths.

Unlike most pH sensitive fluorophores, this indicator dye can also be used to measure pH by using dual wavelength emission spectroscopy following a broadband excitation. The peak emission is shifted to longer wavelengths as the sample pH increases in alkalinity (Fig. 3.22). A difference spectra ( $\mathrm{pH} 6.0-\mathrm{pH} 8.0$ ) shows that the maximal change with pH occurs at wavelengths of 435 nm and 486 nm , respectively (Fig. 3.23). By plotting the ratio of emission wavelengths ( 486 nm $/ 435 \mathrm{~nm}$ ) as a function of pH , a calibration curve can be obtained which can be exactly fit, given the sparse data, by a fourth order polynomial (Fig. 3.24).

As in the case of the excitation spectra, the wavelength of peak emission remains constant with temperature (Fig. 3.25), while the intensities of fluorescence at 435 and 486 nm , as well as their ratio ( $486 \mathrm{~nm} / 435 \mathrm{~nm}$ ), appear to be a function of temperature. Unlike the temperature behavior of the emission spectrum, however, the functional form of the relationship among the variables of ratio, temperature and pH appears to be considerably more complex (Fig. 3.26).

The behavior of the emission ratio, with respect to temperature, parallels closely the behavior of fluorescence lifetime. A minimum sensitivity is seen to occur at pH 7.0 , with increased temperature sensitivity the greater the departure from this pH in either direction. Also, as in the lifetime experiments, the temperature sensitivity changes sign as neutral pH is approached. At acidic pH , the ratio increases with increasing temperature, while it tends to decrease, with increasing temperature, at more alkaline pH . The average pH error introduced by temperature over range of this study was $\pm 0.02 \mathrm{pH}$ units $/{ }^{\circ} \mathrm{C}$. In a more restricted pH range ( $\mathrm{pH} 6.5-\mathrm{pH} 7.5$ ), a maximum error of $\pm 0.03 \mathrm{pH}$ units $/{ }^{\circ} \mathrm{C}$ occurred at the pH extremes. Unfortunately, as was discussed above with regard to the excitation spectrum, the emission wavelengths chosen in order to optimize pH sensitivity are also those that possess a high degree of temperature sensitivity (Fig. 3.27).

The behavior of the dual emission ratios as a function of temperature seem to be largely dependent upon the 486 nm component. This wavelength is approximately that of the peak emission wavelength of the dianionic species. A plot of the temperature difference spectrum ( $49.4-30.6^{\circ} \mathrm{C}$ ) illustrates the change in sign and magnitude of the 486 nm fluorescence as the sample pH is altered (Fig. 3.28). This result is probably due to the effect of temperature on this systems complex equilibria. The sample pH determines the fractional contribution of each equilibrium constant to the overall temperature sensitivity of the resultant ratio.

### 3.5. Conclusions

The studies described above suggest that the pH sensitive fluorophore $1,4-$ DHPN can be employed in many different ways as the basis of a pH sensitive optrode for use in the physiological pH and temperature ranges. Each approach presents advantages as well as disadvantages with respect to instrument design, accuracy, and precision.

A simple absorption technique could be used near the peak absorption wavelength. Changes in pH could then be measured by measuring the changes in absorbance at this wavelength. This technique, while affording simplicity in instrument design, requires the use of complex sensor geometry for acceptable measurement sensitivity. Furthermore, changes in indicator concentration will manifest themselves as pH measurement errors. On the other hand absorbance peak detection while not suffering form concentration dependent problems, still requires complex optrode designs, as well as sophisticated optical and electronic processing in order to be able to detect the absolute position of the peak wavelength. Use of a scanning diode array may make this approach more feasible. It should be noted however, that this technique exhibits extremely poor pH sensitivity in the physiological ( $\mathrm{pH} 6.5-\mathrm{pH} 7.5$ ) range.

The measurement of pH using the fluorescence lifetimes of 1,4-DHPN has several attractive features, especially when frequency domain techniques are employed using either phase delay or modulation ratio at a fixed frequency. The measurement is fast, has good precision, and minimum temperature sensitivity near physiological pH . In addition, the maximum pH sensitivity occurs at slightly acidic values of pH ( $\mathrm{pH} 6-7$ ). This is most likely due to the larger difference in lifetimes observed between the neutral and anionic species ( 4 ns ) than between the anionic and dianionic species ( 2.3 ns ). This increased sensitivity at acidic values of pH make this technique attractive for hyperthermia studies since the microenvironment of
most tumors is slightly acidic. Disadvantages of this technique are its dependence on sophisticated and expensive instrumentation. Extremely stable frequency synthesizers, as well as light sources capable of being modulated at frequencies as high as 1 GHz are usually required. Another disadvantage is the complex shape of the pH calibration curve obtained using this technique. Obtaining a pH value from a measurement of phase or modulation would impose a considerable processing burden, resulting in the need for fast microprocessors in order to achieve a reasonable real time data acquisition rate.

Fluorescence excitation spectroscopy is a poor choice for a pH measurement technique when utilizing 1,4-DHPN. Its maximum pH sensitivity is in the $7-8 \mathrm{pH}$ range. However, useable sensitivity is exhibited down to pH 6 . The need for sequential excitation makes for complex instrumentation, as well as poor precision due to the nonsimultaneous acquisition of the data needed to compute the desired ratio. In addition, the large temperature sensitivity of this technique makes temperature correction necessary at almost all pH values.

Dual wavelength emission spectroscopy appears to offer the most promise for use in making physiological pH measurements with a 1,4-DHPN based optrode. Sensitivity in the $7-8 \mathrm{pH}$ range is $\geq 1$, with sufficient sensitivity down to pH 6 . Simple instrumentation can be employed to acquire simultaneous dual wavelength intensity data, following a single broadband excitation. This makes the computed ratio relatively independent of parameter fluctuations in either the measurement system or the sample. Furthermore, a single fiber can be used for both excitation and emission. In addition, the temperature sensitivity is the smallest of all the techniques studied across the physiological $6.5-7.5 \mathrm{pH}$ range, with an absolute minimum temperature sensitivity occurring at pH 7 .

For the above reasons, a pH sensitive optrode using 1,4-DHPN was constructed and instrumentation was built in order to take advantage of its unique dual emission wavelength characteristics.

## 1,4-dihydroxyphthalonitrile



Figure 3.1. Three forms of the fluorescent probe 1,4-DHPN in solution.


Figure 3.2. Oxidation of the fluorescent probe 1,4-DHPN in solution.


Figure 3.3. Uncorrected titration curve for a 2 mM solution of 1,4-DHPN in a 50/50 ethanol/water solvent.


Figure 3.4. Blank corrected titration curve for a 2 mM solution of 1,4 -DHPN in a $50 / 50$ ethanol/water solvent.


Figure 3.5. Concentration dependence of the absorption spectrum of 1,4-DHPN in 305 mOsm phosphate buffer at $\mathrm{pH}=7.32$.


Figure 3.6. Concentration dependence of the absorbance of 1,4-DHPN measured at a peak wavelength of 381 nm , in 305 mOsm phosphate buffer at $\mathrm{pH}=7.32$.


Figure 3.7. The pH dependence of the 381 nm extinction coefficient of a 0.1 mM solution of 1,4-DHPN.


Figure 3.8. The pH dependence of the absorption spectrum of a 0.1 mM solution of 1,4-DHPN.


Figure 3.9. The pH dependence of the peak absorption wavelength of a 0.1 mM solution of 1,4-DHPN.


Figure 3.10. Frequency dependence of the fluorescence phase of a 0.002 mM solution of 1,4-DHPN in 305 mOsm phosphate buffer as a function of pH .


Figure 3.11. Frequency dependence of the fluorescence modulation ratio of a 0.002 mM solution of $1,4-$ DHPN in 305 mOsm phosphate buffer as a function of pH .


Figure 3.12. Calculated fluorescence lifetime of a 0.002 mM solution of $1,4-$ DHPN in 305 mOsm phosphate buffer as a function of pH .


Figure 3.13. Fluorescence modulation ratio and phase of a 0.002 mM solution of $1,4-$ DHPN in 305 mOsm phosphate buffer as a function of pH . Measurements were taken at a frequency of 30 MHz .


Figure 3.14. Temperature dependence of the fluorescence phase of a 0.001 mM solution of 1,4-DHPN in 305 mOsm phosphate buffer as a function of pH . Measurements were taken at a frequency of 30 MHz .


Figure 3.15. Temperature dependence of the fluorescence modulation ratio of a 0.001 mM solution of $1,4-$ DHPN in 305 mOsm phosphate buffer as a function of pH . Measurements were taken at a frequency of 30 MHz .


Figure 3.16. The pH dependence of the excitation spectrum of a 0.001 mM solution of 1,4-DHPN in 305 mOsm phosphate buffer. Emission measurements were taken at 455 nm .


Figure 3.17. Excitation pH difference spectrum ( $\mathrm{pH} 6-8$ ) of a 0.001 mM solution of 1,4-DHPN in 305 mM phosphate buffer. Emission measurements were taken at 455 nm .


Figure 3.18. The pH dependence of the excitation ratio ( $414 \mathrm{~nm} / 365 \mathrm{~nm}$ ) of a 0.001 mM solution of 1,4 -DHPN in 305 mOsm phosphate buffer. Emission measurements were taken at 455 nm .


Figure 3.19. Temperature dependence of the excitation spectra of a 0.001 mM solution of $1,4-$ DHPN in 305 mOsm phosphate buffer at pH 7.0 . Emission measurements were taken at 455 nm .


Figure 3.20. Temperature dependence of the excitation ratio ( $414 \mathrm{~nm} / 365 \mathrm{~nm}$ ) of a 0.001 mM solution of $1,4-\mathrm{DHPN}$ in 305 mOsm phosphate buffer as a function of pH . Emission measurements were taken at 455 nm .


Figure 3.21. The pH and temperature excitation difference spectrums for a 0.001 mM solution of 1,4-DHPN in 305 mOsm phosphate buffer. Emission measurements were taken at 455 nm . All temperatures were measured in degrees Celsius.


Figure 3.22. The pH dependence of the emission spectrum of a 0.001 mM solution of 1,4-DHPN in 305 mOsm phosphate buffer. Excitation was at 387 nm .


Figure 3.23. Emission pH difference spectrum ( $\mathrm{pH} 6-8$ ) of a 0.001 mM solution of 1,4-DHPN in 305 mOsm phosphate buffer. Excitation was at 387 nm .


Figure 3.24. The pH dependence of the emission ratio ( $486 \mathrm{~nm} / 435 \mathrm{~nm}$ ) of a 0.001 mM solution of 1,4-DHPN in 305 mOsm phosphate buffer. Excitation was at 387 nm.


Figure 3.25. Temperature dependence of the emission spectrum of a 0.001 mM solution of 1,4-DHPN in 305 mOsm phosphate buffer at pH 7.0. Excitation was at 387 nm.


Figure 3.26. Temperature dependence of the emission ratio ( $486 \mathrm{~nm} / 435 \mathrm{~nm}$ ) of a 0.001 mM solution of 1,4-DHPN in 305 mOsm phosphate buffer as a function of pH . Excitation was at 387 nm .


Figure 3.27. The pH and temperature emission difference spectrums for a 0.001 mM solution of 1,4-DHPN in 305 mOsm phosphate buffer. Excitation was at 387 nm . All temperatures were measured in degrees Celsius.


Figure 3.28. The pH dependence of the temperature emission difference spectrums of a 0.001 mM solution of 1,4-DHPN in 305 mOsm phosphate buffer. Excitation was at 387 nm . All temperatures were measured in degrees Celsius.

## CHAPTER 4 INSTRUMENT DESIGN

### 4.1. Introduction

Instrumentation was designed to enable rapid optical measurement of pH , via the fluorophore 1,4-DHPN, using the dual emission ratio technique discussed in Chapter 3. Frequently in spectroscopic work, the short-term stability of the lightsource intensity limits measurement speed and accuracy [80]. Time dependent concentration changes may also introduce errors in the measurement of the desired reaction. The dynamic ratio technique, if correctly applied, will yield increases in both measurement speed and accuracy.

A gated integrator approach, using a pulsed source and analog detection, was chosen for simultaneous signal acquisition on each channel. In general, the choice between gated integration (boxcar averaging) and phase sensitive detection (lock-in detection) is based on the time behavior of the signal [72]. If the signal is fixed frequency and has a duty cycle greater than or equal to $50 \%$, lock-in detection is the best technique to use since the noise collected during a long gating time can easily swamp the signal. On the other hand, if the signal has a duty cycle less than $50 \%$, such as the pulse from a flashlamp, then a gated integrator can be utilized to detect the signal only when it is present. Consequently all noise occurring outside of the gating interval is rejected.

Analog detection was chosen on the basis of the expected signal level. At very low light intensities, photon counting works well since the input discriminator tends to reduce front end noise. At high light intensities, analog detection works better because analog inputs are less prone to saturation than those of a counter. For moderate light levels such as are inherent in the present application, analog detection combined with a front end optimized for low noise operation seems to be
the best choice. A low noise front end is necessary since this input noise adds to the ever present Poisson-counting distribution noise to degrade the overall signal-tonoise ratio (SNR) of the system.

Two additional features were incorporated into the instrument design in an attempt to improve the overall SNR. The first is the addition of an optional hardware module preceding the integrator stage. This module consists of a high Q , fixed frequency bandpass filter, followed by an rms-to-dc converter. The addition of this module has several advantages. First, it is extremely effective in reducing ac and dc background due to stray light reaching the detector. Second, broadband random noise is reduced and discrete electrical frequencies, such as pickup from 60 Hz sources, are eliminated with the appropriate selection of center frequency. Third, dc offset errors such as those introduced by photodetector dark current and temperature effects are eliminated. Finally, detection of either a fundamental or harmonic of the gating pulse, allows processing to take place in a higher frequency region where $1 / \mathrm{F}$ noise from detectors and amplifiers is minimal. The only disadvantage in the use of this module is that sensitivity is reduced due to the loss of signal intensity caused by the reduction in overall harmonic content. Detection and processing of a single frequency is valid, since it can be demonstrated that all the photons in the light pulse contribute to the measurements at each frequency [3].

The second option for further increasing the SNR is the capability of preforming signal averaging in software. In its simplest form, signal averaging is just the summation of signals in memory. If the noise is truly random, it will have a mean value of zero and a constant rms value. After $n$ summations, the rms signal amplitude will have increased by $n$, while the noise will have increased by only $n^{1 / 2}$. Thus, the SNR at any point is improved by a factor of $\mathrm{n}^{1 / 2}$ [101].

### 4.2. Overall System Design

The overall design (Fig. 4.1) was kept as modular as possible, both internally and externally, for easy system access and modification. The main subsystems have been placed into four separate enclosures. These subsystems are the excitation source, optics, photodetectors, and electronics. Connections between subsystems are achieved either by means of $500 \mu \mathrm{~m}$ core optical fibers, for optical subsystems, or shielded coaxial cable terminated with BNC connectors for electrical subsystems. Control over the total system state and parameter settings is achieved using an Apple 2E microcomputer.

Step index optical fibers (Polymicro Technologies, FHP 500/600/630) were used for this study. These fibers have a synthetic silica core of $500 \mu \mathrm{~m}$ diameter encased within a doped silica cladding containing a thin polyimide outer buffer coating (Fig. 4.2). These materials give a durable low fluorescence optical fiber of approximately $630 \mu \mathrm{~m}$ in total diameter, a numerical aperture of 0.22 resulting in a full angle acceptance cone of $25.4 \mathrm{C}^{\circ}$, and an attenuation of less than $50 \mathrm{~dB} / \mathrm{km}$ at a wavelength of 400 nm . These fibers are placed within black Teflon sleeving (Alpha Corp.), in order to ease handling and reduce coupling of stray light into the fiber, before being terminated in LFR (Amp, Inc.) style optical connectors of the appropriate size.

The general operation of this instrument is as follows. Light energy from the flashlamp subsystem is bandpass filtered in the near uv and focused into the input of an optical fiber (Fig. 4.3) using a filter and a series of lenses contained within this module. Both the flash rate and intensity are under computer control. The distal end of this fiber connects to the direct pH input of the optical subsystem. Contained within the first of two modules inside of this subsystem is a dichroic beam splitter (LPF 1) which reflects the short excitation wavelengths ( $<420 \mathrm{~nm}$ ) toward the sensor output. These wavelengths excite the 1,4 -DHPN present near the tip of the
sensor fiber and the backscattered fluorescence is returned by the same fiber to the dichroic beamsplitter. Since the fluorescence is red shifted, relative to the excitation wavelengths, transmission of fluorescence ( $>430 \mathrm{~nm}$ ) out of the first module and into the second module takes place. Within the second module, an achromatic beamsplitter (BS) and narrowband interference filters (F1 and F2) isolate the two wavelengths of interest and couple them into two separate output fibers.

The other end of these output fibers can either be connected to photomultipliers contained within the photodetector module or to PIN photodiodes contained within the electronics module. In the case of photomultipliers, the electrical output signals are coupled into the external inputs of the electronics module by means of coaxial cables. The optical amplification of each tube can be separately controlled by the computer. For PIN photodiodes, the electrical signal is directly connected to the front end electronics within the electronics module.

The electronics module (Fig. 4.4) contains two identical electronic channels which simultaneously process each optical signal. The current from each photodetector is first converted to a voltage. This voltage is then either integrated directly or patched, via coaxial cables, into and out of a second optional module for bandpass filtering and rms-to-dc conversion before integrating. The integrated pulse is then passed to a sample and hold stage, after appropriate gain equalization and polarity correction.

Output of the sample and hold circuitry of each channel is routed out of the electronics subsystem and is sequentially multiplexed onto the input of an A/D converter card contained within the Apple 2E microcomputer. The digitized signals are then further processed, displayed and stored.

A detailed description of each subsystem is given below.

### 4.3. Excitation Source

A bulb type Xenon flashlamp (EG\&G FX-198), operating at a low repetition rate, is used as a synchronous source of excitation. The efficiency of this type of flashlamp can be as high as $15 \%$ when operated with $0.1-1.0$ joules of input energy. Light output is very rich in blue and ultraviolet and is made up of a line structure superimposed on top of a strong high temperature continuum [1]. Approximately $11 \%$ of the optical energy is confined in the $300-400 \mathrm{~nm}$ range of interest. Flash duration, measured at one-third peak amplitude, can be calculated to be approximately $3.2 \mu \mathrm{~s}$ for the value of the discharge capacitor selected. Flash delays after triggering of less than $2.0 \mu \mathrm{~s}$, timing jitter of less than 200 ns , and pulse output amplitude variation of less than $5 \%$ have all been reported for this type of flashlamp.

The output of this lamp (Fig. 4.5) is collimated by a parabolic reflector (Melles Griot, 02 RPM 006) with a focal length of 10.2 mm adjusted to coincide with the electrodes of the flashlamp. The near parallel beam from this source is then condensed and shaped by a pair of plano-convex lenses (Melles Griot, 01 LPX 281 and 01 LPX 108), with respective focal lengths of 200 and 50 mm , arranged in a confocal fashion with the larger lens closest to the flashlamp and plane surfaces facing each other. This gives a parallel light beam of smaller diameter which is then filtered by a shortpass filter (Dell Optics Co.) with an average transmission of $25 \%$ from $340-380 \mathrm{~nm}$ and blocked to an average O.D. of 6 from $430-1000 \mathrm{~nm}$. This parallel near uv component is then focused onto the launch end of the output optical fiber via an aspheric condensing lens (Melles Griot, 01 LAG 000) with a focal length of 8.5 mm . The aspheric surface minimizes spherical aberration, allowing a much shorter focal length for a given diameter than a spherical lens of equal spherical aberration. This results in low f-numbers, thus maximizing the collecting area of the lens and concentrating more energy into the fiber located at its focus. All
flashlamp optics are mounted in a brass cylinder with a diameter of 15 cm and an overall length of approximately 69 cm . Provision is made to allow slight adjustment of the long focal length lens with respect to the flashlamp in order to maximize energy coupling into the optical fiber.

The flashlamp electronics (Fig. 4.6) are housed in a $23 \times 15 \times 13 \mathrm{~cm}$ aluminum box secured to one end of the brass cylinder. This enclosure contains a 24 V at 2.5 A low voltage supply (Power One ), a 300-1500 V programmable HV supply (EG\&G PS-350), a flashlamp Lite-Pac trigger transformer (EG\&G FYD506), and two $1 \mu \mathrm{~F}$ energy storage capacitors, connected in parallel, and rated at 2 kV . The high voltage is programmed over its full range by means of a 2-10 VDC external reference supplied by the Apple 2 E microcomputer via an 8 bit $\mathrm{D} / \mathrm{A}$ card (Applied Engineering) residing in slot 3. Since the flashlamps maximum average power is rated at 10 W in free air, this puts restrictions on the flash frequency/output energy combinations, and thus indirectly on the maximum high voltage for any selected flash frequency (Figs. 4.7 and 4.8). Software allows the output energy of the flashlamp to be selected in the interval 0.1-2.25 joules. The computer then sets the high voltage to the appropriate value for the required energy and computes the highest permissible flash frequency selected to the closest 4 Hz increment, in the range $4-100 \mathrm{~Hz}$. Manual override of this frequency selection is also allowed. The flashlamp trigger signal is also supplied by the computer system via a single output line of a parallel interface card (John Bell Engineering, Inc.), residing in slot 2 . This drive capability of this TTL level is increased by means of a line driver residing on the interface board at the back of the electronics subsystem. Both the flashlamp reference voltage and the boosted trigger signal are available via BNC connectors mounted on the front panel of the electronics subsystem.

### 4.4. Optical Subsystems

The optical subsystem is composed of two separate modules: a dedicated sensor module and an optical detector module. These modules are similar in overall design and were fabricated out of black Deldrin to decrease the weight of the optical system and to prevent stray reflections from degrading performance. Each module (Fig. 4.9) currently measures approximately $5 \times 5 \times 10 \mathrm{~cm}$ with a T-shaped 2.54 cm diameter optical path machined in the plastic. Each of the three openings on a module interfaces with a standard fiber optic connector through a specially designed adaptor. On the back side of each adaptor is a holder which can accommodate a 12 mm diameter lens and up to two 12.5 mm normal incidence filters. The distance between lens and fiber can be manually adjusted. This allows alignment of the lens so that the amount of light coupled to the optical fiber can be maximized. Currently an aspheric condensing lens with a focal length of 8.5 mm (Melles Griot, 01 LAG 000) is being used.

Along the long axis of the module is a moveable mount for holding either a 19 mm diameter longpass filter or beamsplitter at a $45^{\circ}$ incidence angle. All optics in this system, except for the lenses, were custom made by Dell Optics Company (North Bergen, N.J.) according to provided specifications. Optical coupling into and out of each module is accomplished by means of the optical fibers previously described. The modular design of such a system allows compartmentalization of the optical assembly. This makes the system extremely flexible since changes within a module can be easily made and their effect on total system performance can readily be measured. Furthermore, all modules can be made optically unique in spite of being mechanically similar.This enables complex systems to be built for performing specialized measurements.

The operation of the optical subsystem is as follows. Energy from the excitation module is focused by an aspheric condensing lens into the input of the
sensor module. This module consists of a dichroic beam splitter that is used to reflect excitation wavelengths below 420 nm toward the sensing fiber. Fluorescence from this fiber at wavelengths longer than 430 nm are transmitted by this dichroic device and coupled into the exit fiber. The broadband fluorescence output from this fiber then enters the detector module where it is again beamsplit, but this time by a 50/50 achromatic beamsplitter. This beamsplitter divides the fluorescence signal into two components. Each component is passed through a 10 nm narrowband interference filter centered at either 434 or 488 nm . Each of these filters has a peak transmission in the passband of approximately $70 \%$ and is blocked outside of this band to an O.D. of 4. The output from these filters is then coupled out of the optical subsystem and into the appropriate photodetectors using the usual optical fibers.

### 4.5. Photodetectors

Either PIN photodiodes or photomultiplier tubes may be selected as the photodetectors in this system. The PIN photodiodes make use of an extra high resistance (intrinsic) I layer between the P and N layers. This increases the width of the depletion region resulting in lower junction capacitance. As a result, the speed of this device is much faster than a conventional PN photodiode. Extremely low noise and low dark current are also characteristics of PIN devices.

The PIN photodiodes used in this system (Hamamatsu, S1722-01) are uv enhanced silicon photodiodes. Both diodes are housed in TO-8 style packages and are located within the electronics subsystem. They have a surface area of 13.2 $\mathrm{mm}^{2}$, a radiant sensitivity of $0.20 \mathrm{~A} / \mathrm{W}$ at 450 nm ., a typical dark current of 30 nA , a junction capacitance of approximately 12 pF , and a shunt resistance of approximately $100 \mathrm{M} \Omega$. The noise characteristics of this device are set by the sum of the thermal noise, caused by the shunt resistance, and the shot noise resulting from the dark current and photocurrent. The signal characteristics are determined
by the junction capacitance, effective surface area, and radiant sensitivity. Taken together, these parameters determine the SNR of the device and the theoretical lower limit of light detection by this photodetector. The minimum signal power on a detector that produces an rms SNR of 1 is defined as the noise equivalent power (NEP) for that detector. The lower the NEP, the more sensitive is the detector. For the photodiodes used in this system the NEP has been found to be approximately $100 \mathrm{fW} / \mathrm{NHz}$, when measured at the wavelength of peak radiant sensitivity. The actual lower limit of detectability, for such a device in an actual circuit, will be primarily determined by the characteristics of the optical signal and the performance of the front end circuitry [98]. Both photodiodes are operated in the current mode using a low noise operational amplifier that effectively holds the photodiode voltage at zero. This is an optimal configuration from the standpoints of response linearity and noise generation.

In order to detect very low intensity optical signals with acceptable SNR, the photomultiplier subsystem should be used (Fig. 4.10). This subsystem contains two 13 mm diameter head-on photomultiplier tubes (Hamamatsu, R1463-01), two regulated programmable HV supplies (Hamamatsu, C1309-04), two voltage divider socket assemblies (Hamamatsu, E849-35), and one 15 V at 1 A low voltage supply (Power One). These components are housed in an aluminum enclosure to which optical input and electrical input and output connections can be made. The gain of the photomultiplier tubes is controlled by the Apple 2E microcomputer, via the programmable HV supplies, using the 8 bit D/A card discussed in Section 4.3. The software controlled $0-10 \mathrm{~V}$ outputs of this $\mathrm{D} / \mathrm{A}$ card are available via the front panel of the electronics subsystem. This voltage is divided down, within the photomultiplier enclosure, allowing high voltages ranging from (-190)-(-1100) V to be generated. This allows current gains ranging from 4 to over $2 \times 10^{6}$ using this subsystem.

The NEP of photomultiplier tubes is limited primarily by dark current and its associated noise. The photomultipliers used in this system have a multialkali cathode, an anode radiant sensitivity of $5.1 \times 10^{4} \mathrm{~A} / \mathrm{W}$ at 420 nm , and a typical dark current of 10 nA . Using a typical current gain of $1.0 \times 10^{6}$, a NEP of approximately $1 \mathrm{fW} / \mathrm{NHz}$ can be calculated. Thus, the NEP of the photomultipliers is at least 100 times lower than that of the PIN photodiodes. This is due to the dynode chain amplification of the photomultipliers being essentially noiseless. As a consequence, the amplified shot noise of the photocathode becomes the primary noise component [22]. As in the case with PIN photodiodes, the front end circuitry plays a crucial role in determining the actual lower limit of detection.

### 4.6. Electronics Subsystem

The computer controlled electronic subsystem has two principle components, the main module and the optional narrowband filter module. Each of these modules is electrically shielded by being placed inside of a cast aluminum box. These two boxes are then mounted inside of a standard rack mount enclosure. All inputs and outputs are routed from the appropriate aluminum box to the front panel of the rack mount enclosure via BNC connectors and shielded cables. All digital control lines are interfaced between the electronic subsystem and the computer by means of an interface board mounted on the rear panel of the electronics subsystem (Appendix A). The final analog outputs of the electronic subsystem are multiplexed onto the 12 bit $\mathrm{A} / \mathrm{D}$ converter card (Applied Engineering) residing in slot 5 of the Apple 2E microcomputer.

Two identical electrical channels are available within each module and operate as follows (Fig. 4.11). The low level signal current from either the PIN photodiodes or the photomultipliers are manually selected via a toggle switch (SW1) on the front panel of the electronics enclosure. These currents are directed
into a low noise, low drift, (Burr-Brown, OPA101BM) operational amplifier (U1) configured as a current to voltage converter. The gain of this I/V stage is software selectable in four fixed decade steps. The frequency response of this stage is set at 16 kHz , via lowpass filtering, regardless of the gain setting. This value was chosen as a compromise between minimizing noise and preserving signal characteristics so as to lessen the necessary integration time. A noise analysis of this front end [98] shows that the input voltage noise of the operational amplifier ( 8 $\mathrm{nV} / \sqrt{\mathrm{Hz}}$ ) undergoes a capacitive gain at higher frequencies and becomes the major component of front end output noise. The next most important noise component has been found to be the Johnson noise of the feedback resistors. Output offsets of this front end amplifier can be zeroed via a front panel potentiometer (R5) which forms part of a current injection circuit that has been optimized for low drift.

The output of the previous stage can either be coupled directly to a software controlled Miller integrator stage (U2), or fed out of this module as input to the optional narrowband filter module. The Miller integrator starts integration immediately after the flashlamp trigger, using an operator selected integration period. This stage is implemented using an operational amplifier (Burr-Brown, 3528AM) selected for its low bias current characteristic. This enables one to use long integration times without amplifier saturation. This also allows for long holding times, such as those used in multiple integrations, without objectionable voltage droop due to leakage. Use of a low leakage current polystyrene integrating capacitor also adds to the stability of this stage.

The output of the integrator stage is coupled to a variable gain stage (U3) built around a standard operational amplifier (LM741C). The configuration of this amplifier is switch selectable (SW2) between inverting and noninverting to accommodate both photodiode and photomultiplier type detectors. Four feedback positions are available, each with its own separately adjustable potentiometer. These
are selected in tandem with the switchable I/V ranges, thus allowing separate calibration of each $I / V$ range as well as the ability to balance the electrical gain between channels. The amount of gain can be varied over approximately one decade.

The output of the variable gain stage is coupled to a digitally controlled sample/hold amplifier (U4) (Burr-Brown, SHC298AM) and routed out of the electronics subsystem via a BNC connector mounted onto the rear panel.

The dc level from the sample/hold amplifier is multiplexed onto one of 16 channels of a 12 bit A/D card (Applied Engineering). This card features a $25 \mu \mathrm{~s}$ conversion time and resides in slot 5 of the Apple 2E computer controller.

The narrowband filter module (Fig. 4.12) consists of a very low noise, high Q, Butterworth bandpass filter (U1) (A.P. Circuit Corporation, APB-6-Q1224 Hz ), centered around 24 Hz with a 2 Hz bandwidth. The output of this stage is coupled through a high pass filter to a low level DC/RMS converter (U2) (Analog Devices, AD636KD). A unity gain inverting amplifier (U3) (LM741C) is used to achieve correct signal polarity for coupling back into the integrator stage of the main module. A front panel offset adjustment potentiometer (R9) can be used to zero the overall system output when this module is in operation. A frequency of 24 Hz was chosen based upon spectral measurement of noise density. This frequency is a compromise between minimum noise and optimum pulse frequency. When this module is used, signal-to-noise ratio can be improved by taking advantage of the synchronous bandlimited nature of the signal over the uncorrelated wideband background noise.

### 4.7. Software

The computer control, data acquisition, and output display software is written in combined Applesoft BASIC and 6502 assembly language, running under
the Apple DOS 3.3 operating system, on an Apple 2E microcomputer [26]. During system initialization the DOS 3.3 operating system is relocated to high RAM, thus freeing an additional 10.5 Kbytes of low RAM for use in storing the necessary programs and variables. Relocation is accomplished through use of Memory Management System (MMS) software (On-Line Systems, Coarsegold, CA) executed by the HELLO program during system boot (Appendix B). Upon termination, the HELLO program executes the STARTUP program. This program sets the memory range to be used by Applesoft BASIC, loads the required machine language programs, and then loads and executes the main BASIC program.

The BASIC program (Appendix C) is approximately 1000 lines, with calls to assembly language routines (Appendix D) which occupy approximately 4 Kbytes of memory. In addition, approximately 2 Kbytes of memory are set aside for data storage and communication between high and low level language routines.

Assembly language is used, due to its speed, for both overall system control and data acquisition. Assembly language however is extremely inefficient for communication involving display screen usage and for implementing complex numerical procedures. For these tasks, interpreted Applesoft BASIC was chosen since this language is readily available in the system ROM and allows interactive testing and debugging. Assembly language source code was stored in a sequential text file format and edited, assembled, and debugged using utilities available in the DOS Programmer's Tool Kit Volume II (Apple Computer, Inc.).

Overall system integration is achieved by means of a coordinated interaction between 6502 assembly routines and BASIC routines. The BASIC language supports lower level process control through the use of CALL statements which execute assembly subroutines, PEEK statements which read memory locations, and POKE statements which write memory locations. Thus, control of overall
program execution is possible in BASIC by use of shared memory locations to exchange both variables and data.

The assembly language routines are incorporated into four distinct phases which are called from the main BASIC program at the appropriate point. Phase 1 initializes system hardware and sets all system acquisition parameters with the exception of system gain. Phase 2 sets the overall system gain using either a manual or automatic calibration routine. Phase 3 is responsible for measuring initial system offsets which are subsequently used to correct raw measurement data. Phase 4 initiates the actual measurement of corrected data using a specific acquisition procedure.

The BASIC user interface is detailed in flowchart form (Fig. 4.13). It is comprised of eight distinct sections: initialization, offset adjustment, flashlamp control parameters, integrator control parameters, photomultiplier control parameters, system gain determination, measurement routines, and printer routines. The first six sections configure the system for actual operation and execute in a sequential fashion. Within the final two sections, the user can select from any of the available options.

The measurement and printer routine software supports fixed time point single sample sweep measurements, multiple sample (signal averaging) sweep measurements, variable time interval data file acquisition, fixed time point data file acquisition, statistical processing of data files, and hardcopy output of both data and statistical results.

In single sweep measurements, 10 successive data points are taken. For each data point, a voltage proportion to the integrated signal on each channel is displayed. The ratio of these voltages ( $\mathrm{CH} 0 / \mathrm{CH} 1$ ) is also displayed, as well as the percent change in ratio for each successive data point. In addition, the mean and standard deviation of each of these parameters, for the entire data set, is computed
and presented. The multiple sweep measurement routine works in an identical fashion, except that each data point displayed is a user selected average of from (1)(100) separate data points.

In the data file mode, the user can choose either to acquire 200 single data points in succession at a particular point in time, or acquire up to 200 single or averaged data points separated by a fixed time interval. This interval is selectable in units of either hours, minutes, or seconds, with values ranging from (1)-(32,767). Precision of this time interval is insured by use of hardware generated timing interrupts from a Timemaster II H.O. clock card (Applied Engineering) residing in slot 4 of the Apple 2E microcomputer. Regardless of the mode selected, eight parameters are stored to disk. These include baseline levels for channel zero and channel one as well as raw measurement data for each channel, the ratio ( $\mathrm{CH} / \mathrm{CH1}$ ), and standard deviations for channel zero, channel one, and the ratio. In the case of fixed time point data files, baseline levels will be equal for all data points in the file and all standard deviations will be set equal to zero.

The statistical processing routine reports minimum and maximum values for each of the parameters in the data file, along with the mean, variance, and standard deviation associated with those parameters. Hardcopy printed output of this information is supported via an Epson FX-100 printer attached to the computer system by means of a Parallel Pro interface card (Applied Engineering) located in slot 1.


Figure 4.1. Modular overview of the entire fluorescence ratio based pH measurement system.

## Fiber Silica


Figure 4.2. Diagram of a typical step index silica core optical fiber.


Figure 4.3. Optical diagram of the fluorescence ratio based pH measurement system.

Figure 4.4. Single channel electrical block diagram of the fluorescence ratio based pH measurement system.


Figure 4.5. Optical diagram of the flashlamp excitation subsystem.



Figure 4.7. High voltage versus maximum safe flashlamp firing rate.


Figure 4.8. Input energy versus maximum safe flashlamp firing rate.

Figure 4.9. Diagram of a typical optical module.

Figure 4.10. Schematic diagram of the photomultiplier electronics.

Figure 4.11. Schematic diagram of the wideband electronics module of a single electrical channel.


Figure 4.12. Schematic diagram of the narrowband electronics module of a single electrical channel.


Figure 4.13. Functional flowchart of the the measurement system software (modified from Ehlert, 1988).




Figure 4.13. Continued.


Figure 4.13. Continued.


Figure 4.13. Continued.


Figure 4.13. Continued.


Figure 4.13. Continued.


Figure 4.13. Continued.


Figure 4.13. Continued.

## CHAPTER 5 <br> INSTRUMENT PERFORMANCE

### 5.1. Introduction

The overall performance of the optical pH measurement system and its suitability for making in vivo pH measurements under hyperthermic conditions were investigated.

The performance of individual subsystems were first evaluated in order to determine how well each achieved its specific design goal. The information obtained from these studies points out the strengths and weaknesses of the selected design scheme. These details should prove to be of value in optimizing subsystem performance for later improvements of this measurement system.

Next, performance evaluation of the overall measurement system using a single optical fiber immersed in solutions of free dye at various concentrations and values of pH , subjected to different temperatures was conducted. These measurements allow determination of system performance in differing sample environments. From this data, pH measurement errors introduced by temperature and pH changes, such as those occurring under hyperthermic conditions, can be determined. Knowledge of errors in pH measurement, resulting from inadequate signal intensity due to low concentrations of fluorophore, are of importance in designing a suitable pH sensing optrode. In addition, the impact of various software controlled system parameters on the pH measurement data were studied. This information will aid in selecting the optimum instrument settings needed in order to achieve a specified measurement criteria.

Design and fabrication of pH sensitive optrodes were considered from the viewpoints of size, stability, sensitivity, response time, and ease of production. An
important component of the sensor design is the choice of a semipermeable membrane. Several membranes were investigated, both in sheet and tubular form. The membranes were evaluated for their ability to easily exchange hydrogen ions, while at the same time restricting leakage of $1,4-$ DHPN. The size of the actual optrode was kept as small as practically possible to minimize diffusion limited response time. This allows in vivo use with minimal physiological perturbation.

Since none of the membranes tested had adequate differential permeability, encapsulation of 1,4-DHPN by means of liposomes was studied as a method of limiting the leakage of 1,4 -DHPN from the optrode. The effect of this encapsulation technique on the kinetics of hydrogen ion transport, and thus sensor response time, was also studied.

Finally, animal studies were conducted in order to evaluate the suitability of the entire optical pH measurement system for in vivo use. Measurements of pH in the blood, leg muscle, and tumor of white rats were made under stable and varying physiological conditions. In this fashion, the steady state and dynamic characteristics of this in vivo measurement system could be investigated.

### 5.2. Subsystem Evaluation

### 5.2.1. Excitation Source

### 5.2.1.1. Procedure

The flashlamp energy storage capacitators were charged to numerous values of energy ranging from 0.0-2.25 joules by controlling their maximum charging voltage via the Apple 2E microcomputer. Optical energy was measured at several points internal and external to the excitation subsystem using a radiometer (Photodyne Inc., Model 66XLA) in an energy mode. The sensing head of this instrument (Photodyne Inc., Model 420) is set to read actual energy or power
when used at a wavelength of 400 nm . Appropriate correction factors for other wavelengths were read from a supplied calibration table and applied to data measurements whenever appropriate. Reported data values were the average of 101 flashes taken at a frequency of 8 Hz .

### 5.2.1.2. Results and Discussion

A plot of electrical energy input versus optical energy output (Fig. 5.1) shows a nonlinear relationship. Both efficiency and output sensitivity are highest at low input energies. The decrease in efficiency seen at higher energies is due to excessive $I^{2}$ R heating losses generated by the high peak currents associated with these energies. From this graji., overall conversion efficiency is seen to be very low, on the order of $0.002 \%$ for 1 joule of electrical input energy. This is several orders of magnitude lower than the $15 \%$ figure appearing in the manufacturers literature. Several explanations for this result are possible. Electrical losses resulting from energy dissipation in the storage capacitators and associated wiring could play a minor role in this reduced efficiency. More importantly, the measurement error could be large due to limitations in the instrumentation used and the measurement technique employed. Flashlamp energy output measurements were obtained by placing the sensor head as close to the top of the glass flashlamp bulb as was physically possible. Due to the large area of the source, the sensor could only measure a fraction of the optical energy being produced. An integrating sphere would be necessary for a more accurate determination. Furthermore, the broadband nature of the source introduces uncorrected wavelength dependent measurement errors. A spectroradiometer would be needed to accurately determine its spectral energy distribution.

The optical energy present in the excitation subsystem output fiber, with and without the shortpass filter, was subsequently measured (Fig. 5.2) and an estimate
of the peak available power was derived based on a $t_{1 / 3}=3.2 \mu \mathrm{~s}$ pulse width. An input energy of 1 joule at a frequency of 8 Hz was used in these determinations. Broadband optical energy was coupled into the output fiber by the flashlamp optics with an efficiency of approximately $14 \%$. This coupling drops to approximately $0.3 \%$ when the shortpass filter is used to restrict the spectral distribution of the optical output. This figure is consistent with the fact that approximately $4 \%$ of the flashlamp energy is contained within the bandpass of this filter which has a bandpass transmission of approximately $25 \%$. The overall electrical-to-optical conversion efficiency for this resultant narrowband excitation source is calculated from these measurements to be approximately $8 \times 10^{-6} \%$. Clearly, while this is a workable means of obtaining the desired optical excitation, it is far from being efficient.

### 5.2.2. Optics

### 5.2.2.1. Procedure

The optical attenuation properties of each optical module, in the optical subsystem, was measured using the optical energy from the flashlamp excitation source. The flashlamp was operated with 1 joule of input energy at a frequency of 8 Hz . Optical energy was measured using the Photodyne radiometer in an energy mode.

The shortpass filtered optic $A$ source was first coupled into the input of the sensor module and the energy output from the sensor fiber was measured. From this measurement, the efficiency of input coupling to the excitation fiber can be determined.

Next, the shortpass filter was removed from the excitation source and broadband optical energy was directed into the sensor fiber connector. Optical energy measurements were taken at the output of the sensor module, at the 488 nm
and 434 nm outputs of the detector module, and at the respective front panel optical connectors. From these measurements, the attenuation in the return fluorescence path can be determined.

### 5.2.2.2. Results and Discussion

The efficiency of excitation source coupling into the sensor fiber is calculated to be slightly greater than $4 \%$. Thus, starting with 1 joule ( $312,500 \mathrm{~W}$ peak) of electrical energy at the flashlamp (Fig. 5.2), only 3.7 nJ ( 1.2 mW peak) of restricted wavelength excitation energy is available at the sensor tip.

A study of attenuation in the fluorescence return path (Fig. 5.3) indicates that approximately $25 \%$ of the return signal present in the sensor fiber is coupled out of the first module. Since only about $82 \%$ of the broadband input signal is at an appropriate wavelength for reflection out of this module, actual efficiency can be calculated to be closer to about $30 \%$.

If both narrowband interference filters are removed from the optical detector module, the coupling efficiency from input to either module output is on the order of $6 \%$. This efficiency declines to approximately $0.1 \%$ when the narrowband interference filters are restored. Since each filter has approximately $70 \%$ transmission within its passband, and only $2.8 \%$ of the flashlamp energy is reportedly within each of these bands, the above $0.1 \%$ efficiency measured with these filters is consistent with the design parameters. For a narrowband light source concentrated within the bandpass of either of these interference filters, the overall efficiency of the detector module was determined to be approximately $4 \%$. If this same narrowband source is input to the sensor fiber, the overall return efficiency for the filtered wavelengths is on the order of $1 \%$.

An additional attenuation of approximately $50 \%$ is contributed by the use of the front panel optical connectors. Each optical connector contributes
approximately 1.5 dB to overall signal attenuation. Thus, at the front panel connectors, the optical signal energy is on the order of $0.5 \%$ of that present for the same signal at the sensor tip.

### 5.2.3. Photodetectors

### 5.2.3.1. Procedure

Photodetector responsivity (V/W) was measured at the output of the front end current to voltage converter (I/V) using an optical source consisting of a blue LED (Siemens, LDB5410), with a peak wavelength of 480 nm ; the intensity of this source was varied between 0.0 and 6.5 nW by application of a constant dc voltage to the LED. The transresistance gain of the I/V stage was fixed at $1 \times 10^{7}$ throughout the course of this study. Optical power was measured using a radiometer (Photodyne, Model 88XLC) with an optical head (Photodyne, Model 420) calibrated to display actual power when used at a wavelength of 400 nm . An appropriate wavelength dependent correction factor was applied to the displayed power readings in order to obtain the actual power at the selected wavelength. Noise measurements were obtained at the output of the I/V stage using a wideband ac rms voltmeter (Hewlett Packard, Model 7478A).

### 5.2.3.2. Results and Discussion

The dc output voltage of the front end $I / V$ stage was plotted as a function of the optical power applied to the detector. Four different photomultiplier gains ranging from ( $1 \times 10^{3}$ ) $-\left(1 \times 10^{6}\right)$ were used (Fig. 5.4). The response of the photodiodes was also measured (Fig. 5.5). The optical responsivity in V/W was determined from the slope of a regression fit to each of these curves. Using the slope of these curves and the measured ac rms noise level at the I/V output for each gain level, the optical power needed to achieve a SNR=1 at this measurement point
was calculated (Fig. 5.6). The NEP for each gain was also calculated and plotted by dividing the previous results by the square root of the $I / V$ bandwidth $(16 \mathrm{kHz})$.

The NEP measured at the $I / V$ output for the PIN photodiodes was $1.6 \times 10^{-}$ 12. This is only a factor of 6 worse than the typical NEP quoted by the manufacturer. These diodes are capable of detecting 0.2 nW of 480 nm optical power with a $S N R=1$ given the current front end design.

For photomultipliers, a NEP of $3.9 \times 10^{-16}$ can be measured at the front end for a current amplification of $1 \times 106$. This allows detection of 50 fW with a SNR=1. Thus, it is seen that when operated near maximum gain the photomultipliers are approximately 4000 times more sensitive than the PIN photodiodes at the wavelengths of interest.

As in the case with photodiodes, the measured NEP at the I/V output is again very close to the calculated value for the detector alone. This is a good indication that the detector noise, rather than the front end noise, is the major limitation for the detection of very low light levels with this system.

### 5.2.4. Electronics

### 5.2.4.1. Procedure

The electronics subsystem was calibrated using photomultiplier tubes for an equal gain of $1 \times 10^{11} \mathrm{~V} / \mathrm{W}$ on either channel. This gain figure was obtained via the internal calibration potentiometers, with the photomultiplier current gain set at 10,000 , the $\mathrm{I} / \mathrm{V}$ transresistance gain set at $1 \times 10^{7}$, and the integration time set at 100 $\mu \mathrm{s}$. Other ranges were calibrated to their appropriate gains relative to this standardized range.

Drift and stability tests were run on the system using a photomultiplier current gain of 10,000 , a transresistance $I / V$ gain of $1 \times 10^{7}$, and an integration time of $100 \mu \mathrm{~s}$. The optical inputs to the photomultipliers were capped off to
eliminate all light and the system offsets were tracked from power-up, using the primary electronics module, with and without the optional narrowband electronic filter module.

The optical signals were obtained and measured using the same procedures outlined in Section 5.2.3.1. The total system responsivity curves were also generated and fitted in an analogous fashion. Again, a front end gain of $1 \times 10^{7}$ was used with the integration time set at $25 \mu \mathrm{~s}$.

### 5.2.4.2. Results and Discussion

With the primary wideband electronics module in operation, substantial system drift occurs throughout the first hour of operation (Fig. 5.7). This drift is approximately -16 mV for channel 0 and +8 mV for channel 1 . For strong signal levels the ratio error introduced by this drift is minor. However, under weak signal conditions, or when photodiodes are used, insufficient warm up time could introduce substantial measurement errors. If the narrowband filter module is used (Fig. 5.8), warm up drift is practically eliminated. In either case, adequate system stability is achieved once a steady state operating point has been reached.

Total system output voltage was plotted against optical input power at 480 nm for both photomultipliers (Fig. 5.9) and photodiodes (Fig. 5.10). From the slope of these curves total system responsivity (V/W) can be determined. These values will be used later in determining the actual fluorescence energy levels arriving at the detectors for a specific concentration of 1,4-DHPN located near the sensor tip.

### 5.2.5. Conclusion

With a flashlamp input energy of 1 joule ( $312,500 \mathrm{~W}$ ), only $82.1 \mathrm{~nJ}(26 \mathrm{~mW})$ of excitation at the appropriate wavelength can be obtained. Most of the energy loss appears to result from inefficient electrical-to-optical energy conversion.

However, as previously mentioned, problems with the optical measurement technique employed make this conclusion subject to question. In any event, the optical excitation source used in this measurement system, while functional, is extremely inefficient. Further research and development effort is needed in order to produce a more compact, efficient, and powerful source of optical excitation.

The optical modules used in this system appear to work properly. However, further work to develop more efficient optical coupling could substantially improve overall performance. Only about 4\% of the optical energy coupled into the sensor module appears at the tip of the sensor fiber. Thus, for a 1 joule ( $312,500 \mathrm{~W}$ ) electrical input to the flashlamp, only $3.7 \mathrm{~nJ}(1.2 \mathrm{~mW})$ is available to excite the pH dependent fluorophore. Furthermore, only about $1 \%$ of the fluorescence signal at any wavelength is available at the output connectors of the detector module.

The photodetectors used in this system appear to be capable of achieving close to their theoretical limit of sensitivity when measured at the output of the front end I/V stage. This indicates that both the photodetector subassembly and the electronics subassembly have acceptable levels of performance. Stability of the electronics subassembly is good and use of narrowband filtering virtually eliminate the problem of baseline drift.

### 5.3. Measurement System Evaluation

### 5.3.1. Solution Studies

### 5.3.1.1. Procedure

A simultaneous potentiometric and optical titration was performed using a 2 mM solution of $2 \mathrm{mM} 1,4-$ DHPN in a $50 / 50$ ethanol/water solvent. This mixture was titrated against an NaOH solution, whose concentration had previously been determined to be 17.3 mM . The pH during the entire course of the titration was
monitored using a Beckman Model 71 pH meter. The fluorescence signals at 488 and 434 nm ; as well as their ratio, were monitored using a single silica optical fiber. This fiber was attached to the sensor input of the optical pH measurement system. The system was configured for 8 Hz operation using 1 joule of input energy. Integration time was set for $100 \mu \mathrm{~s}$, with an $\mathrm{I} / \mathrm{V}$ transresistance gain of 1 x $10^{7}$ and a photomultiplier current gain of 4000 on each channel. Each data point reflects the average of 100 samples, with 2 minutes of stabilization time allowed between data points.

Sensitivity studies were conducted by varying the concentration of $1,4-$ DHPN from $10 \mathrm{mM}-5 \mu \mathrm{M}$, at pH values of $6.0,7.0$, and 8.0 , respectively. Instrument settings were essentially the same as those used during titration, except that the photomultiplier current gain was readjusted as necessary. Each data point obtained represents the average of 200 samples. In the wideband versus narrowband electronic processing studies, the frequency and energy settings of the flashlamp were changed to 24 Hz at 0.4 joules as required by the fixed frequency design of the narrowband filters. All other system parameters were unchanged.

Studies of the effect of temperature on ratio measurements were carried out at temperatures from $10-70^{\circ} \mathrm{C}$. Measurements were obtained at pH values of 5.94, 7.08, and 8.22. Instrument settings and measurement parameters were identical to those used in the sensitivity studies. The concentration of $1,4-$ DHPN was fixed at $100 \mu \mathrm{M}$ throughout these measurements.

The effect of flashlamp frequency and input energy on measured ratios was investigated using a $100 \mu \mathrm{M}$ solution of 1,4 -DHPN at pH values of $5.93,6.45,7.15$, 7.70, and 8.17. For the energy dependent studies, input energies ranging from 2.25-0.5 joules were used at a frequency of 8 Hz . For the frequency dependent studies, frequencies ranging from (64)-(8) Hz were used with 1 joule of input energy. Integration time was fixed at $100 \mu \mathrm{~s}$ and $\mathrm{I} / \mathrm{V}$ transresistance gain was fixed
at $1 \times 10^{7}$ throughout these measurements. Again, 200 samples were averaged for each reported data point.

An investigation of the effect of integration time and sample averaging on the standard deviation of measured ratios was conducted. For the integration studies, a $100 \mu \mathrm{M}$ solution of $1,4-\mathrm{DHPN}$ at a pH of 7 was used. The flashlamp was provided with 1 joule of input energy at a frequency of 8 Hz . A transresistance gain of 1 x 107 was selected for the front end I/V stage. Integration times of $25,100,400$, and $1000 \mu$ s were used and the usual 200 samples per data point were obtained. For the sample averaging studies, identically $\mathrm{I} / \mathrm{V}$ gain settings, dye concentration, pH , and flashlamp parameters were chosen. Integration time was fixed at $100 \mu \mathrm{~s}$ and averages and standard deviations of 10,100 , and 1000 samples were obtained.

Finally, the effect of optical fiber characteristics on instrument sensitivity, as measured by the standard deviation of measured ratios, was studied. The instrument settings and measurement parameters were similar to those used in the integration time study, with the integration time fixed at $100 \mu \mathrm{~s}$. A $50 \mu \mathrm{M}$ solution of $1,4-$ DHPN, at pH 7 , was used in testing all optical fibers. The optical fibers tested included the standard FHP 500/600/630 silica fiber with a flat polished termination, as well as terminations with integral spherical lenses having apparent focal lengths of 0.8 and 2.0 mm . In addition, a smaller FHP 320/385/415 fiber, with a flat polished termination, was evaluated.

### 5.3.1.2. Results and Discussion

An acid base titration of 1,4-DHPN was performed and a simultaneous plot of both potentiometrically determined pH and optical fluorescence ratio $(488 / 434)$, versus volume of titrant added, was constructed (Fig. 5.11). The shape of the potentiometrically determined titration curve is typical of that obtained when a weak diprotic acid is titrated with a strong base. Details of this titration curve have
already been discussed in Section 3.2.2. The optical fluorescence ratio curve has a very different shape. It is a continuous smooth curve with no apparent inflection points. A plot of this ratio versus pH can be nicely fit by a fourth order polynomial over the pH 5 to 9 range (Fig. 5.12). Mathematically simpler fits, such as those obtained with a single exponential, give acceptable accuracy, especially when a narrower pH range is under investigation. Since this is the case in the physiological studies to be conducted, a single exponential fit will be used for system calibration in the sensor evaluation and animal testing that follow.

Plotting the measured fluorescence intensity at both 488 and 434 nm against the volume of added titrant (Fig. 5.13), as well as against pH (Fig. 5.14), shows the fluorescence intensity at 434 nm decreases rapidly as the pH is made more alkaline. Since this wavelength is primarily associated with the monoanionic form of the fluorophore, such a result is not unexpected. However, since a wavelength of 488 nm is primarily associated with the dianionic form of the fluorophore, the fluorescence intensity measured at this wavelength should be expected to increase. The slight decrease actually observed is probably due to the fact that the expected increase in fluorescence intensity at this wavelength, due to the increased concentration of the dianionic species, is offset by the volume dilution effects of the added titrant. Thus, the apparent decrease in intensity is most likely an artifact introduced by the experimental procedure. The impact of volume dilution effects on the fluorescence ratio measured in this experiment should also be considered if an accurate calibration curve is desired.

Studies of the relationship between fluorescence intensity and concentration at a pH of 7 , indicate that fluorescence intensity at either 488 or 434 nm is an approximately linear function, for concentrations of 1,4-DHPN at less than about 1 mM (Fig. 5.15). For concentrations of fluorophore below 1 mM , the fluorescence ratio (488/434) appears to be independent of concentration (Fig. 5.16). However,
the standard deviation of the measured ratio is found to decrease approximately proportional to the cubed root of the concentration (Fig. 5.17). This result is somewhat puzzling since it would normally be expected that the standard deviation would decrease as the square root of the number of particles present in solution and thus, as the square root of the concentration. This discrepancy can either be an artifact introduced by attempting to fit a theoretical curve to a limited number of data points, or it could be real and related to apparent changes in solution volume being sensed at the probe tip. Further theoretical and experimental work is required in order to resolve this incongruity.

For any concentration the pH error will always be greater the more alkaline the solution. This is a direct consequence of the small amount of 434 nm fluorescence present in these solutions. For the present measurement system operated with an input energy of 1 joule, a flash rate of 8 Hz , an integration time of $100 \mu \mathrm{~s}$, and an $\mathrm{I} / \mathrm{V}$ transresistance gain of $1 \times 10^{7}$, a 1 mM solution of $1,4-$ DHPN is required to obtain a pH measurement with a standard deviation of less than 0.1 pH unit. This result was obtained at neutral pH , when 200 samples were taken. Given the above set of operating parameters, this is close to the minimum pH error that can be obtained with this system regardless of concentration. A plot of pH errors for other concentrations and pH values can easily be constructed (Fig. 5.18).

From the total system output voltage for the 488 or 434 nm channel, using a 10 mM concentration of 1,4-DHPN, the system sensitivity curves (Figs. 5.9 and 5.10) can be used to calculate the narrowband optical energy incident upon the photodetectors. Energies of $7.6 \mathrm{fJ}(2.4 \mathrm{nW})$ at 488 nm and $5.54 \mathrm{fJ}(1.7 \mathrm{nW})$ at 434 nm can be calculated in this fashion. These intensities are high enough to be resolved by the system photodiodes, but an SNR of approximately 10 is all that can
be expected. If lower concentrations of fluorophore are to be used or a better SNR is desired, photomultipliers must be employed.

If the efficiency of the fluorescence return path is assumed to be on the order of $0.5 \%$ at each measured wavelength, then approximately $1 \mathrm{pJ}(300 \mathrm{nW})$ of narrowband optical energy is returned through the sensor fiber within each detected emission band. Assuming a sensor tip excitation energy of 3.7 nJ ( 1.2 mW ), approximately $0.03 \%$ of this energy is recovered as fluorescence emission within each filtered bandpass. The spectral and spatial distribution of the fluorescence emission accounts for the majority of the observed energy reduction.

Studies were conducted to determine the effect of narrowband filtering on measurement accuracy and precision. For the same concentration of fluorophore, system output levels with the narrowband filter module in place were reduced to approximately $5 \%$ of those obtained using the wideband module. This results in a factor of 20 reduction in available sensitivity with equivalent accuracy (Fig. 5.19). However, even though the measured system output was substantially reduced, the standard deviation of the ratio was virtually identical for a given concentration of dye (Fig. 5.20). These results suggest that the use of narrowband filtering may result in increased measurement precision when adequate signal strengths are obtainable. The ability of such filtering to eliminate dc offsets, such as those caused by stray light and component drift, may offer additional measurement advantages in some cases.

A detailed study of the effect of temperature on optically measured pH values was conducted using this measurement system. The intensity of fluorescence at both 488 nm (Fig. 5.21) and 434 nm (Fig. 5.22) decreased with increasing temperature. However, the rate of change of fluorescence with temperature is seen to increase, with increasing alkalinity, for 488 nm fluorescence and decrease for 434 nm fluorescence. This effect, as previously noted, tends to minimizes the the
temperature dependent pH measurement error near neutral pH (Fig. 5.23). Over the 6.5 to 7.5 pH range, a maximum standard deviation of 0.007 pH units $/{ }^{\circ} \mathrm{C}$ was calculated. By calibrating this system at $40^{\circ} \mathrm{C}$, hyperthermia studies over a $\pm 10^{\circ} \mathrm{C}$ temperature range can be conducted with a maximum standard pH error of only $\pm$ 0.07 units. The ratio error introduced by temperature dependent effects, over the above temperature and pH ranges, is smaller than the intrinsic system measurement error for the measurement conditions utilized in this experiment (Fig. 5.24). A ratio standard deviation of less than 0.1 would be needed, at all of the above pH values, before temperature dependent effects would be of equivalent significance. A plot of the fluorescence ratio and its standard deviation, as a function of both pH and temperature (Fig. 5.25), clearly demonstrates an increase in temperature dependent measurement errors at alkaline pH , as well as a decrease in measurement precision with increasing temperature.

Several system parameters were evaluated for their impact on measurement precision as determined by the standard deviations of the measured fluorescence ratios. If the flashlamp firing frequency is increased, with input energy held constant, it can be seen that while the mean ratio remains essentially unaltered (Fig. 5.26), its standard deviation increases with increasing frequency and pH (Fig. 5.27). Furthermore, the standard deviation of the ratio is found to increase in proportion to the square root of the frequency (Fig. 5.28). A similar behavior of mean ratio (Fig. 5.29) and its standard deviation (Fig. 5.30) is noted if the input energy per flash is decreased at a constant flash frequency. The decrease in the standard deviation of the measured ratio again appears to be proportional to the square root of the input energy (Fig. 5.31). Thus, the best measurement precision is obtained with the lowest flashlamp frequency and highest input energy. These values however, must be consistent with both the maximum flashlamp dissipation requirement and the maximum time allotted for acquisition of the required number
of samples. Since the number of emitted photons is proportional to the flashlamp energy, and the standard deviation has been shown to decrease as the square root of the number samples (photons), it seems reasonable to expect that the standard deviation of the fluorescence ratio would decrease as the square root of the available flashlamp energy. The frequency dependence of the standard deviation is best explained by a linear relationship existing between flashlamp frequency and energy. As the flashlamp frequency is increased beyond that recommended for a given input energy, its output energy will drop due to the limited time available for recharging the discharge capacitators. Thus, in reality the frequency dependence of the standard deviation of the fluorescence ratio is really an energy dependence.

A study of the effect of sample averaging and integration time on the measured ratio yielded expected results. The standard deviation of the measured ratio decreased as the square root of the number of samples being averaged (Fig. 5.32). An integration time of $100 \mu \mathrm{~s}$ appears to be near optimum for use with this measurement system. For values much less than this number, the standard deviation increases due to the small percentage of the overall signal being integrated (Fig. 5.33). For integration times much greater than $100 \mu \mathrm{~s}$, the standard deviation of the measurement ratio increases due to a predominate contribution from the system noise.

Finally, the impact of sensor fiber variation on measurement precision was studied. While an increase in the standard deviation of the ratio was observed for a $36 \%$ reduction in fiber core diameter (Fig. 5.34), smaller differences were observed when flat ended fibers were compared with similar fibers incorporating spherical lenses at their tip. Both lensed fibers gave slightly better standard deviations than the flat ended fiber. The shorter focal length of 0.8 mm gave the best overall measurement precision. Since this focal length is closest to the radius of curvature the lens, near optimum beam collimation would be expected. The
significance of these differences can not be tested since only a single averaged data point was acquired with each fiber tested. These results do tend to suggest however, that by using a large diameter sensor fiber, with a collimated output, small but noticeable improvements in measurement precision can be achieved.

### 5.3.2. Optrode Studies

### 5.3.2.1. Procedure

For free fluorophore studies, a $100 \mu \mathrm{M}$ solution of 1,4-DHPN was prepared in a normal saline solution ( $0.9 \%$ ). The pH of this solution was adjusted to be in the 5.5 to 6.0 pH range by addition of small volumes of either NaOH or HCl . The tubing or membrane to be tested was then filled with this dye solution. The end of the tubing was tied off with thread and the entire length was immersed in a vial containing 10 ml of either normal saline, or 305 mOsm phosphate buffer, at a pH of either 7.0 or 8.0 . Both the fluorescence ratio ( $488 \mathrm{~nm} / 434 \mathrm{~nm}$ ), and the optical intensity at each wavelength was monitored as a function of time. The flashlamp was set at a frequency of 8 Hz with an input energy of 1 joule. A $100 \mu \mathrm{~s}$ integration time was used with an $I / V$ gain of $1 \times 10^{7}$. A 100 sample average was reported for each data point, requiring an acquisition time of 12.5 seconds.

For the liposome encapsulation studies, large unilamellar vesicles (LUV) were prepared from a 125 mg of a $4: 1$, by weight, mixture of dipalmitoylphosphatidylcholine (DPPC) and dipalmitoylphosphatidylglycerol (DPPG) (Avanti Polar Lipids), using the reverse evaporative phase process [58]. Varying amounts of Gramicidin A (Sigma Chemicals) were added to the lipid phase prior to preparation. This amount ranged from 0 to 2 mole percent of the total lipid present. The aqueous phase consists of a $10 \mathrm{mM} 1,4-$ DHPN dye solution. Sucrose was added to this solution ( $1.04 \mathrm{~g} / 10 \mathrm{ml}$ ) in order to achieve an approximately isosmotic concentration, and the final pH was adjusted to
approximately 7 by the addition of a small volume of NaOH . The prepared LUV were dialyzed overnight in the cold, against normal saline, in order to remove unencapsulated dye. Prior to using these LUV in sensor studies, they were diluted 1:1 with normal saline. Testing was then carried out in a manner identical to that used in the free fluorophore studies.

### 5.3.2.2. Results and Discussion

Several small diameter tubes of various materials were studied in order to measure their differential permeability properties. In order to be of value in pH sensor fabrication, such tubing should possess a large permeability to hydrogen ions while at the same time offering a low permeability to the small, 160 molecular weight, pH sensitive fluorophore. Small size is desirable in order to both maintain acceptable response time and minimize tissue injury while performing in vivo measurements. In addition, mechanical rigidity and a simple construction allow easy handling and fabrication.

The materials investigated using free dye were TFE Sub-Lite-Wall Teflon (AWG 18) (Zeus, Inc.), Nafion Perfluorinated Tubing (815X) (Perma Pure Products, Inc.), polysulfone tubing (P10-43) (Amicon Corp.), cellulose (Spectra/Por 2, MWCO 12-14K) (Spectrum Medical Industries, Inc.), and Cuprophan ( 150 PM ) (Enka-Glanzstoff, AG). All materials except cellulose and Cuprophan were commercially available in tubular form. Cellulose and Cuprophan tubes were made using a polyurethane based epoxy (Master Bond EP30DP-1), by a modification of a published procedure [74]. All tubes had an inner diameter of approximately 1 mm with walls of varying thickness. The thickness of the walls were as follows: Teflon ( $40 \mu \mathrm{~m}$ ), Nafion ( $150 \mu \mathrm{~m}$ ), polysulfone ( $1000 \mu \mathrm{~m}$ ), cellulose ( $30 \mu \mathrm{~m}$ ), and Cuprophan ( $15 \mu \mathrm{~m}$ ). The diffusive permeability of both polysulfone and Cuprophan, to a variety of different solutes, have been measured
[52]. Although no permeability data for hydrogen ions could be found, respective permeabilities to sucrose ( 343 Daltons) of $3.21 \times 10^{-4} \mathrm{~cm} / \mathrm{s}$ and $1.76 \times 10^{-4} \mathrm{~cm} / \mathrm{s}$ were measured. For smaller solutes, the permeability of both membranes increase slightly. Cuprophan appears to have a slightly higher permeability than polysulfone for solutes of less than 60 in molecular weight.

A plot of fluorescence ratio (488/434) versus time (Fig. 5.35), for free dye loaded tubes of the above material, indicates that the Teflon and Nafion tubes have very low hydrogen ion permeability as measured by a step change in external solution pH . Polysulfone has a slightly higher permeability, while both the cellulose and Cuprophan tubes show the highest permeability. Unfortunately, as indicated by the higher standard deviation of the fluorescence ratios with time, the tubes with higher hydrogen ion permeability also show substantial permeability to the selected fluorescent indicator dye. This result is clearly seen when the fluorescence intensities are plotted against time (Figs. 5.36 and 5.37). From these figures it can be concluded that dye leakage increases proportional to hydrogen ion permeability. Since Cuprophan appears to leak dye at a slightly faster rate than cellulose, its hydrogen ion permeability can also be inferred to be slightly higher. From this data, Cuprophan would appear to be the membrane of choice from a hydrogen ion permeability standpoint, however, its near total leakage of free dye with a 10 minute time period severely limits the utility of this simple sensor. Furthermore, its flaccid structure makes handling and use difficult.

The mechanical rigidity of the polysulfone tubes is a desirable feature for development of a practical sensor. The slow leakage of free dye from this tubing allowed stable pH reading for up to approximately 1 hour. However, the fairly slow response time of this tubing ( $\tau=6.1$ minutes), prohibits its use in making dynamic pH measurements. A further complication in sensing pH with this tubing is apparent when the fluorescence ratio ( $488 \mathrm{~nm} / 434 \mathrm{~nm}$ ) is plotted to its steady state value
(Fig. 5.38). The final value of the ratio is lower in the tubing than in free solution. This could either be indicative of a more acidic internal environment or a perturbation of the pK values of the sensing fluorophore. The highly acidic sulfonic acid group of this membrane is the most probable cause of this deviation.

In order to determine whether cellulosic type membranes would similarly perturb the steady state fluorescence ratio, 2.5 mm diameter semi-micro dialysis tubing (Spectrum Medical Industries, Part \# 132600) was used. This tubing was selected since it had both a small volume, so as to not exceed the capacity of the external phosphate buffer, while at the same time possessing a volume large enough to diffusion limit the leakage of 1,4-DHPN. This restricted leakage provides acceptable signal strength when the steady state ratio is achieved. Plots of both fluorescence ratio ( $488 \mathrm{~nm} / 434 \mathrm{~nm}$ ) versus time (Fig. 5.39) and pH versus time (Fig. 5.40), obtained using a micro pH electrode (Lazar Research Labs, Model PAR-146), indicate that the expected final value of both pH and fluorescence ratio are reached, as steady state is approached. Thus, it can be concluded that no detectable membrane related perturbations, of either pH or fluorophore pK values, result from cellulosic tubing.

In order to test the feasibility of a rigid Cuprophan based sensor, three different sizes of pipettes were cut so as to present a uniform cross sectional area along their entire length. At one end of the resulting cylinder, a flat Cuprophan membrane was stretched and securely attached around the circumference using a small amount of epoxy (Master Bond, EP30DP-1). These cylinders were then filled to approximately the same height with free dye solution. The optical sensor fiber was next inserted into the open end of the cylinder and secured, with epoxy, at a selected distance from the membrane. Disposable pipettes with volumes of $10 \mathrm{ml}, 1$ ml , and $50 \mu \mathrm{l}$, and respective cylinder internal diameters of $7.9 \mathrm{~mm}, 3.2 \mathrm{~mm}$, and 1.1 mm , were employed (Fig. 5.41).

Measurements of the fluorescence ratio ( $488 \mathrm{~nm} / 434 \mathrm{~nm}$ ) as a function of time were made using the previously described system settings, upon immersion of these sensors into a pH 8 phosphate buffer. The results indicate that the smaller the sensor, the more rapid the response time (Fig. 5.42), with the 1.1 mm sensor appearing to reach its correct steady state value in about 10 minutes. These results are consistent with those that would be expected in a diffusion limited sensor response. Further confirmation of this hypothesis is obtained when the distance between the membrane and the sensor fiber is varied, at a constant solution volume, using the 1.1 mm sensor (Fig. 5.43). Response varies from several minutes at a distance of 1 mm from the membrane to virtually undetectable in 30 minutes at a distance of 8 mm . The effect of varying the volume of the filling solution, at a constant probe to membrane distance, was next investigated using these same 1.1 mm sensors (Fig. 5.44). No measurable difference in response time could be determined, given the large measurement error margin introduced by rapid dye leakage from this sensor. This result again argues for diffusion limited response, as opposed to mass transfer limited response, within this sensor. Large membrane surface area is desirable, in order to insure that a sufficiently large number of hydrogen ions cross the membrane, per unit time. By keeping the solution volume low, rapid internal concentrations changes can occur with passage of only relatively few hydrogen ions through the membrane. Unfortunately however, since measurement sensitivity and SNR increase in proportion to the number of dye molecules being interrogated, a physical limit is imposed on the minimum acceptable sensor volume.

In order to gain an understanding of the importance of the Cuprophan membrane permeability on sensor response, the Nusselt number for mass transfer was computed. Specifically, this number compares the intensity of mass flux at the
membrane with the specific flux by pure molecular diffusion in a fluid layer of thickness L [96]. In this case it is given by the equation

$$
\begin{equation*}
\mathrm{Nu}=\mathrm{PL} / \mathrm{D} \tag{6.1}
\end{equation*}
$$

where
$\mathrm{P}=$ membrane permeability ( $\mathrm{cm} / \mathrm{s}$ )
$\mathrm{D}=$ molecular diffusion coefficient ( $\mathrm{cm}^{2} / \mathrm{s}$ )
$\mathrm{L}=$ thickness of fluid layer ( cm )
Nusselt numbers much greater than one imply that mass transfer is primarily diffusion limited. For Nusselt numbers much less than one, membrane permeability becomes the limiting factor. Since Cuprophan permeability data was not available for hydrogen ions, a calculation was done using sucrose as a representative small molecular weight molecule. The diffusion coefficient for sucrose, in aqueous solution, has been reported to be $4.586 \times 10^{-6} \mathrm{~cm}^{2} / \mathrm{s}$ [ 5 ]. Using this number for a diffusion coefficient and the value previously reported for Cuprophan sucrose permeability, a solution thickness of 0.26 mm gives a Nusselt number equal to unity. Thus, it can be concluded that for optical fibers placed much closer to the membrane than 0.26 mm , membrane permeability properties would dominate the response time observed. Since the fiber tip is usually placed several millimeters from the membrane in order to provide reasonable signal levels, it again becomes clear that the observed response is primarily limited by the diffusion time for hydrogen ions in solution. A value of $9.34 \times 10^{-5} \mathrm{~cm}^{2} / \mathrm{s}$ has been reported for the diffusion coefficient of hydrogen ions in solution at $25^{\circ} \mathrm{C}$ [53]. Using the simple result from the solution of the diffusion equation, that the average distance that a molecule travels varies as the square root of the elapsed time [5], it can be calculated that approximately 54 seconds are required for a hydrogen ion to diffuse to a position approximately 1 mm away from the sensor membrane. Thus, for practical
membrane to fiber distances, response times of several minutes should be expected for this type of sensor.

In order to be able to minimize the degree of tissue damage associated with in the in vivo application of this type of sensor, several smaller diameter sensors were fabricated. These were made from both 20 and $15 \mu \mathrm{l}$ capillary tubes, with respective inner diameters of 0.64 and 0.58 mm . The outer diameter of these tubes were 0.89 and 0.86 mm , respectively. Free dye response studies with these sensors (Fig. 5.45) suggest that they have slightly shorter response times, as would be expected based upon diffusional principles. However, the very rapid dye leakage from all of these capillary sensors makes a quantitative comparison difficult.

In order to slow the rapid leakage of 1,4 -DHPN from these sensors, prior encapsulation in DPPC/DPPG liposomes was tried. Even though the intrinsic permeability of liposomal membranes to hydrogen ions is several orders of magnitude higher than for other small monovalent ions (approximately $1 \times 10^{-4}$ $\mathrm{cm} / \mathrm{s}$ ), proton flux has been found be limited by the development of a diffusion potential resulting from the restricted flow of counterion currents [21]. For this reason, the small peptide antibiotic ionophore Gramicidin A was added to the lipid phase during liposome preparation. This ionophore is specific for monovalent cations and translocates them across the lipid membrane by means of a dimeric formed hydrophilic channel. Discrimination among various monovalent cations is not high, but an anomalously high permeability of the channel to hydrogen ions has been reported [43]. The kinetics of Gramicidin induced hydrogen ion permeability has been found to be an increasing function of the Gramicidin level, up to the point where approximately 8-10 Gramicidin molecules are incorporate in each vesicle [17]. At these levels, transmembrane hydrogen ion equilibration times of less than 1 ms have been reported [12].

Studies were conducted with suspensions of liposomes in order to determine both the Gramicidin levels necessary to achieve an acceptable response time, and the liposome concentration necessary to produce adequate signal intensity for proper fluorescence ratio determination. Liposomes containing Gramicidin levels of 0.0 , 0.2 , and 2.0 mole percent (of total quantity) were prepared as previously described. The initial internal pH of these liposomes was approximately 6.7 as measured by the fluorescence ratio at $t=0$. A $1 / 2$ dilution was made at this time using phosphate buffer at a pH of 7.5 . The fluorescence ratio as a function of time was recorded using the previously described measurement system. The results (Fig. 5.46) indicate first that a quantity of fluorophore, sufficient for optical detection, was encapsulated within the aqueous compartment of the LUV. Secondly, at a Gramicidin level of 2.0 mole percent, hydrogen ion equilibration time was well below the time resolution of the measurement system. In addition (Fig. 5.47) a maximum liposome dilution of approximately $1 / 2$ was necessary for accurate fluorescence ratio determination.

A study was next made using standard 2.5 mm cellulosic dialysis tubing, as described previously, in order to mimic a large scale sensor. Results (Figs. 5.48 and Fig. 5.49) indicate that even at the 0.2 mole percent Gramicidin level hydrogen ion diffusion time, and not liposomal hydrogen ion permeability, is the dominant factor limiting sensor response.

The more practical polysulfone and Cuprophan capillary sensors were next investigated by replacing the free fluorophore in the previous studies with liposome (using 2.0 mole percent Gramicidin) encapsulate fluorophore.

When a sensor was fabricated using fluorophore containing liposomes within the 1.1 mm polysulfone tubing (Fig. 5.50), it can again be noted that, in addition to a slow response time, the expected steady state ratio is never attained (Fig. 5.51). From the measured fluorescence ratio, it appears that the fluorophore
is experiencing a local environment which is slightly more acidic than the external solution. Since the fluorophore is now encapsulated and shielded from direct interaction with the polysulfone, alterations in fluorophore pK values seem an unlikely explanation. It now seems more plausible that the sulfonic acid groups of the polysulfone tubing are becoming ionized, resulting in a more acidic localized environment on the inside of this tubing.

Another explanation for the discrepancy in measured fluorescence ratios is that either the local internal liposomal environment is more acidic than its external environment, or that the negatively charged liposome is again perturbing the pK values of the fluorophore. Phase fluorometry studies of free 1,4 -DHPN and Gramicidin free liposome encapsulated fluorophore at pH 7 (Figs. 5.52 and 5.53), show that even at a liposome dilution of $1 / 10,000$ in phosphate buffer, the the apparent fluorescently determined pH of the liposomal encapsulated dye can be as much as 0.5 pH units lower than that for free dye in solution. A similar result is obtained when fluorescence ratios are used to compare free fluorophore with liposome encapsulated fluorophore (Fig. 5.54). A lower fluorescence ratio and thus, an apparently lower pH is measured for the liposomal encapsulated dye. However, pH errors resulting from the low liposomal dilutions (1/2) used in this experiment, coupled with the large standard deviation of the measured ratios preclude a quantitative analysis of the results. Since environmental variables may exert a quantitatively different effect on fluorescence emission than they exert on lifetime, the quantitative pH difference measure by the ratio technique may differ from that measured using phase or modulation variables. In any event, for accurate pH determinations such effects must be taken into account during system calibration.

In order to investigate the effect of temperatures in the normal to hyperthermic range on liposome dye retention, polysulfone based liposomal
sensors were placed in pH 7 phosphate buffer at temperatures of 25,37 , and $45^{\circ} \mathrm{C}$, respectively. The fluorescence intensities at 434 and 488 nm were then measured as a function of time and plotted (Figs. 5.55 and 5.56). These studies indicate that even though increased loss of dye occurs at higher temperatures, sufficient dye is retained to enable measurement times well in excess of 1 hour, at a temperature of $45{ }^{\circ} \mathrm{C}$.

Finally, studies were conducted using Cuprophan capillary sensors containing liposome (with 2.0 mole percent Gramicidin) encapsulated fluorophore. Both 50 and $20 \mu \mathrm{l}$ capillary tubes were filled with a $1 / 2$ dilution of liposomes and normal saline and placed in a pH 8 phosphate buffer at $t=0$. The $50 \mu \mathrm{l}(1.1 \mathrm{~mm}$ inner diameter) capillary sensor (Fig. 5.57) achieved the expected steady state value for the fluorescence ratio with an exponential time constant of 3.2 minutes (Fig. 5.58). A plot of the corresponding pH response of this sensor is also shown (Fig. 5.59). The $20 \mu \mathrm{l}$ ( 0.635 mm inner diameter) capillary sensor also reached the expected steady state ratio, and had a time constant of 3.3 minutes (Fig. 5.60). Performance of these sensors appears similar both in terms of response time and measurement error. In either case a steady state fluorescence ratio is attained within approximately 15 minutes.

### 5.3.3. Conclusion

A practical pH sensitive optical sensor was built and tested. This sensor consists of a capillary small tube containing the pH sensitive fluorophore 1,4DHPN, which was encapsulated in DPPC/DPPG liposomes containing 2.0 mole percent Gramicidin. The pH sensing end of the tube contains a thin Cuprophan membrane. This membrane allows hydrogen ions to pass into the tube, while at the same time preventing external leakage of the dye encapsulated liposomes. Response time and measurement accuracy, as well as precision, of these sensors were tested
and appear acceptable for steady state or slowly changing systems.The thermal stability of liposomal based sensors was evaluated and determined to be suitable for use during hyperthermia studies lasting in excess of 1 hour. Finally, the small size of these sensors should permit in vivo pH measurements to be made, with minimal resultant tissue injury.

### 5.4. Animal Testing

### 5.4.1. Procedure

Whole blood was obtained from male Fisher 344 rats by laceration of the brachial artery. This blood was collected in a heparinized syringe and placed at room temperature in a glass vial containing 1000 units/ml of heparin. Liposomal based pH sensors were prepared as previously described using both 1.1 mm inner diameter polysulfone tubing and a $15 \mu \mathrm{l}$ ( 0.58 mm inner diameter, 0.86 mm outer diameter) capillary tube. Blood pH was determined to be 7.24 at room temperature, using a Beckman Model 71 pH meter with a standard combination electrode. A measurement system calibration curve was determined from five data points using a 1 mM solution of $1,4-$ DHPN dissolved in 305 mOsm phosphate buffers ranging between pH 6.0 and 8.0 in 0.5 pH unit increments. The usual standard system measurement parameters were used.

### 5.4.2. Results and Discussion

The polysulfone sensor appeared to achieve a steady state with an exponential response time of 19.8 minutes (Fig. 5.61) compared to 6.1 minutes obtained during the free dye studies. Also as expected, the predicated value of the fluorescence ratio was not attained. The standard deviation and thus, the precision of these measurements was very good. This is a direct result of the large tubular volume of the sensor. At 60 minutes the pH value calculated from the fluorescence ratio was
$7.06 \pm .004$. In contrast, the $15 \mu \mathrm{l}$ Cuprophan capillary sensor reached a steady state response with an exponential time constant of 10.0 minutes (Fig. 5.62), compared to 3.2 minutes for the $20 \mu \mathrm{l}$ liposome based sensors previously discussed. The expected steady state ratio was achieved with this sensor even though the measurement precision, due to volume restrictions, was not as good. At 60 minutes a pH of $7.24 \pm .06$ could be computed from the calibration curve using the measured ratio and its standard deviation.

The most interesting result obtained from this study is that the exponential response time increased by a factor of approximately 3 , for both sensors, when going form 305 mOsm phosphate buffer to a more physiological fluid (blood). Even though the $15 \mu$ sensor was not tested directly in phosphate buffer, it is difficult to believe that it should possess a response substantially different from the $20 \mu \mathrm{l}$ sensor that was tested. In any case, a faster not slower response would be expected. One explanation for the response time differences is the development of a diffusion potential across either the sensor membrane or across the liposomal membrane. Such a potential would limit net proton flux across these membranes, and thus prevent hydrogen ion equilibrium from being reached. Since both sensor membranes are fairly permeable to small molecules and ions, a Donnan equilibrium, set up by the large charged protein molecule present in blood, may contribute to the development of this membrane potential. In fact, the negatively charged liposomes contained within the sensor, may also play a significant role in establishing this potential.

The liposomal membrane by itself is reported to be 6 to 10 orders of magnitude less permeable to other ions than it is to hydrogen [12]. This restriction is eased for most monovalent cations due to the presence of Gramicidin in the liposomal membrane. However, this membrane still remains relatively impermeable to anions and divalent cations. Here as before, the selective
permeability properties of the this membrane may provide the source of a hydrogen ion flux limiting diffusion potential.

Another explanation for the observed increase in response time may be due to the blocking effect of divalent cations on the alkali ion permeability of the Gramicidin channel. $\mathrm{Ca}^{++}$, and to a lesser extent $\mathrm{Mg}^{++}$, have been shown to reduce the conductance of the Gramicidin channel when present in concentrations of 0.11.0 M [6]. Rat blood plasma contains $2.6 \mathrm{mM} \mathrm{Ca}^{++}$and $1.1 \mathrm{mM} \mathrm{Mg}^{++}$[75]. Even though these levels are substantially less than those used in the channel blocking study reported above, they may be sufficient to account for the increase in optrode response time seen in whole blood.

### 5.4.3. Conclusion

A limited biological study of this optical pH measurement system indicates that with small liposomal based Cuprophan sensors, accurate and precise steady state pH measurements can be obtained. While in vitro testing of these sensors demonstrates an adequate response time for measuring slow pH changes, this response time rapidly degenerates for measurements in physiological fluids, such as whole blood. Further research is needed if an understanding of this phenomenon is to be obtained, and a practical solution realized.


Figure 5.1. Optical energy as a function of electrical input energy. Measurements were taken at several points in the flashlamp excitation subsystem.


Figure 5.2. Optical energy and peak optical power measured at several locations along the excitation path. EL.IN = electrical input, FL.OUT = directly in front of flashbulb, NF.OUT = at subsystem output without shortpass filter, F.OUT $=$ at subsystem output with shortpass filter, SENS.OUT = at the sensor output. Peak optical power was calculated using a duration of $3.2 \mu \mathrm{~s}$.


Figure 5.3. Optical energy and peak optical power measured at several locations along the emission path. The unfiltered flashlamp was input to the sensor port with 1 J of electrical input energy. $1=$ output of unfiltered excitation module, $2=$ output of sensor module, $3=$ unfiltered $90^{\circ}$ output of detector module, $4=$ unfiltered $0^{\circ}$ output of detector module, $5=488 \mathrm{~nm}\left(90^{\circ}\right)$ output of detector module, $6=434\left(0^{\circ}\right)$ output of detector module, $7=$ front panel 488 nm connector, $8=$ front panel 434 nm connector.


Figure 5.4. The $\mathrm{I} / \mathrm{V}$ output versus optical input power as a function of photomultiplier current gain. Excitation source was a blue LED ( 480 nm ) operated in a dc mode. An I/V transresistance gain of 10 M was used.


Figure 5.5. The I/V output versus optical input power for PIN photodiodes. Excitation source was a blue LED ( 480 nm ) operated in a dc mode. An I/V transresistance gain of 10 M was used.


PHOTODETECTOR
Figure 5.6. Minimum detectable optical power leveis at the I/V output using PIN photodiodes and photomultipliers at various current gains settings. Excitation source was a blue LED ( 480 nm ) operated in a dc mode. An I/V transresistance gain of 10 M was used.


Figure 5.7. Drift and stability of the wideband electronics module using photomultiplier tubes operated at a current gain of 10,000 . A system integration time of $100 \mu \mathrm{~s}$ and an $I / V$ transresistance gain of 10 M were used. No optical signal was
applied.


Figure 5.8. Drift and stability of the narrowband electronics module using photomultiplier tubes operated at a current gain of 10,000 . A system integration time of $100 \mu \mathrm{~s}$ and an I/V transresistance gain of 10 M were used. No optical signal was


Figure 5.9. System output voltage versus optical input power as a function of photomultiplier current gain. Excitation source was a blue LED ( 480 nm ) operated in a dc mode. An I/V transresistance gain of 10 M and an integration time of 25 $\mu \mathrm{s}$ were used.


Figure 5.10. System output voltage versus optical input power for PIN photodiodes. Excitation source was a blue LED ( 480 nm ) operated in a dc mode. An I/V transresistance gain of 10 M and an integration time of $25 \mu$ s were used.


Figure 5.11. Potentiometric pH and optical fluorescence ratio for a titration of a 2 mM solution of 1,4 -DHPN in a $50 / 50$ ethanol/water solvent. Optical system parameters were set to 1 J at 8 Hz with a $100 \mu \mathrm{~s}$ integration time. The titrant was 17 mM NaOH .


Figure 5.12. Fluorescence ratio versus pH for titration of a 2 mM solution of 1,4-DHPN in a $50 / 50$ ethanol/water solvent. A fourth order polynomial was used to fit the data.


Figure 5.13. System output at 488 nm and 434 nm , as a function of the volume of titrant added, during the titration of a 2 mM solution of $1,4-\mathrm{DHPN}$ in a $50 / 50$ ethanol/water solvent.


Figure 5.14. System output at 488 and 434 nm , as a function of pH , during the titration of a 2 mM solution of 1,4-DHPN in a $50 / 50$ ethanol/water solvent.


Figure 5.15. System output at 488 and 434 nm as a function of 1,4-DHPN concentration, in 305 mOsm phosphate buffer at pH 7.0 .


Figure 5.16. Fluorescence ratio versus concentration of 1,4-DHPN in 305 mOsm phosphate buffer at various values of pH .


Figure 5.17. A log-log plot of the standard deviation of the fluorescence ratio as a function of concentration of $1,4-$ DHPN in 305 mOsm phosphate buffer at various values of pH . The slope of these curves was determined, by regression fit, to average to 0.32 .


1,4-DHPN CONCENTRATION (M)
Figure 5.18. The standard deviation of pH as a function of the concentration of 1,4-DHPN, in 305 mOsm phosphate buffer at various values of pH , as calculated from ratio data.


Figure 5.19. The effect of wideband and narrowband processing on the fluorescence ratio versus concentration curves. All concentrations of $1,4-$ DHPN were prepared in 305 mOsm phosphate buffer at pH 7.0 .


Figure 5.20. The effect of wideband versus narrowband processing on the standard deviation of the fluorescence ratio as a function of the concentration of 1,4-DHPN in solution. All solutions were prepared in 305 mOsm phosphate buffer at pH 7.0.


Figure 5.21. Effect of temperature on the system output at 488 nm as a function of pH . All solution were prepared in 305 mOsm phosphate buffer at a 1,4 -DHPN concentration of 0.1 mM . Slopes tend to increase as the pH is made more alkaline.


Figure 5.22. Effect of temperature on the system output at 434 nm as a function of pH . All solution were prepared in 305 mOsm phosphate buffer at a $1,4-$ DHPN concentration of 0.1 mM . Slopes tend to decrease as the pH is made more alkaline.


Figure 5.23. Calculated standard deviation of pH measurements $/{ }^{\circ} \mathrm{C}$ as a function of pH . Note minimum near pH 7.0 . All measurements were taken using a 0.1 mM concentration of 1,4-DHPN dissolved in 305 mOsm phosphate buffer. System parameter of 1 J at 8 Hz and an integration time of $100 \mu \mathrm{~s}$ were used.


Figure 5.24. Fluorescence ratio as a function of pH at various temperatures. All measurements were taken using a 0.1 mM concentration of $1,4-\mathrm{DHPN}$ dissolved in 305 mOsm phosphate buffer. System parameters of 1 J at 8 Hz and an integration time of $100 \mu \mathrm{~s}$ were used. Note that the temperature induced measurement error is less than the intrinsic measurement error at all values of pH .


Figure 5.25. Fluorescence ratio as a function of temperature at various values of pH . All measurements were taken using a 0.1 mM concentration of $1,4-\mathrm{DHPN}$ dissolved in 305 mOsm phosphate buffer. System parameters of 1 J at 8 Hz and an integration time of $100 \mu \mathrm{~s}$ were used. All temperatures were measured in degrees Celcuis. Note the larger error at higher values of both temperature and pH .


Figure 5.26. Fluoresence ratio versus pH as a function of flashlamp frequency. All measurements were taken using a 0.1 mM solution of 1,4 -DHPN dissolved in 305 mOsm phosphate buffer at pH 7.0 . An input energy of 1 J and an integration time of $100 \mu$ s were used.


Figure 5.27. Standard deviation of the fluorescence ratio versus pH as a function of flashlamp frequency. All measurements were taken using a 0.1 mM solution of 1,4-DHPN dissolved in 305 mOsm phosphate buffer at pH 7.0 . An input energy of 1 J and an integration time of $100 \mu \mathrm{~s}$ were used.


Figure 5.28. A plot of the standard deviation of the ratio versus the square root of flashlamp frequency. All measurements were taken using a 0.1 mM solution of 1,4-DHPN dissolved in 305 mOsm phosphate buffer at pH 7.0 . An input energy of 1 J and an integration time of $100 \mu \mathrm{~s}$ were used.


Figure 5.29. Fluorescence ratio versus pH as a function of input energy. All measurements were taken using a 0.1 mM solution of $1,4-$ DHPN dissolved in 305 mOsm phosphate buffer at pH 7.0 . A flashlamp frequency of 8 Hz and an integration time of $100 \mu \mathrm{~s}$ were used.


Figure 5.30. Standard deviation of the fluorescence ratio versus pH as a function of input energy. All measurements were taken using a 0.1 mM solution of $1,4-\mathrm{DHPN}$ dissolved in 305 mOsm phosphate buffer at pH 7.0 . A flashlamp frequency of 8 Hz and an integration time of $100 \mu \mathrm{~s}$ were used.


Figure 5.31. A plot of the standard deviation of the ratio versus the square root input energy. All measurements were taken using a 0.1 mM solution of $1,4-\mathrm{DHPN}$ dissolved in 305 mOsm phosphate buffer at pH 7.0 . An input energy of 1 J and an integration time of $100 \mu$ s were used.


Figure 5.32. A plot of the standard deviation of the ratio versus the square root of the number of samples. All measurements were taken using a 0.1 mM solution of 1,4-DHPN dissolved in 305 mOsm phosphate buffer at pH 7.0 . System parameters of 1 J at 8 Hz and an integration time of $100 \mu \mathrm{~s}$ were used.


Figure 5.33. A graph of the standard deviation of the ratio versus system integration time. All measurements were taken using a 0.1 mM solution of $1,4-\mathrm{DHPN}$ dissolved in 305 mOsm phosphate buffer at pH 7.0 . System parameters of 1 J at 8 Hz were used.


Figure 5.34. A graph of the standard deviation of the ratio versus fiber type. All measurements were taken using a 0.1 mM solution of $1,4-$ DHPN dissolved in 305 mOsm phosphate buffer at pH 7.0 . System parameters of 1 J at 8 Hz and an integration time of $100 \mu$ s were used. LGF-NL $=500$ micron core with no lens, LGF-.8L $=500$ micron core with spherical lens of $f=0.8 \mathrm{~mm}, \mathrm{LGF}-2 \mathrm{~L}=500$ micron core with spherical lens of $f=2.0 \mathrm{~mm}$, and $\mathrm{TG}-\mathrm{NL}=320$ micron core with no lens.


Figure 5.35. Fluorescence ratio response time curves for 1 mm inner diameter tubes of various materials when placed in a 305 mOsm phosphate buffer at $\mathrm{pH}=6.88$. The interior of the tubes was filled with a 0.1 mM solution of 1,4 -DHPN dissolved in normal saline.


Figure 5.36. Response time curves, at 488 nm , for 1 mm inner diameter tubes of various materials when placed in a 305 mOsm phosphate buffer at $\mathrm{pH}=6.88$. The interior of the tubes was filled with a 0.1 mM solution of $1,4-\mathrm{DHPN}$ dissolved in normal saline.


Figure 5.37. Response time curves, at 434 nm , for 1 mm inner diameter tubes of various materials when placed in a 305 mOsm phosphate buffer at $\mathrm{pH}=6.88$. The interior of the tubes was filled with a 0.1 mM solution of $1,4-$ DHPN dissolved in normal saline.


Figure 5.38. Fluorescence ratio response time curve for a 1 mm inner diameter polysulfone tube when placed in a 305 mOsm phosphate buffer at $\mathrm{pH}=6.88$. The interior of the tubes was filled with a 0.1 mM solution of 1,4 -DHPN dissolved in normal saline. Sensor time constant was 6.1 minutes.


Figure 5.39. Fluorescence ratio response time curves for 2.5 mm diameter semi-micro dialysis tubing when placed in 305 mOsm phosphate buffers at various values of pH . The interior of the tubes was filled with a 0.1 mM solution of 1,4-DHPN dissolved in normal saline.


Figure 5.40. The pH response time curves for 2.5 mm diameter semi-micro dialysis tubing when placed in 305 mOsm phosphate buffers at various values of pH . The interior of the tubes was filled with normal saline and the internal pH was monitored using a micro pH electrode.


Figure 5.41. Construction of a capillary-based Cuprophan sensor using free fluorophore dissolved in normal saline.


Figure 5.42. Fluorescence ratio response time curves of Cuprophan sealed tubes of various diameters. The interior of the tubes was filled with a 0.1 mM solution of 1,4-DHPN dissolved in normal saline.


Figure 5.43. Fluorescence ratio response time curves of a 1.1 mm inner diameter capillary based Cuprophan sensor, with a constant solution volume, using a fiber positioned at different distances from the membrane. The interior of the tubes was filled with a 0.1 mM solution of $1,4-$ DHPN, dissolved in normal saline.


Figure 5.44. Fluorescence ratio response time curves of a 1.1 mm inner diameter capillary based Cuprophan sensor, with different solution volumes, using a fiber positioned at a constant distance from the membrane. The interior of the tubes was filled with a 0.1 mM solution of $1,4-\mathrm{DHPN}$ dissolved in normal saline.


Figure 5.45. Fluorescence ratio response time curves of various size capillary based Cuprophan sensors using a fiber positioned 1 mm from the membrane. The interior of the tubes was filled with a 0.1 mM solution of $1,4-\mathrm{DHPN}$ dissolved in normal saline. A 305 mOsm test solution at pH 8.0 was used.


Figure 5.46. Gramacidin A dependent fluorescence ratio response time curves. Measurements were taken using a 1:1 dilution of $4: 1$ DPPC/DPPG LUV in 305 mOsm phosphate buffer. The LUV contained 303 mM sucrose and 10 mM 1,4-DHPN within their aqueous compartments. Various quantities of Gramicidin A (based upon mole $\%$ of total lipid) were added to the lipid phase. Note, the $2.0 \%$ Gramicidin curve should reach the same endpoint as the $0.2 \%$ curve. The observed shift is due to differences in measurement system calibration. A pH 7.5 buffer was used.


Figure 5.47. Detectibility of DPPC/DPPG LUV containing $10 \mathrm{mM} 1,4-$ DHPN and 303 mM sucrose. A 0.2 mole \% quantity of Gramicidin A was added to the lipid phase during preparation. Dilutions were made using 305 mOsm phosphate buffer at pH 7.5 .


Figure 5.48. Fluorescence ratio response time curves of free and 2.5 mm diameter microdialysis tubing bound DPPC/DPPG LUV containing 10 mM 1,4-DHPN and 303 mM sucrose. A 0.2 mole $\%$ quantity of Gramicidin A was added to the lipid phase during preparation. The test solution was 305 mOsm phosphate buffer at pH 7.5 .


Figure 5.49. Fluorescence ratio response time curves of free and 2.5 mm diameter microdialysis tubing bound DPPC/DPPG LUV containing $10 \mathrm{mM} 1,4-$ DHPN and 303 mM sucrose. A 2.0 mole \% quantity of Gramicidin A was added to the lipid phase during preparation. The test solution was 305 mOsm phosphate buffer at pH 7.5. The high steady state ratio is due to differences in system calibration.


Figure 5.50. Construction of an LUV based polysulfone sensor.


Figure 5.51. Fluorescence ratio response of a 1 mm inner diameter LUV based polysulfone sensor. The LUV contained a 2 mole \% quantity of Gramicidin A added to the lipid phase. The aqueous phase consisted of 10 mM 1,4-DHPN and 303 mM sucrose. The test solutions were 305 mOsm phosphate buffers at several different values of pH .


Figure 5.52. Fluorescence phase versus frequency of DPPC/DPPG LUV and free 1,4-DHPN in solution. The free dye was dissolved in 305 mOsm phosphate buffer at pH 7.0 , to a concentration of 0.002 mM . LUV contained 10 mM 1,4-DHPN dissolved in normal saline. These LUV were diluted $1 / 10,000$ in 305 mOsm phosphate buffer at pH 7.0 before use. No Gramicidin was added.


Figure 5.53. Fluorescence modulation ratio versus frequency of DPPC/DPPG LUV and free $1,4-$ DHPN in solution. The free dye was dissolved in 305 mOsm phosphate buffer, at pH 7.0 , to a concentration of 0.002 mM . LUV contained 10 $\mathrm{mM} 1,4-\mathrm{DHPN}$ dissolved in normal saline. These LUV were diluted $1 / 10,000$ in 305 mOsm phosphate buffer at pH 7.0 before use. No Gramicidin was added.


Figure 5.54. Fluorescence ratio versus pH of free fluorophore and LUV encapsulated fluorophore. Free dye was diluted to a concentration of 0.1 mM in 305 mOsm phosphate buffer at various values of pH . LUV contained a 2 mole $\%$ quantity of Gramicidin A in their lipid phase and 10 mM of 1,4-DHPN, dissolved in 303 mM sucrose, within their aqueous compartment. Liposomes were diluted 1:1 with phosphate buffer before measurement.


Figure 5.55. System response at 488 nm as a function of temperature for LUV based 1 mm inner diameter polysulfone sensors. LUV contained a 2.0 mole $\%$ quantity of Gramicidin A in their lipid phase and 10 mM 1,4-DHPN, dissolved in 303 mM sucrose, within their aqueous compartment. The test solution was 305 mOsm phosphate buffer at pH 7.0 .


Figure 5.56. System response at 434 nm as a function of temperature for LUV based 1 mm inner diameter polysulfone sensors. LUV contained a 2.0 mole \% quantity of Gramicidin A in their lipid phase and 10 mM 1,4-DHPN, dissolved in 303 mM sucrose, within their aqueous compartment. The test solution was 305 mOsm phosphate buffer at pH 7.0.

## LUV Entrapping 1, 4 DHPN



Figure 5.57. Construction of a LUV based Cuprophan capillary sensor.


Figure 5.58. Fluorescence ratio response curve for a LUV based 1.1 mm inner diameter Cuprophan capillary sensor. LUV contained a 2.0 mole $\%$ quantity of Gramicidin $\mathbf{A}$ in their lipid phase and $10 \mathrm{mM} 1,4-$ DHPN, dissolved in 303 mM sucrose, within their aqueous compartment. The test solution was a 305 mOsm phosphate buffer at pH 7.8 . The sensor time constant was 3.2 minutes.


Figure 5.59. The pH response curve for the sensor of Figure 5.58 calculated by means of a standard ratio curve.


Figure 5.60. Fluorescence ratio response curve for a LUV based 0.64 mm inner diameter Cuprophan capillary sensor. LUV contained a 2.0 mole $\%$ quantity of Gramicidin $\mathbf{A}$ in their lipid phase and $10 \mathrm{mM} 1,4-$ DHPN dissolved in 303 mM sucrose, within their aqueous compartment. The test solution was a 305 mOsm phosphate buffer at pH 8.1. The sensor time constant was 3.3 minutes.


Figure 5.61. Fluorescence ratio measurement of an LUV based 1 mm inner diameter polysulfone sensor in whole blood at $25^{\circ} \mathrm{C}$. LUV contained a 2.0 mole $\%$ quantity of Gramicidin A in their lipid phase and $10 \mathrm{mM} 1,4$-DHPN, dissolved in 303 mM sucrose, within their aqueous compartment. The time constant of this sensor was 19.8 minutes and it achieved a calculated pH of 7.06 at the end of 60 minutes. The measured pH of the whole blood was $\uparrow .24$.


Figure 5.62. Fluorescence ratio measurement of an LUV based 0.58 mm inner diameter Cuprophan capillary sensor in whole blood at $25^{\circ} \mathrm{C}$. LUV contained a 2.0 mole \% quantity of Gramicidin A in their lipid phase and $10 \mathrm{mM} 1,4-$ DHPN, dissolved in 303 mM sucrose, within their aqueous compartment. The time constant of this sensor was 10.0 minutes and it achieved a calculated pH of 7.24 at the end of 60 minutes. The measured pH of the whole blood was 7.24 .

## CHAPTER 6 <br> SUMMARY AND RECOMMENDATIONS FOR FURTHER STUDY

A fluorescence emission ratio based fiber optic pH measurement system was developed and evaluated for potential use during clinical hyperthermia of tumors. The behavior of the pH sensitive fluorophore 1,4-DHPN was studied in solution as a function of both temperature and pH . The pH dependence of the absorption spectrum, excitation spectrum, emission spectrum, and lifetime, as given by frequency domain modulation and phase measurements, were investigated for use as a basis in developing an optical pH measurement system. The temperature and pH dependent behavior of either the fluorescence emission spectrum or the fluorescence lifetime were determined to be well suited for the intended application. The temperature dependence of both lifetime and fluorescence emission appear to be near a minimum around pH 7 . When pH was determined from lifetime, using either modulation or phase measurement techniques, an average temperature induced pH error of +0.05 pH units $/{ }^{\circ} \mathrm{C}$ was introduced. Maximum sensitivity of this technique was obtained in the $6-7 \mathrm{pH}$ range, with a calculated measurement precision of $\pm 0.04 \mathrm{pH}$ units over the $6-8 \mathrm{pH}$ range. On the other hand, measurement of pH using emission wavelengths ratios shows a maximum sensitivity in the $7-8 \mathrm{pH}$ range. The average temperature induced pH error was calculated to be $\pm 0.02 \mathrm{pH}$ units $/{ }^{\circ} \mathrm{C}$. While both of these measurement techniques permit a simple single fiber sensor with adequate sensitivity and precision for use during induced hyperthermia, the simpler electronics and signal processing required in the emission ratio technique was attractive from an instrumentation viewpoint. As a result instrumentation was designed and built based on this measurement technique.

The resultant fluorescence emission ratio based pH measurement system consists of a flashlamp excitation source whose filtered narrowband energy is coupled by means of an optical fiber into an optical sensor module. This module contains a dichroic beamsplitter which reflects the excitation wavelengths into the sensor fiber. The pH dependent red shifted fluorescence signal travels back up the same fiber and is transmitted by the dichroic beamsplitter out of the sensor module and into a detector module. Here, the signal is split, and the two wavelengths of maximum pH sensitivity are selected by means of interference filters. These two optical signals are then transmitted to the photodetectors by means of optical fibers. The resultant electronic signals are processed separately in the electronic subsystem, digitized, and read into the memory of an Apple 2E microcomputer for subsequent digital processing and storage. All system and acquisition parameters are controlled by the Apple 2E microcomputer, using a program written in combined Applesoft BASIC and 6502 assembly language.

The weakest link in this measurement system appears to be in the conversion of input energy to optical excitation at the appropriate wavelength. Only about one part in $10,000,000$ of the energy input to the flashlamp appears as output in the desired wavelength range. Since the standard deviation of the measured emission ratio has been shown to decrease as the square root of the excitation energy, a more efficient signal source would improve both sensitivity and precision of the measurement system. More efficient methods of coupling energy both into and out of optical fibers should also be found. Presently only about $4 \%$ of the available flashlamp energy gets coupled into the sensor fiber. In addition, less than $1 \%$ of the total fluorescence energy, at any desired detection wavelength, finds its way to an appropriate photodetector.

The photodetector and electronics modules work well. Overall sensitivity of these system approach their theoretical limit. At 480 nm the photodiodes are
capable of detecting optical intensities of better than 1 nW . The photomultiplier tubes can detect intensities approaching 0.1 pW .

Using a fluorophore concentration of 10 mM , about 2 nW of fluorescence signal appear at each photodetector, when 1 joule of input excitation energy is supplied by the flashlamp. This is just slightly above the minimum detectable signal level using PIN photodiodes. For adequate SNR, photomultipliers must be used with the present optical configuration. Improvements in the excitation source and optical subsystem could make photodiode detection possible.

The present measurement system was found to be capable of measuring the pH of neutral solutions of 1 mM fluorophore, with a standard deviation of better than $\pm 0.1 \mathrm{pH}$ unit, using a flashlamp input energy of 1 joule. This standard deviation is based upon a 200 sample measurement.

The pH measurement standard deviation, as a function of temperature, was found to be approximately 0.01 pH units $/{ }^{\circ} \mathrm{C}$. When calibrated at $40^{\circ} \mathrm{C}$, the temperature induced measurement error is less than the the intrinsic system error over the temperature range usually employed in clinical hyperthermia.

Several materials were evaluated for use in fabricating pH sensitive optrodes. Materials having good hydrogen ion permeability coupled with good fluorophore retention were sought. A material with these properties was not found.

Cuprophan was selected for optrode fabrication based solely on its demonstrated permeability properties for hydrogen ions. Optrodes measuring less than 1 mm in diameter were fabricated by sealing the ends of small capillary tubes with a thin sheet of Cuprophan. These tubes were then filled with a 0.1 mM solution of fluorophore dissolved in normal saline. The time constants of such sensors, upon immersion in 305 mOsm phosphate buffers at various values of pH , were determined to be limited by hydrogen ion diffusion time and not membrane permeability, for reasonable membrane to optical fiber distances. The
measurement lifetime of these optrodes was limited to several minutes due to rapid leakage of the fluorophore across the membrane.

In order to extend the measurement lifetime of these optrodes, LUV made from a 4:1 mixture of DPPC/DPPG were used to encapsulate the pH sensitive fluorophore. The aqueous compartment of these LUV consisted of a 10 mM solution of 1,4-DHPN in 303 mM sucrose to achieve a near isosmotic internal environment. The kinetics of hydrogen ion transport in these LUV was found to be extremely slow. Gramicidin A was added to the lipid phase during preparation in order speed hydrogen ion transport across the membrane of these liposomes. At a Gramicidin A concentration equivalent to 2.0 mole percent of the total lipid present, the kinetics of hydrogen ion transport, across these liposomal membranes, was no longer found to be limiting.

Cuprophan capillary optrodes were again fabricated and filled with a 1:1 suspension of the previously discussed liposomes in normal saline. The response time of these membranes was again determined by immersion in 305 mOsm phosphate buffer at pH 8 . The expected steady state ratio for these optrodes was reached with an exponential time constant of approximately 3 minutes. Steady state ratio stability in excess of 1 hour was achieved, even at hyperthermic temperatures.

In order to test the physiological applicability of this optical pH measurement system, the previously discussed optrode was immersed into heparinized whole blood at $25^{\circ} \mathrm{C}$. The pH of this sample was measured with a standard pH meter and determined to be $7.24 \pm 0.02$. The calculated pH from the optical measurement system, at the end of 60 minutes, was found to be $7.24 \pm 0.06$. An exponential time constant of approximately 10 minutes was measured.

This research has demonstrated the feasibility of a fluorescence emission ratio based pH measurement system, with physiological applicability under hyperthermic conditions. Stability, accuracy, and precision of this measurement
system appear good. Response time in phosphate buffer is marginally acceptable for studies of pH dynamics during hyperthermia. However, further degradation by an approximate factor of 3 in whole blood cannot be tolerated. The mechanisms responsible for this anomaly bears further investigation. Preliminary indications are that the hydrogen ion flux across the liposomal membranes is again becoming a limiting factor. One explanation for this effect is that a flux limiting diffusion potential is being established across the liposomal membrane by charged impermeant molecules and ions that are present in the blood, but not in the buffers used in preliminary testing. Another explanation is that the Gramicidin channels are being blocked by divalent cations, most probably $\mathrm{Ca}^{++}$, which are present in the blood plasma. In either case, further modification of either the lipid membrane or the aqueous compartment may become necessary if reasonable in vivo time constants are to be achieved.

Finally, in order to achieve response times substantially under 1 minute, permeability limited optrodes need to be designed. This would presumably involve copolymerization of the fluorophore with a polymer on a thin film or covalent attachment of the fluorophore to a suitable support. While copolymerization of 1,4DHPN may be possible, lack of linkable side chains on this fluorophore prevent direct covalent linkages to a supporting structure. However, a newly synthesized SNARF series of pH sensitive emission fluorophores (Molecular Probes) may be the solution. The existing optical pH measurement system can be easily adapted to utilize this new fluorophore by a change in optical filters.

## APPENDIX A <br> PHYSICAL CONTROL LINE CONNECTIONS

| External PC |  |
| :---: | :---: |
| Interface Board | Control Line |
| Connections | Description |
| J1, Pin 1 | 10 M Feedback Resistor-CH0 |
| J1, Pin 2 | 1 M Feedback Resistor-CH0 |
| J1, Pin 3 | 100 K Feedback Resistor-CH0 |
| J1, Pin 4 | 10 K Feedback Resistor-CH0 |
| J1, Pin 5 | 10 M Feedback Resistor-CH1 |
| J1, Pin 6 | 1 M Feedback Resistor-CH1 |
| J1, Pin 7 | 100 K Feedback Resistor-CH1 |
| J1, Pin 8 | 10 K Feedback Resistor-CH1 |
| J1, Pin 11 | Shielded Cable GND |
| J2, Pin 11 | Shielded Cable GND |
| J3, Pin 1 | Integrator SW2-CH0 |
| J3, Pin 2 | Integrator SW1-CH1 |
| J3, Pin 5 | S/H Mode-CH0 \& CH1 |
| J3, Pin 7 | Spare 1 |
| J3, Pin 8 | Spare 2 |
| J4, Pin 8 | Flashlamp Trigger |
| J4, Pin 11 | Shielded Cable GND |
| P1, Pin 1 | A/D Input-CH0 |
| P1, Pin 2 | A/D Input-CH1 |
| P2, Pin 1 | Flashlamp Reference |
| P2, Pin 2 | PMT Reference-CH0 |
| P2, Pin 3 | PMT Reference-CH1 |
| P2, Pin 10 | Shielded Cable GND |

# APPENDIX B <br> SYSTEM INITIALIZATION PROGRAMS 

## PROGRAM HELLO

## 10 D\$=CHR\$ (4) <br> 20 PRINT D\$;"BRUN MMS"

## PROGRAM STARTUP

10 POKE 116,164
20 POKE 115,255
30 D $=$ CHR $\$(4)$
40 PRINT D\$;"BLOAD PH2.OBJ0" 50 PRINT D\$;"RUN PH2-REM" 60 END

## APPENDIX C APPLESOFT BASIC CONTROL PROGRAM

```
10 REM FILENAME: PH2-BASIC
20 REM
30 DIM ST(4,200): REM STATISTICAL ARRAY 40 DS\% \(=-23296\) : REM ASSEMBLY ROUTINE DUMMY SECTOR ADDRESS
\(50 \mathrm{OF} \%=\mathrm{DS} \%+32\)
\(60 \mathrm{UF} \%=\mathrm{DS} \%+42\)
70 INFO\% \(=-18432\) : REM ASSEMBLY ROUTINE DATA AREA
80 CD\$ = CHR\$ (4): REM CTRL-D
90 PRINT CD\$;"PR\#3": REM 80 COLUMN MODE
100 CALL - 23040: REM ASSEMBLY ROUTINE INITIALIZATION ADDRESS
110 REM ***********************************************************
120 HOME
130 PRINT
140 VTAB 8
150 PRINT SPC( 25);"ENTER 'B' TO BEGIN PROGRAM"
160 VTAB 10
170 PRINT SPC( 20);"ENTER 'Q' TO QUIT PROGRAM PEACEFULLY"
180 VTAB 15
190 HTAB 25
200 INPUT "PLEASE ENTER A CHOICE ";A\$
210 IF A\$ = "Q" THEN 730
220 IF A\$ < > "B" THEN 120
```



```
240 HOME
250 VTAB 5
260 PRINT SPC( 10);"DOES ANY OF THE ELECTRONIC SUBSYSTEM OFFSET
CIRCUITRY"
270 PRINT SPC( 25);"REQUIRE ADJUSTMENT? ";OFFSET\$
280 VTAB 10
290 HTAB 20
300 INPUT "PLEASE ENTER YOUR RESPONSE (Y/N) ";OFFSET\$
310 IF OFFSET\$ = "N" THEN 570
320 IF OFFSET\$ = "Y" THEN 360
330 VTAB 15
340 PRINT SPC( 25);"PLEASE ENTER 'Y' OR 'N"'
350 GOTO 280
360 POKE DS\% \(+26,1\)
370 GOSUB 750: REM ADJUSTMENT SELECTION
380 POKE DS\% + 26,0
390 HOME
400 VTAB 5
410 PRINT SPC( 30);"DO YOU WISH TO: "
420 PRINT
430 PRINT SPC( 10);"(1) PERFORM FURTHER ADJUSTMENTS TO THE OFFSET CIRCUITRY"
440 PRINT SPC( 10);"(2) CONFIGURE THE SYSTEM FOR ACTUAL MEASUREMENTS"
450 VTAB 10
460 HTAB 20
```

```
470 INPUT "ENTER YOUR CHOICE (1/2): ";CHOICE%
480 IF CHOICE% = 1 THEN OFFSET$ = "Y": GOTO 530
490 IF CHOICE% = 2 THEN OFFSET$ = "N": GOTO }57
500 VTAB }1
510 PRINT SPC( 25);"PLEASE ENTER '1' OR '2"'
520 GOTO 450
530 POKE DS% + 26,1
540 GOSUB 750
550 POKE DS% + 26,0
560 GOTO 390
570 POKE DS% + 26,0
5 8 0 \text { GOSUB 1330: REM FLASH LAMP PARAMETERS}
590 GOSUB 1770: REM INTEGRATION PARAMETERS
6 0 0 ~ H O M E ~
6 1 0 ~ V T A B ~ 5 ~
620 PRINT SPC( 20);"DO YOU WISH TO USE PHOTOMULTIPLIER TUBES? "
6 3 0 ~ V T A B ~ 1 0 ~
6 4 0 ~ H T A B ~ 2 1 ~
650 INPUT "PLEASE ENTER YOUR RESPONSE (Y/N) ";KEY$
660 IF KEY$ = "N" THEN G0 = 1:G1 = 1: GOSUB 9560:GOTO 680: REM SET CURRENT
GANN TO 1 AND FORMAT
670 GOSUB 8860: REM PHOTOMULTIPLIER GAIN PARAMETERS
680 CALL INT (256* PEEK (DS% + 0) + PEEK (DS% + 1) + 0.5): REM PHASE 1
690 GOSUB 2130: REM GAN CALIBRATION SELECTION
700 CALL INT (256* PEEK (DS% + 2) + PEEK (DS% + 3) + 0.5): REM PHASE }
7 1 0 \text { GOSUB 3050: REM MEASUREMENT SELECTION}
720 GOTO }12
7 3 0 \text { END}
740 REM *****************************************************************
750 HOME
7 6 0 \text { VTAB } 8
770 PRINT SPC( 15);"THE FOLLOWING SCREENS CONFIGURE THE SYSTEM FOR"
780 PRINT SPC( 15);"ADJUSTMENT ONLY. THE SYSTEM MUST BE RECONFIGURED"
790 PRINT SPC( 15);"FOR ACTUAL MEASUREMENTS FOLLOWING ADJUSTMENTS"
800 VTAB }1
810 HTAB }2
8 2 0 ~ I N P U T ~ " P R E S S ~ R E T U R N ~ T O ~ C O N T I N U E " ; B O G U S \$ ~
830 GOSUB 1330: REM FLASHLLAMP PARAMETERS
840 GOSUB 1770: REM INTEGRATION PARAMETERS
850 CALL INT (256* PEEK (DS% + 0) + PEEK (DS% + 1) + 0.5): REM PHASE }
860 GOSUB 2300: REM MANUAL GAIN CALIBRATION
870 CALL INT (256 * PEEK (DS% + 2) + PEEK (DS% + 3) + 0.5): REM PHASE }
80 GOSUB 4360: REM FORMATTING VARIABLES
800 HOME
900 VTAB }
910 PRINT SPC( 10);"DIGITIZED VALUES FOR CHANNELS 0 AND 1 WILL BE
CONTINUOUSLY"
920 PRINT SPC( 10);"DISPLAYED TO THE SCREEN. CHANNEL 0 VALUES ARE
DISPLAYED IN"
930 PRINT SPC( 10);"THE LEFT COLUMN, AND CHANNEL 1 VALUES ARE
DISPLAYED IN THE"
940 PRINT SPC( 10);"RIGHT COLUMN."
950 VTAB }1
```

960 PRINT SPC( 10 );"TO PAUSE THIS ROUTINE AT ANY TIME DURING EXECUTION, SIMPLY"
970 PRINT SPC( 10 );"PRESS THE SPACE BAR. TO RESUME, PRESS THE SPACE BAR AGAIN."
980 PRINT SPC( 10);"TO EXIT THE ROUTINE AT ANY TIME, HIT ANY OTHER KEY."
990 VTAB 15
1000 HTAB 26
1010 INPUT "PRESS RETURN TO BEGIN";BOGUS\$
1020 POKE - 16368,0: REM CLEAR KEYBOARD STROBE
1030 POKE DS\% + 25,1: REM SINGLE SWEEP
1040 POKE DS\% + 21,0: REM INJECTION CURRENT LOCATIONS
1050 POKE DS\% + 22,0
1060 POKE DS\% + 23,0
1070 POKE DS\% + 24,0
1080 HOME
$1090 \mathrm{VT} \%=1$
1100 IF PEEK ( -16384 ) > 127 THEN 1250: REM STROBE BIT SET
1110 CALL $\mathbb{N T}(256 *$ PEEK (DS\% + 6 ) + PEEK (DS\% + 7) + 0.5)
$1120 \mathrm{MO} \%=\operatorname{INT}(256 * \operatorname{PEER}(\mathbb{N F O} \%+80)+$ PEEK $(\mathbb{N F O} \%+81)+0.5)$
1130 TEMPS $=$ STR $\$($ M0\% * AO $/ 4095)$
1140 GOSUB 8710
1150 MOS = TEMP $\$$
$1160 \mathrm{M} 1 \%=\operatorname{INT}(256$ * PEEK $(\mathbb{I N F O} \%+82)+$ PEEK ( $\operatorname{INFO} \%+83)+0.5)$
1170 TEMPS $=$ STR $\$($ M1\% * A1 $/ 4095)$
1180 GOSUB 8710
$1190 \mathrm{M} 1 \$=$ TEMP $\$$
$1200 \mathrm{VT} \%=\mathrm{VT} \%+1$
1210 VTAB VT\%
1220 PRINT SPC( 20);MO\$; SPC( 15);M1\$
1230 IF VT\% > 22 THEN 1080
1240 GOTO 1100
1250 IF PEEK ( -16384 ) $=160$ THEN 1270
1260 IF PEEK ( -16384 ) < > 160 THEN 1300
1270 BOGUS $=$ PEEK ( -16368 ): REM CLEAR STROBE BIT
1280 IF PEEK ( -16384 ) < 127 THEN 1280: REM PAUSE MODE
1290 IF PEEK ( -16384 ) $=160$ THEN 1310
1300 BOGUS $=$ PEEK ( -16368 ): RETURN
1310 BOGUS $=$ PEEK ( -16368 ): GOTO 1100
1320 REM *******************************************************
1330 HOME
1340 VTAB 5
1350 PRINT SPC( 25 );"FLASHLAMP ENERGY SELECTION"
1360 VTAB 8
1370 PRINT SPC( 10 );"THE FLASHLAMP ENERGY MUST BE IN THE INTERVAL 0.1 -
2.25 JOULES"

1380 VTAB 10
1390 HTAB 26
1400 INPUT "ENTER THE FLASHLAMP ENERGY ";ENERGY
1410 IF ENERGY $<0.1$ THEN 1700
1420 IF ENERGY > 2.25 THEN 1700
1430 REF\% $=\operatorname{INT}((25.5 / 150) *$ SQR (ENERGY * 1E + 6) + 0.5): REM FLASHLAMP
REFERENCE
1440 POKE DS\% + 27,REF\%
1450 FBOUND\% $=\mathbb{I N T}(10 / E N E R G Y+0.5)$

```
1460 T1 = NNT (FBOUND% / 4)
1470 T2 = FBOUND% / 4
1480 IF T2 - Tl > 0.1 THEN FBOUND% = FBOUND% - 1: GOTO 1460
1490 FBOUND$ = STR$ (FBOUND%)
1500 HOME
1505 VTAB }
1510 PRINT SPC( 25);"FLASH LAMP FREQUENCY SELECTION"
1515 VTAB }
1520 PRINT SPC( 10);"WARNING: THE MAXIMUM FREQUENCY WHICH CAN BE
SAFELY USED FOR"
1525 PRINT SPC( 10);"THE ENERGY SELECTED IS ";FBOUND$;" HZ. CHOOSING A
HIGHER FREQUENCY"
1530 PRINT SPC( 10);"MAY LEAD TO A REDUCTION IN FLASHLAMP LIFETIME."
1535 VTAB }1
1540 PRINT SPC( 10);"THE FREQUENCY MUST BE IN THE INTERVAL 8HZ-100HZ
WITH A"
1545 PRINT SPC( 10);"RESOLUTION OF 4HZ (I.E. 8, 12, 16, 20, 24, ...)"
1550 VTAB }1
1555 HTAB 21
1560 INPUT "ENTER THE FLASH LAMP FREQUENCY ";LAMP%
1565 IF LAMP% < 8 THEN }173
1570 IF LAMP% > 100 THEN }173
1575 T1 = INT (LAMP% / 4)
1580 T2 = LAMP% / 4
1585 IF T2 - T1 > 0.1 THEN 1730
1590 POKE DS% + 8,LAMP%: REM FLASH LAMP FREQUENCY
1600 HOME
1610 VTAB 5
1620 PRINT SPC(25);"MEASUREMENT DELAY PARAMETER SELECTION"
1630 VTAB }
1640 PRINT SPC( 08);"THE FLASHLAMP WILL BE TRIGGERED SEVERAL TIMES
PRIOR TO BEGINNING DATA"
1650 PRINT SPC( 08);"ACQUISITION. THE RANGE OF ACCEPTABLE VALUES
EXTENDS FROM 1 TO 125."
1660 VTAB }1
1670 HTAB }2
1680 INPUT " ENTER DESIRED NUMBER OF DUMMY FLASHES ";DUMMY%
1681 IF DUMMY% < }1\mathrm{ THEN }173
1682 IF DUMMY% > }125\mathrm{ THEN 1730
1683 POKE DS% + 109,2 * DUMMY%: REM DUMMY FLASHES
1690 RETURN
1700 VTAB }2
1710 PRINT SPC( 25);"PLEASE ENTER ANOTHER VALUE"
1720 GOTO 1380
1730 VTAB }2
1740 PRINT SPC( 25);"PLEASE ENTER ANOTHER VALUE"
1750 GOTO 1550
1751 VTAB }2
1752 PRINT SPC( 25);"PLEASE ENTER ANOTHER VALUE"
1753 GOTO 1660
1760 REM ***************************************************************
1770 HOME
1780 VTAB }
1790 PRINT SPC( 20);"INTEGRATION PARAMETER SELECTION"
```

```
1800 VTAB }
1810 PRINT SPC(05);"AN INTEGRATION PERIOD EXTENDS FROM 15 USEC TO 1000
USEC IN"
1820 PRINT SPC(25);"5 USEC INCREMENTS"
1830 VTAB }1
1840 HTAB }2
1850 INPUT "ENTER THE INTEGRATION PERIOD ";PERIOD%
1860 IF PERIOD% < 15 THEN 2060
1870 IF PERIOD% > 1000 THEN }206
1880 T1 = INT (PERIOD% / 5)
1890 T2 = PERIOD% / 5
1900 IF T2 - T1 > 0.1 THEN 2060
1910 POKE DS% +9, INT (PERIOD% * 1.02273 / 5 + 0.5): REM INTEGRATION PERIOD
1920 VTAB }2
1930 PRINT SPC( 80)
1940 VTAB }1
1950 PRINT SPC( 05);"FOR EACH FLASH, THE INTEGRATOR WILL INTEGRATE FOR
A PERIOD OF TIME"
1960 PRINT SPC(05);"AS CHOSEN ABOVE. ADDITIONALLY, THE INTEGRATOR WILL
INGEGRATE OVER"
1970 PRINT SPC(05);"MULTIPLE FLASHES TO OBTAIN A SINGLE SAMPLE VALUE.
THE NUMBER"
1980 PRINT SPC(05);"OF FLASHES TO BE INTEGRATED MUST BE IN THE INTERVAL
1 TO 250."
1990 VTAB }1
2000 HTAB }1
2010 INPUT "ENTER THE NUMBER OF FLASHES TO BE INTEGRATED ";FLSH%
2020 IF FLSH% < 1 THEN 2090
2030 IF FLSH% > 250 THEN }209
2040 POKE DS% + 10,FLSH%: REM NUMBER OF INTEGRATIONS
2050 RETURN
2060 VTAB }2
2070 PRINT SPC(20);"PLEASE ENTER ANOTHER VALUE"
2080 GOTO 1830
2090 VTAB }2
2100 PRINT SPC( 25);"PLEASE ENTER ANOTHER VALUE"
2110 GOTO 1990
2120 REM ****************************************************************
2130 HOME
2140 VTAB 5
2150 PRINT SPC( 05);"THE SYSTEM ALLOWS FOR BOTH MANUAL AND AUTOMATIC
GAIN CALIBRATIONS"
2160 PRINT SPC(05);"BOTH CHANNELS CAN BE INDIVIDUALLY CALIBRATED AND
REQUIRE TWO"
2170 PRINT SPC( 05);"GAIN VALUES: A/D CHANNEL & PROGRAMMABLE GAIN I/V
AMPLIFIER"
2180 VTAB }1
2190 HTAB }2
2200 INPUT "DO YOU PREFER MANUAL CALIBRATION (Y/N) ";MAN$
2210 IF MAN$ = "N" THEN 2260
2220 IF MAN$ = "Y" THEN 2300
2230 VTAB 20
2240 PRINT SPC( 20);"PLEASE ENTER 'Y' OR 'N"'
2250 GOTO 2180
```

```
2260 POKE DS% + 14,0: REM MANUAL CALIBRATION FLAG
2270 VTAB 15
2280 PRINT SPC( 20);"SYSTEM IS PERFORMING CALIBRATION"
2290 RETURN
2300 POKE DS% + 14,1: REM MANUAL CALIBRATION FLAG
2310 HOME
2320 VTAB 2
2330 PRINT SPC( 25);"A/D MANUAL CALIBRATION MENU"
2340 VTAB }
2350 PRINT SPC( 15);"THE A/D IS CONFIGURED FOR UNIPOLAR OPERATION"
2360 VTAB 7
2370 PRINT SPC( 25);"VOLTAGE RANGE"; SPC( 10);"GANN"
2380 VTAB }
2390 PRINT SPC( 27);"0/ +10 V"; SPC( 14);"0"
2400 PRNNT SPC(27);"0/ + 5 V"; SPC(14);"1"
2410 PRINT SPC( 27);"0/ + 2 V'; SPC( 14);"2"
2420 PRINT SPC( 27);"0/ + 1 V"; SPC( 14);"3"
2430 PRINT SPC( 27);"0/ +500 MV"; SPC( 12);"4"
2440 PRINT SPC(27);"0/ +200 MV"; SPC( 12);"5"
2450 PRINT SPC( 27);"0/+100 MV"; SPC( 12);"6"
2460 PRINT SPC( 27);"0/ + 50 MV"; SPC( 12);"7"
2470 VTAB }1
2 4 8 0 ~ H T A B ~ 2 5 ~
2490 INPUT "ENTER CHANNEL O ADD GAIN (0-7) ";A0%
2500 IF A0% < O THEN }292
2510 IF A0% > 7 THEN }292
2520 POKE DS% + 16,A0%: REM CHANNEL O A/D RANGE
2530 VTAB 21
2540 HTAB 25
2550 NNPUT "ENTER CHANNEL 1 A/D GAIN (0-7) ";A1%
2560 IF A1% < O THEN }295
2570 IF A1% > 7 THEN 2950
2580 POKE DS% + 18,A1%: REM CHANNEL 1 AD RANGE
2590 HOME
2600 VTAB }
2610 PRINT SPC( 10);"PROGRAMMABLE GAIN AMPLIFIER MANUAL CALIBRATION
SELECTION MENU"
2620 VTAB 7
2630 PRINT SPC( 10);"THE PROGRAMMABLE GAIN I/V AMPLIFIER PRECEDES THE
INTEGRATOR CIRCUITRY."
2640 PRINT SPC( 10);"IDEALLY, THE INTEGRATED SIGNAL OF INTEREST SHOULD
BE 'CLOSE' TO'
2650 PRINT SPC( 10);"THE A/D RANGE, BUT KEPT JUST BELOW ITS MAXIMUM
VALUE"
2660 VTAB }1
2670.PRINT SPC( 25);"AMPLIFICATION LEVEL"; SPC( 10);"GAIN"
2680 VTAB }1
2690 PRINT SPC( 32);" 10,000"; SPC( 13);"0"
2700 PRINT SPC( 32);" 100,000"; SPC( 13);"1"
2710 PRINT SPC( 32);" 1,000,000"; SPC( 13);"2"
2720 PRINT SPC( 32);" 10,000,000"; SPC( 13);"3"
2730 VTAB }2
2 7 4 0 \text { HTAB } 2 5
2750 INPUT "ENTER CHANNEL O PROGRAMMABLE GAIN (0-3) ";V0%
```

```
2760 IF V0% < O THEN }298
2770 IF V0% > 3 THEN }298
2780 IF VO% = 0 THEN POKE DS% + 15,8: REM CHO PROGRAMMABLE GAIN
2790 IF VO% = 1 THEN POKE DS% + 15,4
2800 IF VO% = 2 THEN POKE DS% + 15,2
2810 IF V0% = 3 THEN POKE DS% + 15,1
2820 VTAB }2
2830 HTAB 25
2840 INPUT "ENTER CHANNEL 1 PROGRAMMABLE GAIN (0-3) ";V1%
2850 IF V1% < O THEN }301
2860 IF V1% > 3 THEN }301
2870 IF V 1% = 0 THEN POKE DS% + 17,128: REM CH1 PROGRAMMABLE GAIN
2880 IF V1% = 1 THEN POKE DS% + 17,64
2890 IF V1% = 2 THEN POKE DS% + 17,32
2900 IF V1% = 3 THEN POKE DS% + 17,16
2910 RETURN
2920 VTAB }2
2930 PRINT SPC( 25);"PLEASE ENTER ANOTHER VALUE"
2940 GOTO 2470
2950 VTAB }2
2960 PRINT SPC(25);"PLEASE ENTER ANOTHER VALUE"
2 9 7 0 ~ G O T O ~ 2 5 3 0
2980 VTAB }2
2990 PRINT SPC( 30);"PLEASE ENTER ANOTHER VALUE"
3000 GOTO 2730
3010 VTAB }2
3020 PRINT SPC( 30);"PLEASE ENTER ANOTHER VALUE"
3030 GOTO 2820
3040 REM *************************************************************
3050 HOME
3060 VTAB }1
3070 PRINT SPC( 10);"THE MULTIPLE SWEEP MEASUREMENT ROUTINE WILL
RETURN AVERAGED
3080 PRINT SPC(10);"VALUES FOR EACH SET OF SAMPLES TAKEN. THE NUMBER
OF SAMPLES
3090 PRINT SPC( 10);"COMPRISING EACH SET MUST BE IN THE INTERVAL 1 TO
100."
3100 VTAB }1
3110 HTAB }2
3120 INPUT "ENTER THE NUMBER OF SAMPLES DESIRED ";SAMPLES%
3130 IF SAMPLES% < 1 THEN }317
3140 IF SAMPLES% > }100\mathrm{ THEN }317
3150 POKE DS% + 12,SAMPLES%: REM NUMBER OF SAMPLES
3160 GOTO 3200
3170 VTAB }2
3180 PRINT SPC( 25);"PLEASE ENTER ANOTHER VALUE"
3190 GOTO 3100
3200 HOME
3210 VTAB 1
3220 PRINT SPC( 30);"MEASUREMENT ROUTINE SELECTION"
3230 VTAB }
3240 PRINT SPC( 10);"CHOICE"; SPC( 25);"DESCRIPTION"
3250 VTAB 5
```

3260 PRINT SPC( 05);"1-SINGLE SWEEP"; SPC( 08);"10 SAMPLES PER CHANNEL WILL BE MEASURED."
3270 PRINT SPC( 27);"THE CORRECTED DATA WILL BE RETURNED."
3280 VTAB 8
3290 PRINT SPC( 05);"2-MULTTIPLE SWEEP"; SPC( 06); STR\$ (SAMPLES\%);" SAMPLES
PER CHANNEL WILL BE MEASURED"
3300 PRINT SPC( 27);"THESE SAMPLES WILL BE AVERAGED TO FORM ONE"
3310 PRINT SPC( 27);"COMPUTATIONAL RESULT. 10 COMPUTATIONAL"
3320 PRINT SPC( 27);"RESULTS WILL BE RETURNED."
3330 VTAB 13
3340 PRINT SPC( 05);"3-DATA FILE"; SPC( 11);"A SPECIFIED NUMBER OF SAMPLES
PEK CHANNEL WILL BE "
3350 FRINT SPC( 27);"MEASURED. THE COMPENSATED DATA WILL BE SAVED TO A
1
3360 PRINT SPC( 27);"DATA FILE ON DISK."
3370 VTAB 17
3380 PRINT SPC( 05);"4-PRINTER"; SPC( 13);"PRINTER OPTION MENU"
3390 VTAB 19
3400 PRINT SPC( 05);"5-NEW CONFIGURATION"; SPC( 03);"RETURNS TO BEGINNING OF PROGRAM."
3410 VTAB 22
3420 HTAB 17
3430 INPUT "ENTER THE NUMBER ASSOCIATED WITH YOUR CHOICE (1-5) ";MODE\%
3440 IF MODE\% < 1 THEN 3520
3450 IF MODE\% > 5 THEN 3520
3460 POKE DS\% + 25,MODE\%: REM MEASUREMENT ROUTINE CHOICE
3470 IF MODE $\%=1$ THEN 3550
3480 IF MODE $\%=2$ THEN 3550
3490 IF MODE $\%=3$ THEN 4860
3500 IF MODE\% $=4$ THEN 5480
3510 IF MODE\% = 5 THEN RETURN : REM TO MAIN SEGMENT
3520 VTAB 23
3530 PRINT SPC( 25);"PLEASE ENTER ANOTHER VALUE"
3540 GOTO 3410
3550 HOME
3552 VTAB 4
3554 HTAB 15
3556 INPUT "DO YOU WISH TO SKIP INJECTION CURRENT MEASUREMENTS (Y/N)
";SKIP\$
3558 IF SKIP\$ = "Y" THEN POKE DS\% + 54,0: GOTO 3570
3560 VTAB 8
3562 PRINT SPC( 15);"DO YOU DESIRE FLASHLAMP TRIGGERING DURING THE
INJECTION"
3564 HTAB 16
3566 INPUT "CURRENT MEASUREMENT (1=NO, 2=YES) ";IJ\%
3568 POKE DS\% + 54,IJ\%: REM INJ.ACTIVE FLAG
3570 VTAB 12
3572 HTAB 21
3580 INPUT "PRESS RETURN WHEN READY TO EXECUTE";BOGUS\$
3582 IF SKIP\$ = "N" THEN 3590
3584 VTAB 16
3586 PRINT SPC( 15);"PREVIOUS INJECTION CURRENT VALUES WILL BE USED"
3588 GOTO 3620
3590 VTAB 16

3600 PRINT SPC(20);"MEASURING INJECTION CURRENT VALUES"
3610 CALL INT ( 256 * PEEK (DS\% + 4) + PEEK (DS\% + 5) + 0.5): REM PHASE 3
3615 IF IJ\% $=2$ THEN VTAB 18: HTAB 21: INPUT "PRESS RETURN WHEN READY TO CONTINUE ";BOGUS\$
3620 VTAB 20
3630 PRINT SPC( 25);"PERFORMING MEASUREMENT "
3640 CALL INT ( 256 * PEEK (DS\% + 6) + PEEK (DS\% + 7) + 0.5): REM PHASE 4
3650 HOME
3660 IF MODE\% $=1$ THEN PRINT SPC( 25);"SINGLE SWEEP RESULTS"
3670 IF MODE \% = 2 THEN PRINT SPC( 25);"MULTIPLE SWEEP RESULTS"
3680 GOSUB 4360
3690 VTAB 3
3700 PRINT SPC( 02);"CHO CURRENT GAIN: ";G0\$; SPC( 07);"CH1 CURRENT GAIN: ";G1\$
3710 PRINT SPC( 02);"CH0 PROGRAMMABLE GAIN: ";P0\$; SPC( 05);"CH1
PROGRAMMABLE GAIN: ";P1\$
3720 PRINT SPC(02);"CHO A/D RANGE: ";A0\$; SPC( 10 );"CH1 A/D RANGE: ";A1\$
3730 JO\% = 256 * PEEK (DS\% + 21) + PEEK (DS\% + 22): REM CHO INJECTION
CURRENT
$3740 \mathrm{~J} 1 \%=256$ * PEEK (DS\% + 23) + PEEK (DS\% + 24): REM CH1 INJECTION
CURRENT
3750 GOSUB 4620: REM FORMAT
3760 PRINT SPC( 02);"CHO INJECTION CURRENT: ";J0\$; SPC( 02);"CH1 INJECTION
CURRENT: ";J1\$
3770 VTAB 8
3780 PRINT SPC( 03);"CHANNEL 0"; SPC( 08);"CHANNEL 1"; SPC( 07);"RATIO (CH0/CH1)"; SPC(08);"\%^RATIO"
3790 VT\% $=10$
3800 BASE $\%=$ INFO $\%$
$3810 \mathrm{C} 0 \%=$ INT $(256$ * PEEK (BASE\% + 0) + PEEK (BASE\% + 1) + 0.5): REM CH0
DATA
$3820 \mathrm{C} 1 \%=$ INT ( 256 * PEEK (BASE\% + 2) + PEEK (BASE\% + 3) + 0.5): REM CH1
DATA
3821 ST( 0,0 ) $=$ C0\%
3822 ST( 1,0 ) = C1\%
3830 GOSUB 4700: REM FORMAT
3840 RTEMP $=(256 *$ PEEK (BASE\% + 4) + PEEK $(B A S E \% ~+5)+$ PEEK $(B A S E \% ~+~ 6) /$
$256+$ PEEK $($ BASE $\%+7) / 65.536 E+3)$
3845 RA = RTEMP * (A0 / A1)
3846 ST $(2,0)=$ RTEMP
3850 BASE $\%=$ BASE $\%+8$
3860 CHANGE $=0$
3870 GOSUB 4780: REM FORMAT
3880 VTAB VT\%
3890 IF PEEK $(\mathrm{UF} \%+0)=0$ THEN VTAB VT\%: HTAB 38: PRINT RA\$
3900 IF PEEK $(U F \%+0)=1$ THEN INVERSE: VTAB VT\%: HTAB 38: PRINT RA\$
3910 NORMAL
3920 IF PEEK $(\mathrm{OF} \%+0)=0$ THEN VTAB VT\%: HTAB 3: PRINT C0\$; SPC( 04);C1\$
3930 IF PEEK $(\mathrm{OF} \%+0)=1$ THEN INVERSE : VTAB VT\%: HTAB 3: PRINT C0\$; SPC(
04);C1\$

3940 NORMAL
3950 VT\% = VT\% + 1
3960 FOR LOOP = 1 TO 9
$3970 \mathrm{C} 0 \%=\operatorname{INT}(256$ * PEEK $($ BASE\% +0$)+$ PEEK $($ BASE\% + 1 $)+0.5)$

```
3980 C1% = INT (256 * PEEK (BASE% + 2) + PEEK (BASE% + 3) + 0.5)
3981 ST(0,LOOP) = C0%
3982 ST(1,LOOP)=C1%
3990 GOSUB 4700
4000 RP = RTEMP: REM PREVIOUS RATIO
4010 RTEMP = (256 * PEEK (BASE% + 4) + PEEK (BASE% + 5) + PEEK (BASE% + 6) /
256 + PEEK (BASE% + 7)/65.536E + 3)
4015 RA = RTEMP * (A0 / A1)
4016 ST(2,LOOP) = RTEMP
4020 BASE% = BASE% + 8
4030 CHANGE = ((RTEMP - RP) / RP) * 100
4040 GOSUB 4780
4050 VTAB VT%
4060 IF CHANGE < 0 THEN PRINT SPC( 57);CHANGE$
4070 IF CHANGE > = 0 THEN PRINT SPC( 58);CHANGE$
4080 IF PEEK (UF% + LOOP) = 0 THEN VTAB VT%: HTAB 38: PRINT RA$
4090 IF PEEK (UF% + LOOP) = }1\mathrm{ THEN INVERSE: VTAB VT%: HTAB 38: PRINT RA$
4 1 0 0 ~ N O R M A L ~
4110 IF PEEK (OF% + LOOP) = 0 THEN VTAB VT%: HTAB 3: PRINT CO$; SPC( 04);C1$
4120 IF PEEK (OF% + LOOP) = }1\mathrm{ THEN INVERSE: VTAB VT%: HTAB 3: PRINT C0$;
SPC( 04);C1$
4 1 3 0 ~ N O R M A L ~
4140 VT% = VT% + }
4 1 5 0 ~ N E X T ~ L O O P ~
4160 M0% = INT (256 * PEEK (BASE% + 0) + PEEK (BASE% + 1) + 0.5)
4170 TEMP$ = STR$ (M0% * A0 / 4095)
4180 GOSUB }871
4190 MO$ = TEMP$: REM CHANNEL 0 MEAN
4200 M1% = INT (256 * PEEK (BASE% + 2) + PEEK (BASE% + 3) + 0.5)
4210 TEMP$ = STR$ (M1% * Al / 4095)
4220 GOSUB }871
4230 M1$ = TEMP$: REM CHANNEL 1 MEAN
4240 MR = (256 * PEEK (BASE% + 4) + PEEK (BASE% + 5) + PEEK (BASE% + 6)/256 +
PEEK (BASE% + 7)/65.536E + 3)
4250 TEMP$ = STR$ (MR * A0/A1)
4260 GOSUB }871
4270 MR$ = TEMP$
4271 VTAB }2
4272 PRINT SPC( 02);M0$; SPC( 04);M1$; SPC( 03);MR$; SPC( 06);" AVG. VALUES"
4 2 7 3 ~ V 0 = 0 . 0
4274 V1 = 0.0
4275 V2 = 0.0
```



```
4277 V0 = V0 + (ST(0,LOOP) - M0%) * (ST(0,LOOP) - M0%)
4278 V1 = V1 + (ST(1,LOOP) - M1%) * (ST(1,LOOP) - M1%)
4279 V2 = V2 + (ST(2,LOOP) - MR) * (ST(2,LOOP) - MR)
4 2 8 0 ~ N E X T ~ L O O P ~
4 2 8 1 ~ V 0 ~ = ~ V 0 / 9 ~
4282 V1 = V1/9
4283 V2 = V2 / }
4284 TEMP$ = STR$ ((A0 / 4095)* SQR (V0))
4285 GOSUB }871
4286 S0$ = TEMP$
4287 TEMP$ = STR$ ((A1 / 4095) * SQR (V1))
```

```
4 2 8 8 \text { GOSUB } 8 7 1 0
4289 S1$ = TEMP$
4290 TEMP$ = STR$ ((A0 / A1)* SQR (V2))
4 2 9 1 ~ G O S U B ~ 8 7 1 0 ~
4292 S2$ = TEMP$
4293 PRINT SPC( 02);S0$; SPC( 04);S1$; SPC( 03);S2$; SPC( 07);"STD. DEV."
4300 VTAB }2
4310 HTAB }2
4320 INPUT "PRESS RETURN WHEN FINSHED VIEWING SCREEN ";BOGUS$
4330 GOTO 3200
4340 REM **************************************************************
4350 REM FILENAME: PH2-BASIC
4360 IF PEEK (DS% + 15) = 8 THEN PO$ = " 10,000": REM CH0 PROGRAMMABLE
GAIN
4370 IF PEEK (DS% + 15) = 4 THEN PO$ = " 100,000"
4380 IF PEEK (DS% + 15) = 2 THEN PO$ = " 1,000,000"
4390 IF PEEK (DS% + 15) = 1 THEN PO $ = " 10,000,000"
4400 IF PEEK (DS% + 16) = 0 THEN A0$ = " +10 V ":A0 = 10000.0: REM CH0 A/D
RANGE
4410 IF PEEK (DS% + 16) = 1 THEN AO$ = " +5 V ":A0=5000.0
4420 IF PEEK (DS% + 16) = 2 THEN AOS = " +2 V ":A0=2000.0
4430 IF PEEK (DS% + 16) = 3 THEN A0$ = " +1 V ":A0 = 1000.0
4440 IF PEEK (DS% + 16) = 4 THEN A0$ = " +500 mV":A0 = 500.0
4450 IF PEEK (DS% + 16) =5 THEN A0$ = " +200 mV":A0 = 200.0
4460 IF PEEK (DS% + 16) = 6 THEN A0$ = " + 100 mV":A0 = 100.0
4 4 7 0 ~ I F ~ P E E K ~ ( D S \% ~ + ~ 1 6 ) ~ = ~ 7 ~ T H E N ~ A 0 \$ ~ = ~ " ~ + 5 0 ~ m V ~ " : A 0 ~ = ~ 5 0 . 0
4480 IF PEEK (DS% + 17) = 128 THEN P1$ = " 10,000": REM CH1 PROGRAMMABLE
GAIN
4490 IF PEEK (DS% + 17) = 64 THEN P1 $ = " 100,000"
4500 IF PEEK (DS% + 17) = 32 THEN P1 $ = " 1,000,000"
4510 IF PEEK (DS% + 17) = 16 THEN P1$ = " 10,000,000"
4520 IF PEEK (DS% + 18) =0 THEN A1$ = " +10 V ":A1 = 10000.0: REM CH1 A/D
RANGE
4530 IF PEEK (DS% + 18) = 1 THEN A1$ = " +5 V ":A1 = 5000.0
4540 IF PEEK (DS% + 18) = 2 THEN A1$ = " +2 V ":A1 =2000.0
4550 IF PEEK (DS% + 18) = 3 THEN A1$ = " +1 V ":A1 = 1000.0
4560 IF PEEK (DS% + 18) =4 THEN A1$ = " +500 mV":A1 = 500.0
4570 IF PEEK (DS% + 18) = 5 THEN A1$ = " +200 mV":A1 = 200.0
4580 IF PEEK (DS% + 18) = 6 THEN A1$ = " +100 mV":A1 = 100.0
4590 IF PEEK (DS% + 18) = 7 THEN A1 $ = " +50 mV ":A1 = 50.0
4600 RETURN
4610 REM *****************************************************************
4620 TEMP$ = STR$ (J0% * A0 / 4095)
4630 GOSUB }871
4640 J0$ = TEMP$
4650 TEMP$ = STR$ (J1% * A1 / 4095)
4660 GOSUB }871
4670 J1$ = TEMP$
4680 RETURN
4690 REM *************************************************************
4700 TEMP$ = STR$ (C0% * A0 / 4095)
4710 GOSUB }871
4720 CO$ = TEMP$
4730 TEMP$ = STR$ (C1% * A1 / 4095)
```

4740 GOSUB 8710
4750 C1\$ = TEMP\$
4760 RETURN

4780 TEMP\$ = STR\$ (RA)
4790 GOSUB 8710
4800 RA $\$=$ TEMP\$
4810 TEMP\$ = STR\$ (CHANGE)
4820 GOSUB 8710
4830 CHANGE $=$ TEMP $\$$
4840 RETURN
4850 REM *************************************************************
4860 HOME
4870 VTAB 4
4880 PRINT SPC( 10);"A DATA FILE WILL BE CREATED CONTAINING ALL SYSTEM
PARAMETERS,"
4890 PRINT SPC( 10);"CORRECTED DATA, AND INJECTION CURRENT FOR A
SPECIFIED "
4900 PRINT SPC( 10 );"NUMBER OF SAMPLES IN EITHER A SINGLE POINT OR INTERVAL MODE."
4901 VTAB 8: HTAB 20
4902 INPUT "WHICH MODE DO YOU PREFER (S=SINGLE,I=INTERVAL) ";KEY\$
4904 IF KEY $=$ = "S" THEN 4910
4906 IF KEY $=$ = "I" THEN 4910
4908 GOTO 4860
4910 VTAB 10
4920 PRINT SPC( 10);"WARNING: IF YOU DESIGNATE THIS DATA FILE WITH AN
ALREADY"
4930 PRINT SPC( 10 );"EXISTING FILE NAME, THE PREVIOUS DATA FILE CONTENTS WILL BE"
4940 PRINT SPC( 10 );"LOST."
4950 VTAB 14
4960 PRINT SPC( 10);"IF YOU DESIRE TO SEE THE DISK CONTENTS BEFORE
NAMING THE"
4970 PRINT SPC( 10);"FILE, SIMPLY ENTER '?' FOR THE FILENAME."
4980 VTAB 17
4990 HTAB 20
5000 INPUT "ENTER THE DESIRED DATA FILE NAME ";FILES
5010 IF FILE $\$$ < > "?" THEN 5080
5020 HOME
5030 PRINT CD\$;"CATALOG,S6,D2": REM DISPLAY DISKETTE CONTENTS
5040 PRINT
5050 HTAB 20
5060 INPUT "PRESS RETURN WHEN READY TO CONTINUE ";BOGUS\$
5070 GOTO 4860
5080 IF KEY\$ = "I" THEN 6980
5081 PRINT
5082 VTAB 19
5083 HTAB 15
5084 INPUT "DO YOU WISH TO SKIP INJECTION CURRENT MEASUREMENTS (Y/N)
";SKIP\$
5085 IF SKIP $\$=$ "Y" THEN POKE DS\% + 54,0: GOTO 5091
5086 VTAB 21

```
5087 PRINT SPC( 15);"DO YOU DESIRE FLASHLAMP TRIGGERING DURING THE
INJECTION"
5088 HTAB }1
5089 INPUT "CURRENT MEASUREMENT (1=NO, 2=YES) ";IJ%
5090 POKE DS% + 54,IJ%: REM INJ.ACTIVE FLAG
5091 IF SKIPS = "N" THEN 5100
5092 VTAB }2
5093 PRINT SPC( 15);"PREVIOUS INJECTION CURRENT VALUES WILL BE USED"
5094 GOTO 5110
5100 PRINT
5101 PRINT SPC( 20);"MEASURING INJECTION CURRENT VALUES"
5 1 0 2 ~ P R I N T ~
5105 CALL INT (256* PEEK (DS% + 4) + PEEK (DS% + 5) + 0.5): REM PHASE 3
5108 IF IJ% = 2 THEN HTAB 20: INPUT "PRESS RETURN WHEN READY TO
CONTINUE ";BOGUS$
5110 PRINT
5120 PRINT SPC( 25);"PERFORMING MEASUREMENT"
5130 CALL INT (256* PEEK (DS% + 6) + PEEK (DS% + 7) + 0.5): REM PHASE }
5140 PRINT
5150 PRINT SPC(20);"DATA FILE BEING WRITTEN TO DISKETTE"
5152 TYPE$ = "SINGLE POINT"
5154 MEAS% = 200
5156 TIME% =0
5158 UN$ = "SECONDS"
5160 PRINT CD$;"OPEN ";FILE$;",S6,D2"
5170 PRINT CD$;"DELETE ";FILE$;",S6,D2"
5180 PRINT CD$;"OPEN ";FILE$;",S6,D2"
5190 PRINT CD$;"WRITE ";FILE$
5192 PRINT TYPE$
5194 PRINT MEAS%
5196 PRINT TIME%
5198 PRINT UN$
5200 PRINT STR$ (ENERGY): REM ENERGY PER FLASH
5210 PRINT STR$ (LAMP%): REM FLASHLAMP FREQUENCY
5220 PRINT STR$ (PERIOD%): REM INTEGRATION PERIOD
5230 PRINT STR$ (FLSH%): REM MULTPLE INTEGRATION COUNT
5240 PRINT GO$: REM CHANNEL O CURRENT GAIN
5250 PRINT G1$: REM CHANNEL 1 CURRENT GAIN
5260 I1% = PEEK (DS% + 15): REM CHANNEL 0 PROGRAMMABLE GAIN
5270 I2% = PEEK (DS% + 16): REM CHANNEL 0 A/D RANGE
5280 I3% = PEEK (DS% + 17): REM CHANNEL 1 PROGRAMMABLE GAIN
5290 I4% = PEEK (DS% + 18): REM CHANNEL 1 A/D RANGE
5291 GOSUB }662
5 2 9 2 ~ P R I N T ~ P 0 ~
5293 PRINT A0
5294 PRINT P1
5295 PRINT Al
5300 IF KEY$ = "I" THEN 7430
5302 J0% =256* PEEK (DS% + 21) + PEEK (DS% + 22)
5304 J1% = 256 * PEEK (DS% + 23) + PEEK (DS% + 24)
5306 J0 = J0% * A0 / 4095
5308 J1 = J1% * A1 / 4095
5340 REM ********************************************************************
5350 D1% = - 18176
```

```
5360 D2% = - 17920
5370 D3% = - 17664
5380 D4% = -17408
5390 FOR SAMPLE = 1 TO 200
5400 C0% = 256* PEEK (D1% + SAMPLE - 1) + PEEK (D2% + SAMPLE - 1) - J0% 
5402 C1% = 256* PEEK (D3% + SAMPLE - 1) + PEEK (D4% + SAMPLE - 1) - J1%
5404 C0= C0% * A0 / 4095
5406 C1 = C1% * A1 / 4095
5408 IF C1% = 0 THEN RA = 1.0E15
5410 IF C1% < > OTHENRA = (C0% / C1%)* (AO/A1)
5412 PRINT J0","J1","C0","C1","RA","0","0","0
5 4 4 0 ~ N E X T ~ S A M P L E ~
5450 PRINT CD$;"CLOSE ";FILE$
5460 GOTO 3200
```



```
5480 HOME
5490 VTAB 1
5500 PRINT SPC( 30);"PRINTER OPTION SELECTION"
5510 VTAB 3
5520 PRINT SPC( 07);"CHOICE"; SPC( 23);"DESCRIPTION"
5530 VTAB }
5540 PRINT SPC(04);"1-DATA FILE"; SPC(05);"THE RESUUTS OF AN EXISTING DATA
FILE CREATED FROM"
5550 PRINT SPC( 20);"THE PREVIOUS MENU WILL BE DUMPED TO THE PRINTER"
5560 VTAB 08
5570 PRINT SPC(04);"2-INTERVAL"; SPC(06);"THE SYSTEM WILL TAKE A USER
SPECIFIED NUMBER OF"
5580 PRINT SPC( 20);"MEASUREMENTS. ADDITIONALLY,THE USER SPECIFIES
THE"
5590 PRNNT SPC( 20);"INTERVAL OF TIME BETWEEN MEASUREMENTS IN
SECONDS."
5600 PRINT SPC( 20);"THE COMPUTATIONAL RESULTS WILL BE DUMPED TO THE"
5610 PRINT SPC( 20);"PRINTER ALONG WITH SYSTEM PARAMTERTERS."
5620 VTAB }1
5630 PRINT SPC( 04);"3-STATISTICS"; SPC( 04);"AN EXISTING DATA FILE WILL BE
POST PROCESSED FOR"
5640 PRINT SPC( 20);"STATISTICAL PARAMETERS."
5650 VTAB }1
5660 PRINT SPC( 04);"4-EXIT"; SPC( 10);"RETURN TO PREVIOUS MENU"
5670 VTAB }2
5680 HTAB }1
560 INPUT "ENTER THE NUMBER ASSOCIATED WITH YOUR CHOICE ";PR%
5700 IF PR% < }1\mathrm{ THEN 5760
5710 IF PR% > 4 THEN 5760
5720 IF PR% = 1 THEN 5800
5730 IF PR% = 2 THEN KEY$ = "BOGUS": GOTO }698
5740 IF PR% = 3 THEN }777
5750 IF PR% = 4 THEN 3200
5760 VTAB }2
5 7 7 0 \text { PRINT SPC( 25);"PLEASE ENTER ANOTHER VALUE"}
5780 GOTO 5480
```



```
5800 HOME
5 8 1 0 ~ V T A B ~ 7 ~
```

```
5820 PRINT SPC( 20);"BE SURE THE PRINTER IS 'ON LINE"'
5830 VTAB }1
5840 PRINT SPC( 10);"IF YOU DESIRE TO SEE THE DISK CONTENTS BEFORE
NAMING"
5850 PRINT SPC( 10);"THE FLLE, SIMPLY ENTER '?' FOR THE FLLENAME."
5860 VTAB 15
5870 HTAB 20
5880 INPUT "ENTER THE DESIRED DATA FILE NAME ";FLLE$
5890 IF FLLES < > "?" THEN 5960
5900 HOME
5910 PRINT CD$;"CATALOG,S6,D2"
5920 PRINT
5930 HTAB 20
5940 INPUT "PRESS RETURN TO CONTINUE ";BOGUS$
5950 GOTO 5800
5960 VTAB 20
5970 PRINT SPC(22);"PRINTING THE DATA FILE ";FLES
5980 PRINT CD$;"PR# 1": REM ACTIVATE PRINTER
5990 PRINT CDS;"OPEN ";FLLES;",S6,D2"
6000 PRINT CDS;"READ ";FILES
6010 GOSUB 6180: REM SYSTEM PARAMETERS
6020 PRNT "SAMPLE#"; SPC(06);"CH0 INJECTION"; SPC( 10);"CH1 INJECTION"; SPC(
10);"CHANNEL 0 DATA"; SPC( 10);"CHANNEL 1 DATA"; SPC( 13);"RATIO"
6 0 3 0 \text { PRINT}
6040 FOR LOOP = }1\mathrm{ TO MEAS%
6042 INPUT J0,\1,C0,C1,RA,S0,S1,S2
6044 TEMP$ = STR$ (J0)
6046 GOSUB }871
6048 J0$ = TEMP$
6050 TEMP$ = STR$ (J1)
6052 GOSUB }871
6054 J1$ = TEMP$
6056 TEMP$ = STR$ (C0)
6058 GOSUB }871
6060 C0$ = TEMP$
6062 TEMP$ = STR$ (C1)
6064 GOSUB }871
6066 C1$ = TEMP$
6068 TEMP$ = STR$ (RA)
6070 GOSUB 8710
6072 RA$ = TEMP$
6090 GOSUB 8780: REM FORMAT
6100 PRINT SPC( 04);LOOP$; SPC( 09);0$; SPC( 09);11$; SPC( 09);C0$; SPC( 09);C1$;
SPC(09);RAS
6 1 1 0 ~ N E X T L O O P
6 1 2 0 ~ G O S U B ~ 6 8 8 0 ~
6130 PRINT CHR$ (140): REM FORM FEED
6140 PRINT CD$;"PR# 3": REM REACTIVATE 80 COLUMN
6150 PRINT CD$;"CLOSE ";FLE$
6160 GOTO 5480
```



```
6180 PRINT "FILENAME: ";FLLE$
6 1 9 0 \text { PRINT}
6 1 9 1 ~ I N P U T ~ T Y P E \$ ~
```

```
6192 PRINT "FILE TYPE: ";TYPE$
6 1 9 3 \text { INPUT MEAS\%}
6194 PRINT "NUMBER OF MEASUREMENTS: ";MEAS%
6 1 9 5 ~ I N P U T ~ T I M E \% ~
6 1 9 6 ~ P R I N T ~ " T I M E ~ B E T W E E N ~ M E A S U R E M E N T S : ~ " ; T I M E \% ~
6 1 9 7 \text { INPUT UN\$}
6 1 9 8 \text { PRINT "MEASUREMENT UNIT: ";UN\$}
6 2 0 0 ~ I N P U T ~ I N \$ ~
6210 PRINT "ENERGY PER FLASH: ";IN$;" JOULES"
6220 INPUT IN$
6230 PRINT "FLASH LAMP FREQUENCY: ";NN$;" HZ"
6 2 4 0 ~ I N P U T ~ I N \$ ~
6250 PRINT "INTEGRATION PERIOD: ";IN$;" USEC"
6 2 6 0 ~ I N P U T ~ I N \$ ~
6270 PRINT "MULTIPLE INTEGRATION COUNT: ";IN$
6280 PRINT
6 2 9 0 \text { GOSUB } 6 4 6 0
6300 INPUT G0$
6 3 1 0 ~ I N P U T ~ G 1 \$ ~
6320 PRINT "CHANNEL 0 CURRENT GAIN: ";G0$; SPC( 04);"CHANNEL 1 CURRENT
GAIN: ";G1$
6330 GOSUB }656
6 3 4 0 \text { GOSUB } 6 6 2 0
6350 PRINT "CHANNEL 0 PROGRAMMABLE GAIN: ";PO$; SPC( 02);"CHANNEL 1
PROGRAMMABLE GAIN: ";P1$
6360 PRINT "CHANNEL O A/D RANGE: ";A0$; SPC( 12);"CHANNEL 1 ADD RANGE:
";A1$
6 3 7 0 \text { PRINT}
6 4 4 0 ~ R E T U R N
```



```
6460 T1$ = PO$
6470 T2 = A0
6480 T2$ = A0$
6490 T3$ = P1$
6500 T4 = A1
6510 T4$ = A1$
6520 T5$ = G0$
6530 T6$ = G1$
6 5 4 0 ~ R E T U R N
6550 REM *************************************************************
6 5 6 0 ~ I N P U T ~ P O ~
6 5 6 1 ~ I N P U T ~ A O ~
6 5 6 2 ~ I N P U T ~ P 1 ~
6 5 6 3 \text { INPUT A1}
6 5 6 4 ~ I F ~ P 0 ~ = ~ 1 0 0 0 0 ~ T H E N ~ I 1 \% ~ = ~ 8 ~
6 5 6 5 \text { IF P0 = 100000 THEN I \% \% = 4}
6 5 6 6 ~ I F ~ P 0 ~ = ~ 1 0 0 0 0 0 0 ~ T H E N ~ I 1 \% ~ = 2 ~
6 5 6 7 \text { IF PO = 10000000 THEN I } 1 \% = 1
6 5 6 8 \text { IF A0 = 10000.0 THEN I I\% =0}
6 5 6 9 \text { IF A0 = 5000.0 THEN I2\% =1}
6 5 7 0 ~ I F ~ A 0 ~ = ~ 2 0 0 0 . 0 ~ T H E N ~ I 2 \% ~ = 2 ~
6571 IF A0 = 1000.0 THEN I2% = 3
6572 IF A0 = 500.0 THEN I 2% = 4
6573 IF A0 = 200.0 THEN I2% = 5
```

```
6574 IF A0 = 100.0 THEN 12% = 6
6575 IF A0 = 50.0 THEN 12% =7
6576 IF P1 = 10000 THEN I3% = 128
6577 IF P1 = 100000 THEN 13% = 64
6578 IF P1 = 1000000 THEN I3% = 32
6579 IF Pl = 10000000 THEN I3% = 16
6580 IF A1 = 10000.0 THEN I4% = 0
6581 IF A1 = 5000.0 THEN I4% = 1
6582 IF A1 = 2000.0 THEN 14% = 2
6583 IF A1 = 1000.0 THEN 14% = 3
6584 IF Al = 500.0 THEN 14% = 4
6585 IF A1 = 200.0 THEN 14% = 5
6 5 8 6 \text { IF A1 = 100.0 THEN 14\% = 6}
6587 IF Al = 50.0 THEN I4% = 7
6 6 0 0 \text { RETURN}
6610 REM *************************************************************
6620 IF I1% = 8 THEN PO$ = " 10,000":PO = 10000
6630 IF I1% = 4 THEN PO$ = "' 100,000":PO = 100000
6640 IF I1% =2 THEN P0$ = " 1,000,000":P0 = 1000000
6 6 5 0 \text { IF I1\% = 1 THEN PO\$ = " 10,000,000":PO= = 10000000}
6660 IF 12% = 0 THEN A0$ = "+10 V ":A0 = 10000.0
6670 IF I2% = 1 THEN AO$ = "+5 V ":A0 = 5000.0
6 6 8 0 \text { IF I2\% =2 THEN AO\$ = "+2 V ":A0 = 2000.0}
6600 IF 12% = 3 THEN AO$ = "+1 V ":A0 = 1000.0
6700 IF I2% = 4 THEN A0$ = "+500 mV":A0 = 500.0
6 7 1 0 \text { IF I2\% = 5 THEN A0 \$ = "+200 mV":A0 = 200.0}
6720 IF I2% = 6 THEN A0$ = "+100 mV":A0 = 100.0
6730 IF I2% = 7 THEN A0$ = "+50 mV ":A0 = 50.0
6740 IF I3% = 128 THEN P1$ = ' 10,000":P1 = 10000
6750 IF I3% = 64 THEN P1$ = " 100,000":P1 = 100000
6 7 6 0 \text { IF 13\% = 32 THEN P1\$ =" 1,000,000":P1 = 1000000}
6 7 7 0 \text { IF I3\% = 16 THEN P1\$ = " 10,000,000":P1 = 10000000}
6780 IF I4% = 0 THEN A1$ = "+10 V ":A1 = 10000.0
6790 IF I4% = 1 THEN A1$ = "+5 V ":A1 = 5000.0
6800 IF I4% = 2 THEN A1 $ = " +2 V ":A1 = 2000.0
6810 IF I4% = 3 THEN A1 $ = "+1 V ":A1 = 1000.0
6820 IF I4% = 4 THEN A1$ = "+500 mV":A1 = 500.0
6830 IF I4% = 5 THEN A1 $ = "+200 mV":Al = 200.0
6840 IF I4% = 6 THEN A1$ = "+100 mV":Al = 100.0
6850 IF I4% = 7 THEN A 1 $ = "+50 mV ":A1 = 50.0
6860 RETURN
6870 REM *************************************************************
680 PO$ = T1$
6890 A0 = T2
6900 A0$ = T2$
6910 P1$ = T3$
6920 A1 = T4
6930 A1$ = T4$
6940 G0$ = T5$
6950 G1$ = T6$
690 RETURN
```



```
6 9 8 0 ~ H O M E
7000 PRINT SPC( 22);"INTERVAL MEASUREMENT ROUTINE"
```

7002 IF KEY\$ < > "I" THEN 7010
7004 VTAB 3
7006 HTAB 15
7008 PRINT SPC( 04);"RESULTS WILL BE SAVED TO A DATA FILE"
7010 VTAB 5
7020 HTAB 15
7030 INPUT "ENTER THE DESIRED MEASUREMENT UNIT (H,M,S) ";UN\$
7040 VTAB 7
7050 HTAB 15
7060 INPUT "ENTER THE DESIRED TIME BETWEEN MEASUREMENTS ";TIME\%
7070 VTAB 9
7080 HTAB 15
7090 INPUT "ENTER THE TOTAL NUMBER OF MEASUREMENTS YOU DESIRE
";MEAS\%
7092 VTAB 11
7094 HTAB 15
7096 INPUT "DO YOU WISH TO SKIP INJECTION CURRENT MEASUREMENTS (Y/N)
";SKIP\$
7097 VTAB 14
7098 IF SKIP\$ = "Y" THEN PRINT SPC( 14);"NO INJECTION CURRENT MEASUREMENTS ARE PERFORMED"
7100 VTAB 17
7110 IF KEY\$ < > "I" THEN PRINT SPC( 20);"BE SURE THE PRINTER IS 'ON LINE"'
7140 VTAB 19
7150 HTAB 26
7160 INPUT "PRESS RETURN TO BEGIN ";BOGUS\$
7170 VTAB 22
7180 PRINT SPC( 20);"EXECUTING INTERVAL MEASUREMENT"
7190 IF UNS = "H" THEN UN\$ = "HOURS": POKE DS\% + 100,4: POKE DS\% + 101,12:
REM SELECT HOUR UNITS
7200 IF UN $\$=$ "M" THEN UN\$ = "MINUTES": POKE DS\% + 100,4: POKE DS\% + 101,5:
REM SELECT MINUTE UNITS
7210 IF UN\$ = "S" THEN UN\$ = "SECONDS": POKE DS\% + 100,12: POKE DS\% + 101,4:
REM SELECT SECOND UNITS
7220 POKE DS $\%+25,2$ : REM MULTIPLE SWEEP
7222 IF KEY\$ = "I" THEN 7422
7230 PRINT CD\$;"PR\# 1": REM REACTIVATE PRINTER
7240 PRINT "TOTAL NUMBER OF MEASUREMENTS: "; STR\$ (MEAS\%)
7250 PRINT "TIME BETWEEN MEASUREMENTS: "; STR\$ (TIME\%);" ";UN\$
7260 PRINT
7270 PRINT "ENERGY PER FLASH: "; STR\$ (ENERGY);" JOULES"
7280 PRINT "FLASH LAMP FREQUENCY: "; STR\$ (LAMP\%);" HZ"
7290 PRINT "INTEGRATION PERIOD: "; STR\$ (PERIOD\%);" USEC"
7300 PRINT "MULTIPLE INTEGRATION COUNT: "; STR\$ (FLSH\%)
7310 PRINT "SAMPLES AVERAGED FOR DATA VALUES: "STR\$ (SAMPLES\%)
7320 PRINT
7330 GOSUB 4360
7340 PRINT "CHANNEL 0 CURRENT GAIN: ";GO\$
7350 PRINT "CHANNEL 0 PROGRAMMABLE GAIN: ";PO\$
7360 PRINT "CHANNEL 0 A/D RANGE: ";A0\$
7370 PRINT
7380 PRINT "CHANNEL 1 CURRENT GAIN: ";G1\$
7390 PRINT "CHANNEL 1 PROGRAMMABLE GAIN: ";P1\$
7400 PRINT "CHANNEL 1 A/D RANGE: ";A1\$

```
7410 PRINT
7420 PRINT
7422 TYPES = "INTERVAL"
7424 IF KEY$ = "I" THEN 5160
7430 CHANGE =0
7440 POKE DS% + 102,0: REM CLEAR TIME.OUT FLAG
7450 CALL INT (256 * PEEK (DS% + 103) + PEEK (DS% + 104) + 0.5): REM START
CLOCK
7460 IF PEEK (DS% + 102) = 0 THEN 7460: REM WAIT FOR START OF INTERVAL
7470 FOR LOOP = 1 TO MEAS%
7480 POKE DS% + 102,0: REM CLEAR TIME.OUT COUNTER
7485 POKE DS% + 54,0: REM RESET INJ.ACTIVE FLAG FOR PHASE }
7490 IF SKIP$ = "N" THEN POKE DS% + 54,1: CALL INT (256 * PEEK (DS% + 4) +
PEEK (DS% + 5) +0.5): REM PHASE 3
7500 CALL NNT (256* PEEK (DS% + 6) + PEEK (DS% + 7) + 0.5); REM PHASE }
7510 CALL INT (256* PEEK (DS% + 107) + PEEK (DS% + 108) + 0.5): REM REENABLE
PROCESSOR INTERRUPTS
7520 J0% = 256 * PEEK (DS% + 21) + PEEK (DS% + 22): REM CHO INJECTION
CURRENT
7530 J1% = 256 * PEEK (DS% + 23) + PEEK (DS% + 24): REM CH1 INJECTION
CURRENT
7540 GOSUB 4620: REM FORMAT
7542 J0 = J0% * A0 / 4095
7544 J1 = J1% * Al / 4095
7550 IF KEY$ < > "I" THEN PRINT
7551 BASE% = INFO%
7552 FOR COUNT = 0 TO 9
7553 ST(0,COUNT) = INT (256* PEEK (BASE% + 0) + PEEK (BASE% + 1) + 0.5)
7554 ST(1,COUNT) = INT (256 * PEEK (BASE% + 2) + PEEK (BASE% + 3) + 0.5)
7555 ST(2,COUNT) = (256* PEEK (BASE% + 4) + PEEK (BASE% + 5) + PEEK (BASE%
+6)/256 + PEEK (BASE% + 7)/65.536E03)
7556 BASE% = BASE% + 8
7 5 5 7 \text { NEXT COUNT}
7560 C0% = INT (256 * PEEK (INFO% + 80) + PEEK (INFO% + 81) + 0.5)
7570 C1% = INT (256* PEEK (INFO% + 82) + PEEK (INFO% + 83) + 0.5)
7572 C0 = C0% * A0 / 4095
7574 C1 = C1% * Al / 4095
7580 GOSUB 4700
7590 RP = RTEMP
7600 RTEMP = (256 * PEEK (INFO% + 84) + PEEK (INFO% + 85) + PEEK (INFO% + 86) /
256 + PEEK (INFO% + 87) / 65.536E + 3)
7605 RA = RTEMP * (A0 / A1)
7610 IF LOOP > 1 THEN CHANGE = ((RTEMP - RP) / RP) * 100
7 6 1 1 \text { GOSUB } 4 7 8 0
7 6 1 2 ~ V 0 = 0 . 0
7613 V1 = 0.0
7 6 1 4 ~ V 2 ~ = ~ 0 . 0 ~
7 6 1 5 ~ F O R ~ C O U N T ~ = ~ 0 ~ T O ~ 9 ~
7 6 1 6 ~ V 0 ~ = ~ V 0 ~ + ~ ( S T ( 0 , C O U N T ) ~ - ~ C 0 \% ) ~ * ~ ( S T ( 0 , C O U N T ) ~ - ~ C 0 \% ) ~
7617 V1 = V1 + (ST(1,COUNT) - C1%) * (ST(1,COUNT) - C1%)
7618 V2 = V2 + (ST(2,COUNT) - RTEMP) * (ST(2,COUNT) - RTEMP)
7 6 1 9 \text { NEXT COUNT}
7 6 2 0 ~ V O ~ = ~ V 0 / 9 ~
7621 V1 = V1/9
```

```
7622 V2 = V2 / 9
7623 TEMP$ = STR$ ((A0 / 4095) * SQR (V0)):S0 = (A0 / 4095) * SQR (V0)
7624 GOSUB }871
7625 SO$ = TEMP$
7626 TEMP$ = STR$ ((A1 / 4095) * SQR (V1)):S1 = (A1 / 4095)* SQR (V1)
7627 GOSUB }871
7628 S1$ = TEMP$
7629 TEMP$ = STR$ ((A0 / A1)* SQR (V2)):S2 = (A0 / A1)* SQR (V2)
7630 GOSUB }871
7631 S2$ = TEMP$
7635 IF KEY$ < > "I" THEN PRINT "MEAS #"; STR$ (LOOP); SPC( 03);"CHO INJECTION
CURRENT: ";J0$; SPC( 03);"CHO DATA: ";CO$; SPC( 03);"CH0 STD. DEV. ";S0$
7640 IF KEY$ < > "I" THEN PRINT SPC( 06); SPC( LEN ( STR$ (LOOP))); SPC(
03);"CH1 INJECTION CURRENT: ";J1$; SPC(03);"CH1 DATA: ";C1$; SPC(03);"CH1 STD.
DEV.";S1$
7650 IF KEY$ < > "I" THEN PRINT SPC( 09); SPC( LEN ( STR$ (LOOP)));"RATIO:
";RA$; SPC( 46);"RATIO STD. DEV. ";S2$
7660 IF KEY$ < > "I" THEN PRINT SPC( 09); SPC( LEN (STR$ (LOOP)));"%^RATIO:
";CHANGES
7670 IF KEY$ < > "I" THEN PRINT
7675 IF KEY$ = "I" THEN PRINT J0","J1","C0","C1","RA","S0","S1","S2
7680 IF PEEK (DS% + 102) > TIME% THEN PRINT "MEASUREMENT INTERVAL TOO
SHORT FOR SELECTED PARAMETERS ": GOTO 7710: REM START AGAIN
7690 IF PEEK (DS% + 102) < TIME% THEN 7690: REM WAIT FOR TIME.OUT COUNT
TO EXPIRE
7700 NEXT LOOP
7705 IF KEY$ = "I" THEN PRINT CD$;"CLOSE";FILE$
7710 POKE DS% + 102,0: REM CLEAR TIME.OUT COUNTER
7720 CALL INT (256* PEEK (DS% + 105) + PEEK (DS% + 106) + 0.5): REM DISABLE
CLOCK AND PROCESSOR INTERRUPTS
7730 IF KEY$ < > "I" THEN PRINT CHR$ (140)
7740 IF KEY$ < > "I" THEN PRINT CD$;"PR# 3": REM REACTIVATE 80 COLUMN
7750 IF KEY$ < > "I" THEN GOTO 5480
7755 IF KEY$ = "I" THEN GOTO 3200
7760 REM *************************************************************
770 HOME
7 7 8 0 \text { VTAB } 7
7790 PRINT SPC( 20);"BE SURE THE PRINTER IS 'ON LINE'"
7800 VTAB }1
7810 PRINT SPC( 10);"IF YOU DESIRE TO SEE THE DISK CONTENTS BEFORE
NAMING"
7820 PRINT SPC( 10);"THE FILE, SIMPLY ENTER '?' FOR THE FILENAME."
7830 VTAB }1
7840 HTAB }2
7850 INPUT "ENTER THE DESIRED DATA FILE NAME ";FLLE$
7860 IF FILE$ < > "?" THEN 7930
7870 HOME
7880 PRINT CD$;"CATALOG,S6,D2"
7890 PRINT
7900 HTAB 20
7910 INPUT "PRESS RETURN TO CONTINUE ";BOGUS$
7920 GOTO 7770
7930 VTAB 20
7940 PRINT SPC( 15);"PRINTING THE DATA FILE STATISTICAL RESULTS "
```

```
7950 PRINT CD$;"PR# 1": REM ACTIVATE PRINTER
7960 PRINT CD$;"OPEN ";FILE$;",S6,D2"
7970 PRINT CD$;"READ ";FIES
7980 GOSUB 6180: REM PRINT SYSTEM PARAMETERS
7990 PRINT
8000 M0 = 0.0: REM MEAN VALUE
8010 M1 = 0.0
8011 M2 = 0.0
8012 M3 = 0.0
8013 M4 = 0.0
8020 L0 = 10000: REM LOW VALUE
8030 H0 = - 10000: REM HIGH VALUE
8040 L1 = 10000
8050 H1 = - 10000
8051 L2 = 1.0E15
8052 H2 = - 1.0E15
8053 L3 = 10000
8054 H3 =-10000
8055 L4 = 10000
8056 H4 = - 10000
8060 FOR LOOP = 1 TO MEAS%: REM MEAS% SAMPLES
8070 INPUT J0,J1,C0,C1,RA,S0,S1,S2
8080 ST(0,LOOP) = C0
8090 ST(1,LOOP)=C1
8091 ST(2,LOOP) = RA
8092 ST(3,LOOP) = J0
8093 ST(4,LOOP) = J1
8100 M0 = M0 + ST(0,LOOP)
8110 M1 = M1 + ST(1,LOOP)
8111 M2 = M2 + ST(2,LOOP)
8112 M3 = M3 + ST(3,LOOP)
8113 M4 = M4 + ST(4,LOOP)
8120 IF ST(0,LOOP) < LO THEN LO = ST(0,LOOP)
8130 IF ST(0,LOOP) > H0 THEN H0 = ST(0,LOOP)
8140 IF ST(1,LOOP) < L1 THEN L1 = ST(1,LOOP)
8150 IF ST(1,LOOP) > H1 THEN H1 = ST(1,LOOP)
8151 IF ST(2,LOOP) < L2 THEN L2 = ST(2,LOOP)
8152 IF ST(2,LOOP) > H2 THEN H2 = ST(2,LOOP)
8153 IF ST(3,LOOP) < L3 THEN L3 = ST(3,LOOP)
8154 IF ST(3,LOOP) > H3 THEN H3 = ST(3,LOOP)
8155 IF ST(4,LOOP) < L4 THEN L4 = ST(4,LOOP)
8156 IF ST(4,LOOP) > H4 THEN H4 = ST(4,LOOP)
8 1 6 0 ~ N E X T ~ L O O P
8170 PRINT CD$;"CLOSE ";FILE$
8180 M0 = M0 / MEAS%
8190 M1 = M1 / MEAS%
8191 M2 = M2 / MEAS%
8192 M3 = M3 / MEAS%
8193 M4 = M4 / MEAS%
8200 V0 = 0.0: REM CHANNEL 0 VARIANCE
8210 V1 = 0.0: REM CHANNEL 1 VARIANCE
8211 V2 =0.0: REM RATIO VARIANCE
8212 V3 = 0.0: REM CHANNEL O INJECTION VARIANCE
8213 V4 = 0.0: REM CHANNEL 1 INJECTION VARIANCE
```

```
8220 FOR LOOP = 1 TO MEAS%
8230 V0 = V0 + (ST(0,LOOP) - M0) * (ST(0,LOOP) - M0)
8240 V1 = V1 + (ST(1,LOOP) - M1) * (ST(1,LOOP) - M1)
8241 V2 = V2 + (ST(2,LOOP) - M2) * (ST(2,LOOP) - M2)
8242 V3 = V3 + (ST(3,LOOP) - M3) * (ST(3,LOOP) - M3)
8243 V4 = V4 + (ST(4,LOOP) - M4) * (ST(4,LOOP) - M4)
8250 NEXT LOOP
8260 V0 = V0 /(MEAS% - 1)
8270 V1 = V1 / (MEAS% - 1)
8271 V2 = V 2 / (MEAS% - 1)
8272 V3 = V3 / (MEAS% - 1)
8273 V4 = V4 / (MEAS% - 1)
8280 S0 = SQR (V0): REM CHANNEL O STANDARD DEVIATION
8290 S1 = SQR (V1): REM CHANNEL 1 STANDARD DEVIATION
8291 S2 = SQR (V2): REM RATIO STANDARD DEVIATION
8292 S3 = SQR (V3): REM CHANNEL O INJECTION STANDARD DEVIATION
8293 S4 = SQR (V4): REM CHANNEL 1 INJECTION STANDARD DEVIATION
8300 TEMP$ = STR$ (LO)
8 3 1 0 \text { GOSUB } 8 7 1 0
8320 LO$ = TEMP$
8330 TEMP$ = STR$ (L1)
8 3 4 0 \text { GOSUB } 8 7 1 0
8350 L1$ = TEMP$
8351 TEMP$ = STR$ (L2)
8352 GOSUB }871
8353 L2$ = TEMP$
8354 TEMP$ = STR$ (L3)
8355 GOSUB }871
8356 L3$ = TEMP$
8357 TEMP$ = STR$ (L4)
8 3 5 8 \text { GOSUB } 8 7 1 0
8359 L4$ = TEMP$
8360 TEMP$ = STR$ (H0)
8 3 7 0 \text { GOSUB } 8 7 1 0
8380 HO$ = TEMP$
8390 TEMP$ = STR$ (H1)
8400 GOSUB }871
8410 H1$ = TEMP$
811 TEMP$ = STR$ (H2)
8412 GOSUB }871
8413 H2$ = TEMP$
8414 TEMP$ = STR$ (H3)
8 4 1 5 \text { GOSUB } 8 7 1 0
8416 H3$ = TEMP$
8417 TEMP$ = STR$ (H4)
8 4 1 8 \text { GOSUB } 8 7 1 0
8419 H4$ = TEMP$
8420 TEMP$ = STR$ (M0)
8430 GOSUB }871
8440 MO$ = TEMP$
8450 TEMP$ = STR$ (M1)
8460 GOSUB }871
8470 M1$ = TEMP$
8471 TEMP$ = STR$ (M2)
```

```
8472 GOSUB }871
8473 M2$ = TEMP$
8474 TEMP$ = STR$ (M3)
8 4 7 5 \text { GOSUB } 8 7 1 0
8476 M3$ = TEMP$
8477 TEMP$ = STR$ (M4)
8478 GOSUB }871
8479 M4$ = TEMP$
8480 TEMP$ = STR$ (V0)
8490 GOSUB }871
8500 V0$ = TEMP$
8510 TEMP$ = STR$ (V1)
8520 GOSUB }871
8530 V1$ = TEMP$
8531 TEMP$ = STR$ (V2)
8532 GOSUB }871
8533 V2$ = TEMP$
8534 TEMP$ = STR$ (V3)
8535 GOSUB }871
8536 V3$ = TEMP$
8537 TEMP$ = STR$ (V4)
8 5 3 8 \text { GOSUB } 8 7 1 0
8539 V4$ = TEMP$
8540 TEMP$ = STR$ (S0)
8550 GOSUB }871
8560 SO$ = TEMP$
8570 TEMP$ = STR$ (S1)
8580 GOSUB }871
8 5 8 1 ~ S 1 \$ ~ = ~ T E M P \$ ~
8582 TEMP$ = STR$ (S2)
8583 GOSUB }871
8584 S2$ = TEMP$
8585 TEMP$ = STR$ (S3)
8586 GOSUB }871
8587 S3$ = TEMP$
8588 TEMP$ = STR$ (S4)
8 5 8 9 \text { GOSUB } 8 7 1 0
8590 S4$ = TEMP$
8594 PRINT
8596 PRINT "PARAMETER"; SPC( 11);"CH0 INJECTION"; SPC( 10);"CH1 INJECTION";
SPC( 10);"CHANNEL O DATA"; SPC( 10);"CHANNEL 1 DATA"; SPC( 10);"RATIO"
8597 PRINT
8598 PRINT "MINIMUM"; SPC( 13);L3$; SPC( 09);L4$; SPC( 09);L0$; SPC( 10);L1$; SPC(
10;[L2$
8599 PRINT "MAXIMUM"; SPC( 13);H3$; SPC( 09);H4$; SPC( 09);HO$; SPC( 10);H1$;
SPC( 10);H2$
8600 PRINT
8601 PRINT "MEAN"; SPC( 16);M3$; SPC( 09);M4$; SPC(09);M0$; SPC( 10);M1$; SPC(
10);M2$
8602 PRINT "VARIANCE"; SPC( 12);V3$; SPC( 09);V4$; SPC( 09);V0$; SPC( 10);V1$; SPC(
10);V2$
8603 PRINT "STANDARD DEV."; SPC( 07);S3$; SPC( 09);S4$; SPC( 09);S0$; SPC( 10);S1$;
SPC( 10);S2$
8660 GOSUB }688
```

```
8670 PRINT CHR$ (140)
8680 PRINT CD$;"PR# 3": REM REACTTVATE }80\mathrm{ COLUMN
8690 GOTO 5480
8700 REM ****************************************************************
8710 LGTH% = LEN (TEMP$)
8720 IF LGTH% = 14 THEN RETURN
8730 FOR PAD = 1 TO 14-LGTH%
8740 TEMP$ = TEMP$ + " "
8750 NEXT PAD
8760 RETURN
8770 REM *************************************************************
8780 LOOP$ = STR$ (LOOP)
8790 LGTH% = LEN (LOOP$)
8800 IF LGTH% = 4 THEN RETURN
8810 FOR PAD = 1 TO 4-LGTH%
8820 LOOP$ = LOOP$ + " "
8 8 3 0 ~ N E X T ~ P A D ~
8840 RETURN
8850 REM
**********************************************************************
880 HOME
8870 VTAB 3
8880 PRINT SPC( 24);"PHOTOMULTIPLIER GAIN SELECTION"
8890 VTAB }
8900 PRINT SPC( 24);"THE GAIN OF EACH PHOTOMULTIPLIER "
8910 PRINT SPC(24);"CAN BE ADJUSTED INDEPENDENTLY AND "
8920 PRINT SPC(24);"MUST BE IN THE INTERVAL 4.0-2.0E06 "
8930 VTAB }1
8940 HTAB 25
8950 INPUT "ENTER THE GAIN FOR PHOTOMULTIPLIER CH0 ";G0
8960 IF GO < 4.0 THEN }893
8970 IF G0 > 2.0E06 THEN }893
890 VTAB }1
890 HTAB 25
9000 INPUT "ENTER THE GAIN FOR PHOTOMULTIPLIER CH1 ";G1
9010 IF G1 < 4.0 THEN }898
9020 IF G1 > 2.0E06 THEN }898
9030 ZTO = (.434294482 * LOG (G0 / 3.1622776E - 17))/7.5: REM HV CHO
9040 O0 = 10^ ZT0
9050 ZT1 = (.434294482 * LOG (G1 / 3.1622776E - 17)) /7.5: REM HV CH1
9060 O1 = 10^ ZT1
9070 R0=(1.6127616E - 18)* O0^ 7.5: REM CH0 ANODE RADIANT SENSITIVITY
9080 R1 = (1.6127616E-18)* O1^7.5: REM CH1 ANODE RADIANT SENSITIVITY
9090 W0 = 1.0E-05/R0: REM MAXIMUM CHO INPUT INTENSITY
9100 W1 = 1.0E-05/R1: REM MAXIMUM CH1 INPUT INTENSITY
9110 Z0 = (-O0 + 1100) /650: REM REFERENCE VOLTAGE CHO
9120 Z1 = (-01 + 1100)/650: REM REFERENCE VOLTAGE CH1
9130 X0% = INT ((255* Z0 / 1.423670669) +.5): REM CH0 GAIN CONTROL VALUE
9140 X1% = INT ((255* Z1 / 1.423670669) +.5): REM CH1 GAIN CONTROL
9150 VTAB }1
9160 PRINT SPC(41);"CH0"; SPC( 16);"CHI"
9170 VTAB }1
9180 GOSUB 9560: REM FORMAT
9190 TEMP$ = STR$ (- O0)
```

9200 GOSUB 8710
9210 O0\$ = TEMPS
9220 TEMP $\$=$ STR\$ ( -01 )
9230 GOSUB 8710
9240 O1\$ = TEMP\$
9250 TEMP $=$ STR\$ (R0)
9260 GOSUB 8710
9270 RO\$ = TEMP\$
9280 TEMP\$ = STR\$ (R1)
929 GOSUB 8710
9300 R1\$ = TEMP\$
9310 TEMP\$ = STR\$ (W0)
9320 GOSUB 8710
9330 W0 $\$=$ TEMP $\$$
9340 TEMP\$ $=$ STR\$ (W1)
9350 GOSUB 8710
9360 W1\$ = TEMP\$
9370 TEMP\$ = STR\$ (ZO)
9380 GOSUB 8710
9390 Z0\$ = TEMPS
9400 TEMP\$ = STR\$ (Z1)
9410 GOSUB 8710
9420 Z1\$ = TEMP\$
9430 PRINT SPC( 5);"CURRENT GAIN: "; SPC( 17);G0\$; SPC( 5);G1\$
9440 PRINT SPC(5);"ANODE SENSITIVITY (A/W): "; SPC( 06);R0\$; SPC( 5);R1\$
9450 PRINT SPC( 5);"REFERENCE VOLTAGE (V): "; SPC( 8);Z0\$; SPC( 5);Z1\$
9460 PRINT SPC( 5 );"HIGH VOLTAGE (V): "; SPC( 13);O0\$; SPC( 5);O1\$
9470 PRINT SPC( 5);"MAXIMUM INTENSITY (W): "; SPC( 8);W0\$; SPC( 5);W1\$
9480 PRINT
9490 HTAB 25
9500 INPUT "ARE THESE PARAMETERS OK (Y/N)? ";KEY\$
9510 IF KEYS = "N" THEN 8860: REM IF NO START OVER
9520 POKE DS\% + 28,X0\%: REM WRITE CHO CONTROL BYTE
9530 POKE DS\% +29,X1\%: REM WRITE CHI CONTROL BYTE
9540 RETURN
9550 REM

9560 TEMP\$ = STR\$ (G0)
9570 GOSUB 8710
9580 G0\$ = TEMP\$
9590 TEMP\$ = STR\$ (G1)
9600 GOSUB 8710
9610 G1\$ = TEMP\$
9620 RETURN
9630 REM


## APPENDIX D ASSEMBLY (6502) CONTROL PROGRAM



| $\begin{aligned} & 0000: \\ & 0000: \end{aligned}$ | CODO | 47 | AD.LOW | EQU | \$CODO |  | :LOW BYTE A/D DATA |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 0000: | COD1 | 49 | AD. HIGH | EQU | \$COD1 |  | : HIGE BYTE A/D DATA |
| 0000: | COD2 | 50 | START.CONV | EQU | \$COD2 |  | :START CONVERSION ADDRESS |
| 0000: | COD3 | 51 | AD. CTRU | EQU | \$COD3 |  | ;A/D CONTROL BYTE |
| 0000: |  | 52 | ; |  |  |  |  |
| 0000: |  | 53 | ;*** APPLIED |  | ENGINEERING | G TIME | EMASTER II - SLOT 4 *** |
| 0000: |  | 54 | DRA.CLK |  |  |  |  |
| 0000: | COCO | 55 |  | EQU | \$COCO |  | ;DATA/DIRECTION REGISTER |
| A |  |  |  |  |  |  |  |
| 0000: | $\mathrm{COC1}$ | 56 | CRA.CIK | EQU | \$COC1 |  | : CONTROL REGISTER A |
| 0000: | COC2 | 57 | DRB.CLK | EQU | \$COC2 |  | ; DATA/DIRECTION REGISTER |
| B |  |  |  |  |  |  |  |
| 0000: | COC3 | 58 | CRB.CLK | EQU | \$COC3 |  | - CONTROL REGISTER B |
| 0000: |  | 59 | : |  |  |  |  |
| 0000: | B800 | 60 | SCRATCH | EQU | \$8800 |  | ; GENERAL SCRATCH PAD AREA |
| 0000: | B900 | 61 | SCRATCHO | EQU | \$8900 |  | ; SCRATCH PAD AREA CHO |
| 0000: | BA00 | 62 | SCRATCHO.A | EQU | \$BA00 |  | -CHO AUXILIARY AREA |
| 0000: | BB00 | 63 | SCRATCH1 | EQU | \$BB00 |  | ; SCRATCH PAD AREA CHI |
| 0000: | BCOO | 64 | SCRATCHI.A | EQU | \$BCOO |  | CH1 AUXILILARY AREA |
| 0000: | BDOO | 65 | RAT.INT | EQU | \$BD00 |  | ;RATIO INTEGER PART |
| 0000: | BEOO | 66 | RAT.FRAC | EQU | \$BE00 |  | ;RATIO FRACTIONAI PART |
| 0000: |  | 67 | : |  |  |  |  |
| 0000: | FDED | 68 | COUT | EQU | \$FDED |  | ;FIRMWARE ROUTINE "COUT" |
| 0000: |  | 69 | ; |  |  |  |  |
| 0000: |  | 70 | ; ************** |  | DUMMY | SECTO | OR ****************** |
| 0000: |  | 71 | ; |  |  |  |  |
| 0000: |  | 72 | DSECT |  |  |  |  |
| A500: | A500 | 73 |  | ORG | \$A500 |  |  |
| A500: | 0001 | 74 | PHASE1.H | DS | 1 |  | ;PHASE 1 STARTING ADDRESS |
| A501: | 0001 | 75 | PHASE1.L | DS | 1 |  |  |
| A502: | 0001 | 76 | PHASE2.H | DS | 1 |  | ;PHASE 2 STARIING ADDRESS |
| A503: | 0001 | 77 | PHASE2.L | DS | 1 |  |  |
| A504: | 0001 | 78 | PHASE3.H | DS | 1 |  | ;PHASE 3 STARIING ADDRESS |
| A505: | 0001 | 79 | PHASE3.L | DS | 1 |  |  |
| A506: | 0001 | 80 | PHASE4.H | DS | 1 |  | ;PHASE 4 STARTING ADDRESS |
| A507: | 0001 | 81 | PHASE4.L | DS | 1 |  |  |
| A508: | 0001 | 82 | FLASH.F | DS | 1 |  | ; FLASH LAMP FREQUENCY |
| A509: | 0001 | 83 | INT. PERIOD | DS | 1 |  | ; INTEGRATION PERIOD |
| LENGTH |  |  |  |  |  |  |  |
| A50A: | 0001 | 84 | INT. NUM | DS | 1 |  | ; NO. MULTIPLE |
| INTEGRATIONS |  |  |  |  |  |  |  |
| A50B: | 0001 | 85 | INT. CNT | DS | 1 |  | ; INTEGRATION COUNT |
| A50C: | 0001 | 86 | SAMPLE.NUM | DS | 1 |  | ;NO. SAMPLES DESIRED |
| A50D: | 0001 | 87 | SAMPLE.CNT | DS | 1 |  | ; SAMPLE COUNT |
| A50E: | 0001 | 88 | MANUAL | DS | 1 |  | -MANUAL CALIBRATION FLAG |
| A50F: | 0001 | 89 | CHO.PGAIN | DS | 1 |  | ChO Programmable Gain |
| A510: | 0001 | 90 | CHO.AD.GAIN | DS | 1 |  | CHANNEL 0 A/D GAIN |
| A511: | 0001 | 91 | CHI.PGAIN | DS | 1 |  | CHI PROGRAMMABLE GAIN |
| A512: | 0001 | 92 | CH1.AD.GAIN | DS | 1 |  | CHANNEL 1 A/D GATN |
| A513: | 0001 | 93 | AD.CHO.CTRL | DS | 1 |  | A/D CHANNEL 0 CONTROL |
| A514: | 0001 | 94 | AD.CH1.CTRL | DS | 1 |  | A/D CHANNEL 1 CONTROL |
| A515: | 0001 | 95 | CHO. INJ. H | DS | 1 |  | ; ${ }^{\text {H B TE }}$ INJECTION CURRENT |
| A516: | 0001 | 96 | CHO.INJ.L | DS | 1 |  | L BYTE INJECTION CURRENT |
| A517: | 0001 | 97 | CH1. INJ. H | DS | 1 |  | H BYTE INJECTION CURRENT |
| A518: | 0001 | 98 | CH1.INJ.L | DS | 1 |  | L BYTE INJECTION CURRENT |
| A519: | 0001 | 99 | MEAS.MODE | DS | 1 |  | MEASUREMENT ROUTINE |
| CHOICE |  |  |  |  |  |  |  |


| A51A: | 0001 | 100 | OEFSET.ADJ | DS | 1 | :OFFSET ADJUSTMENT ELAG |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| A51B: | 0001 | 101 | DA.COPY1 | DS | 1 | :D/A CEI CONTROL COPY |
| A51C: | 0001 | 102 | DA.COPY2 | DS | 1 | :D/A CH2 CONTROL COPY |
| A51D: | 0001 | 103 | DA.COPY3 | DS | 1 | :D/A CB3 CONTROL COPY |
| A51E: | 0001 | 104 | DA.COPY4 | DS | 1 | :D/A CH4 CONTROL CORY |
| A51F: | 0001 | 105 | DA.COPY5 | DS | 1 | ;D/A CH5 CONTROL COPY |
| A520: | 0001 | 106 | OF.SW1 | DS | 1 | ;OVERFLOW ERROR FLAGS |
| A521: | 0001 | 107 | OF.SW2 | DS | 1 |  |
| A522: | 0001 | 108 | OF.SW3 | DS | 1 | ; FLAG IS SET IF AN |
| A523: | 0001 | 109 | OF.SW4 | DS | 1 | :OVERFLON WAS DETECTED |
| A524: | 0001 | 110 | OF.SW5 | DS | 1 | ;DURING MEASUREMIENT |
| A525: | 0001 | 111 | OF.SW6 | DS | 1 | ; SWEEP |
| A526: | 0001 | 112 | OF.SW7 | DS | 1 |  |
| A527: | 0001 | 113 | OF.SW8 | DS | 1 |  |
| A528: | 0001 | 114 | OE.SW9 | DS | 1 |  |
| A529: | 0001 | 115 | OF.SW10 | DS | 1 |  |
| A52A: | 0001 | 116 | UF.SW1 | DS | 1 | ;UNDERFLOW ERROR FLAGS |
| A52B: | 0001 | 117 | UF.SW2 | DS | 1 |  |
| A52C: | 0001 | 118 | UF.SW3 | DS | 1 | ; FLAG IS SET IF AN |
| A52D: | 0001 | 119 | UF.SW4 | DS | 1 | ;UNDERFLOW WAS DETECTED |
| A52E: | 0001 | 120 | UF.SW5 | DS | 1 | ; DURING MEASUREMENT |
| A52F: | 0001 | 121 | UF.SW6 | DS | 1 | ; SWEEP |
| A530: | 0001 | 122 | UF. SW7 | DS | 1 |  |
| A531: | 0001 | 123 | UF.SW8 | DS | 1 |  |
| A532: | 0001 | 124 | UF. SW9 | DS | 1 |  |
| A533: | 0001 | 125 | UF. SW10 | DS | 1 |  |
| A534: | 0001 | 126 | SWEEP.CNT | DS | 1 | : SWEEP COUNT |
| A535: | 0001 | 127 | SMPL .AVAIL | DS | 1 | ; SAMPLE AVAILABLE FLAG |
| A536: | 0001 | 128 | INJ.ACTIVE | DS | 1 | ; INJECTION CURRENT FLAG |
| A537: | 0001 | 129 | INJ.STATUS | DS | 1 | ; LEVEL STATUS INDICATOR |
| A538: | 0001 | 130 | MEAS . ACTIVE | DS | 1 | ; GENERAL MEASUREMENT FLAG |
| A539: | 0001 | 131 | DATA.ACTIVE | DS | 1 | ; DATA MEASUREMENT FLAG |
| A53A: | 0001 | 132 | SYNC. OK | DS | 1 | ; SYNCHRONIZATION FIAG |
| A53B: | 0001 | 133 | DATA.SETTIE | DS | 1 | ;DATA SETTLING COUNT |
| A53C: | 0001 | 134 | FLASH. H | DS | 1 | ;FLASHLAMP TIMER VALUES |
| A53D: | 0001 | 135 | FIASH.L | DS | 1 |  |
| A53E: | 0001 | 136 | INT.LOOP | DS | 1 | ;INTEGRATION LOOP |
| VARIABLE |  |  |  |  |  |  |
| A53F: | 0001 | 137 | MEAS.TEMP. H | DS | 1 | ; TEMPORARY MEAS. |
| VARIABLES |  |  |  |  |  |  |
| A540: | 0001 | 138 | MEAS.TEMP.I | DS | 1 |  |
| A541: | 0001 | 139 | REMAIN.H | DS | 1 | ; HIGH BYTE REMAINDER |
| A542: | 0001 | 140 | REMAIN.MH | DS | 1 |  |
| A543: | 0001 | 141 | REMAIN.M | DS | 1 |  |
| A544: | 0001 | 142 | REMAIN.ML | DS | 1 |  |
| A545: | 0001 | 143 | REMAIN.L | DS | 1 | ;LOW BYTE REMAINDER |
| A546: | 0001 | 144 | DIV.QUOT.H | DS | 1 | ; H BYTE DIVIDEND/QUOTIENT |
| A547: | 0001 | 145 | DIV.QUOT.MH | DS | 1 |  |
| A548: | 0001 | 146 | DIV.QUOT.M | DS | 1 |  |
| A549: | 0001 | 147 | DIV.QUOT.ML | DS | 1 |  |
| A54A: | 0001 | 148 | DIV.QUOT.L | DS | 1 | ; $L$ BYTE LIVIDEND/QUOTIENT |
| A54B: | 0001 | 149 | DIVISOR.H | DS | 1 | ;HIGH BYTE DIVISOR |
| A54C: | 0001 | 150 | DIVISOR.MH | DS | 1 |  |
| A54D: | 0001 | 151 | DIVISOR.M | DS | 1 |  |
| A54E: | 0001 | 152 | DIVISOR.ML | DS | 1 |  |
| A54F: | 0001 | 153 | DIVISOR.L | DS | 1 | ;LOW BYTE DIVISOR |
| A550: | 0001 | 154 | DIV.TEMPX | DS | 1 | ; COPY OF X REGISTER |
| A551: | 0001 | 155 | DIV.TEMP.MH | DS | 1 | ; TEMPORARY VARIABLES |


| A552: | 0001 | 156 | DIV.TEMP.M | DS | 1 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| A553: | 0001 | 157 | DIV.TEMP.ML | DS | 1 |  |
| AS54: | 0001 | 158 | DIV.TEMP.L | DS | 1 | - |
| A555: | 0001 | 159 | CHO.SUM. H | DS | 1 | : CHANNEL 0 |
| A556: | 0001 | 160 | CHO.SUM.M | DS | 1 | : SUM LOCATIONS |
| A557: | 0001 | 161 | CHO.SUM.L | DS | 1 |  |
| A558: | 0001 | 162 | CH1.SUM. H | DS | 1 | : CHANNEL 1 |
| A559: | 0001 | 163 | CE1.SUM.M | DS | 1 | :SUM LOCATIONS |
| A55A: | 0001 | 164 | CH1.SUM.L | DS | 1 |  |
| A55B: | 0001 | 165 | RAT.SUM.H | DS | 1 | ;RATIO SUM IOCATIONS |
| A55C: | 0001 | 166 | RAT.SUM.MR | DS | 1 |  |
| A55D: | 0001 | 167 | RAT.SUM.M | DS | 1 |  |
| A55E: | 0001 | 168 | RAT.SUM.ML | DS | 1 |  |
| A55F: | 0001 | 169 | RAT.SUM.I | DS | 1 |  |
| A560: | 0001 | 170 | INVALID | DS | 1 | :SYSTEM ERROR EIAG |
| A561: | 0001 | 171 | DLY.TEMPX | DS | 1 | ; COPY OF X REGISTER |
| A562: | 0001 | 172 | DLY. TEMPY | DS | 1 | ; COPY OF Y REGISIER |
| A563: | 0001 | 173 | DELAY | DS | 1 | ;DELAY PARAMETER |
| A564: | 0001 | 174 | CLK. CNITRL.A | DS | 1 ; | ;CLOCK CONTROL REGISTER A |
| CORY |  |  |  |  |  |  |
| A565: | 0001 | 175 | CLK. CNTRL. ${ }^{\text {B }}$ | DS | 1 ; | ; CLOCK CONTROL REGISTER B |
| COPY |  |  |  |  |  |  |
| A566: | 0001 | 176 | TIME.OUT | DS | 1 ; | ; CLOCK INTERVAL TIME UP |
| COUNT |  |  |  |  |  |  |
| A567: | 0001 | 177 | CLIKSTART. H | DS | 1 ; | ; CLKSTART STARTING ADDRESS |
| A568: | 0001 | 178 | CLKSTART.L | DS | 1 |  |
| A569: | 0001 | 179 | CLKSTOP. H | DS | 1 ; | ;CLKSTOP STARTING ADDRESS |
| A56A: | 0001 | 180 | CIKSTOR.L | DS | 1 |  |
| A56B: | 0001 | 181 | ENABLE. H | DS | 1 ; | ; INTERRUPT ENABLE STARTING |
| ADDRESS |  |  |  |  |  |  |
| A56C: | 0001 | 182 | ENABLE.L | DS | 1 |  |
| A56D: | 0001 | 183 | DUMMY.FLASH | DS | 1 | ;DUMMY FLASHES TO ACHIEVE |
| S.S. |  |  |  |  |  |  |
| 0000: |  | 184 |  | DEND |  |  |
| 0000: |  | 185 |  |  |  |  |
| 0000: |  | 186 | ;********** |  | INITIALIZATION | N ROUTINE ************ |
| - NEXT | OBJECT | FILE | NAME IS JMCl | .0.0 | BJO |  |
| A600: | A600 | 187 |  | ORG | \$A600 |  |
| A600: 78 |  | 188 |  | SEI |  | ;DISABLE PROCESSOR IRQ'S |
| A601:A9 B5 |  | 189 |  | LDA | \#<ISR | ; INTERRUPT VECTOR POINTER |
| A603:8D FF | 03 | 190 |  | STA | \$03FF | ; MSB |
| A606:A9 18 |  | 191 |  | LDA | \#>ISR |  |
| A608:8D FE | 03 | 192 |  | STA | \$03FE | ; LSB |
| A60B:A9 A6 |  | 193 |  | IDA | \#<PHASE1 | ;PHASE1 STARTING ADDRESS |
| A60D: 8D 00 | A5 | 194 |  | STA | PHASEI.H |  |
| A610:A9 B4 |  | 195 |  | IDA | \# $>$ PHASE1 |  |
| A612:8D 01 | A5 | 196 |  | STA | PHASEI.L |  |
| A615:A9 A8 |  | 197 |  | LDA | \#<PHASE2 | ;PHASE2 STARTING ADDRESS |
| A617:8D 02 | A5 | 198 |  | STA | PHASE2.H |  |
| A61A:A9 2D |  | 199 |  | IDA | \# $>$ PHASE 2 |  |
| A61C:8D 03 | A5 | 200 |  | STA | PHASE2.I |  |
| A61F:A9 AA |  | 201 |  | LDA | \#<PHASE3 | ;PHASE3 STARTING ADDRESS |
| A621:8D 04 | A 5 | 202 |  | STA | PHASE3.H |  |
| A624:A9 00 |  | 203 |  | IDA | \# $>$ PHASE3 |  |
| A626:8D 05 | A5 | 204 |  | STA | PHASE3.L |  |
| A629:A9 AB |  | 205 |  | IDA | \#<PHASE4 | ;PHASE4 STARTING ADDRESS |
| A62B:8D 06 | A5 | 206 |  | STA | PHASE4.H |  |
| A62E:A9 61 |  | 207 |  | IDA | \#>PHASE4 |  |






| A82D: |  |  | 437 | ; |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| A82D:AD | OE |  | 438 | PHASE2 | LDA | MANUAL | ; CEECK FOR MANUAL CALIB. |
| A830:F0 | 11 | A843 | 439 |  | BEQ | AUTO | ; NO |
| A832:C9 | 01 |  | 440 |  | CMP | \# 01 |  |
| A834:F0 | 09 | A83F | 441 |  | BEQ | MAN | ; YES |
| A836:A9 | 80 |  | 442 |  | IDA | \#\%10000000 | ; ERRONEOUS CALL TO PHASE |
| 2 |  |  |  |  |  |  |  |
| A838:8D | 60 | 5 | 443 |  | STA | INVALID | ; SYSTEM ERROR BYTE |
| A83B:20 | 7A B | 4 | 444 |  | JSR | ERROR | :SYSTEM ERROR ROUTINE |
| A83E: 60 |  |  | 445 |  | RTS |  | : RETURN FROM BASIC CALI |
| A83F:20 | 58 | 8 | 446 | MAN | JSR | MAN.CAL | :SET UP CONTROL BYTES |
| A842:60 |  |  | 447 |  | RTS |  | : RETURN FROM BASIC CALI |
| A843:20 | 7C | 8 | 448 | AUTO | JSR | AUTO. CALO | ; CAIIBRATE CHANNEL 0 |
| A846:20 | 3E | 9 | 449 |  | JSR | AUTO.CALI | ; CALIBRATE CHANNEL 1 |
| A849:AD | OF A | 5 | 450 |  | IDA | CHO.PGAIN | ;SET UP PROGRAMMABLE GAIN |
| A84C: 0 D | 11 A | 5 | 451 |  | ORA | CH1.PGAIN |  |
| A84F: 8D | 01 | 2 | 452 |  | STA | U1.DRA |  |
| A852:A9 | 00 |  | 453 |  | IDA | \#\%00000000 | ;TRIGGER LEVEL LOW |
| A854:8D | 80 C | 2 | 454 |  | STA | U2.DRB |  |
| A857:60 |  |  | 455 |  | RTS |  | ;RETURN FROM BASIC CALU |
| A858: |  |  | 456 | : |  |  |  |
| A858: |  |  | 457 | ;***** | MANUAI | SYSTEM CALIE | BRATION ROUTINE ****** |
| A858: |  |  | 458 | ; |  |  |  |
| A858:AD | OF | 5 | 459 | MAN.CAL | LDA | CHO.PGAIN | ;SET UP PROGRAMMABLE GAIN |
| A85B: 0 D | 11 A | 5 | 460 |  | ORA | CH1.PGAIN |  |
| A85E: 8D | 01 C | 2 | 461 |  | STA | U1.DRA |  |
| A861:AD | 10 A | 5 | 462 |  | LDA | CHO.AD.GAIN | ; SET UP CHO A/D CONTROL |
| A864:2A |  |  | 463 |  | ROL |  | ; INFORMATION CONTAINED IN |
| A865:2A |  |  | 464 |  | ROL |  | ;LOW ORDER NIBBLE |
| A866:2A |  |  | 465 |  | ROL |  |  |
| A867:18 |  |  | 466 |  | CLC |  | ; SET MULTIPLEXER CHANNEL |
| A868:2A |  |  | 467 |  | ROL |  |  |
| A869:29 | 71 |  | 468 |  | AND | \#\%01110001 |  |
| A86B: 8D | 13 A | 5 | 469 |  | STA | AD.CHO.CTRL | :A/D CHO CONTROL BYTE |
| A86E:AD | 12 A |  | 470 |  | LDA | CHI.AD.GAIN | ; SET UP CHI A/D CONTROL |
| A871:2A |  |  | 471 |  | ROL |  | : INFORMATION CONTAINED IN |
| A872:2A |  |  | 472 |  | ROL |  | ; LOW ORDER NIBBLE |
| A873:2A |  |  | 473 |  | ROL |  |  |
| A874:38 |  |  | 474 |  | SEC |  | :SET MULTIPLEXER CHANNEL |
| A875:2A |  |  | 475 |  | ROI |  |  |
| A876:29 | 71 |  | 476 |  | AND | \#\%01110001 |  |
| A878:8D | 14 A |  | 477 |  | STA | AD.CH1.CTRL | ;A/D CH1 CONTROL BYTE |
| A87B: 60 |  |  | 478 |  | RTS |  |  |
| A87C: |  |  | 479 | ; |  |  |  |
| A87C: |  |  | 480 | ;**** | AU'TOMATI | IC GAIN CALIBR | RATION CHO ROUTINE **** |
| A87C: |  |  | 481 |  |  |  |  |
| A87C:AD | OA A |  | 482 | AUTO.CALO | IDA | INT.NUM | :MULT. INTEGRATION COUNT |
| A87F:8D | OB A |  | 483 |  | STA | INT. CNT |  |
| A882:A9 | 00 |  | 484 |  | LDA | \#00 | ; S/H DATA AVAILABLE |
| A884:8D | 35 A |  | 485 |  | STA | SMPL.AVAIL | ;FLAG INITIALLY ZERO |
| A887:A9 | 88 |  | 486 |  | LDA | \#\%10001000 | ;CHO INITIAL GAIN $=2$ |
| A889:8D | 01 C |  | 487 |  | STA | U1.DRA |  |
| A88C: 20 | 58 B |  | 488 |  | JSR | OUT.SETTLE | ; OUTPUT MUST SETTLE |
| A88F:20 | 2D A |  | 489 |  | JSR | PRE.MEAS | ; PRE-MEASUREMENT |
| A892:A9 | 10 |  | 490 | RANGE. CALO | 0 IDA | \#\%00010000 | ; CHO INIT A/D RANGE $=5 \mathrm{~V}$ |
| A894:8D | 13 A5 |  | 491 |  | STA | AD.CHO.CTRI |  |
| A897:8D | D3 C0 |  | 492 |  | STA | AD. CTRL | ;A/D CONTROL LOCATION |


| A89A:A9 | 31 |  | 493 |  | LDA | \#\%00110001 | :PLACE S/H IN "TRACK" |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| MODE |  |  |  |  |  |  |  |
| A89C: 8D | 81 | C2 | 494 |  | STA | U2.DRA |  |
| A89F:AD | 35 | A5 | 495 | WAIT1. CALO | LDA | SMPL. AVAIL | ; WAIT FOR SAMPLE |
| AVAIIABLE |  |  |  |  |  |  |  |
| A8A2:FO | FB | A89F | 496 |  | BEQ | WAITI. CALO |  |
| A8A4:20 | 6 F | B4 | 497 |  | JSR | CONVERT | ;PERFORM A/D CONVERSION |
| A8A7:AD | D1 | CO | 498 |  | IDA | AD. HIGE |  |
| A8AA:29 | OF |  | 499 |  | AND | \$\%00001111 | ; KEEP ONLY DATA BITS |
| A8AC:C9 | OF |  | 500 |  | CMP | \#\%00001111 | ;CHECK FOR A/D "OVERELOW" |
| A8AE:90 | 07 | A8B7 | 501 |  | BCC | CONT. CALO | ; NO |
| A8B0 : AD | D0 | CO | 502 |  | IDA | AD.ION |  |
| A8B3:C9 | EO |  | 503 |  | CMP | \#\%11100000 |  |
| A8B5: ${ }^{\text {O }}$ | 23 | A8DA | 504 |  | BCS | PGAIN.FIXO | ; YES |
| A8B7:AD | 13 | A5 | 505 | CONT. CALO | IDA | AD.CHO.CTRL |  |
| A8BA:C9 | 70 |  | 506 |  | CMP | \#\%01110000 | ; CHECK IF AT MIN. RANGE |
| A8BC:F0 | 27 | A8E5 | 507 |  | BEQ | PGAIN. CALO | ; YES |
| A8BE:18 |  |  | 508 |  | CLC |  | ;NO, DECREASE A/D RANGE |
| A8BF: 69 | 10 |  | 509 |  | ADC | \#\%00010000 |  |
| A8C1:8D | 13 | A5 | 510 |  | STA | AD.CHO.CTR |  |
| A8C4:8D | D3 | C0 | 511 |  | STA | AD.CTRL |  |
| A8C7:A9 | 31 |  | 512 |  | IDA | \#\%00110001 | ;PLACE S/H IN "TRACK" |
| MODE |  |  |  |  |  |  |  |
| A8C9:8D | 81 | C2 | 513 |  | STA | U2.DRA |  |
| A8CC:AD | OA | A5 | 514 |  | IDA | INT. NUM | ; RESET MULTIPLE |
| A8CF: 8D | OB | A5 | 515 |  | STA | INT.CNT | ; INTEGRATION COUNT |
| A8D2:A9 | 00 |  | 516 |  | IDA | \#00 | ;RESET SAMPLE AVAILABLE |
| A8D4:8D | 35 | A5 | 517 |  | STA | SMPL.AVAIL | ; FLAG |
| A8D7:4C | 9F | A8 | 518 |  | JMP | WAIT1.CAIO |  |
| A8DA:AD | 01 | C2 | 519 | PGAIN.FIXO | IDA | U1.DRA | ; SAVE CHO PROGRAMMABLE |
| A8DD:29 | OF |  | 520 |  | AND | \#\%00001111 | ; GAIN SEITING |
| A8DF: 8D | OF | A5 | 521 |  | STA | CHO.PGAIN |  |
| A8E2:4C | 2A | A9 | 522 |  | JMP | FINAL. CALO | ;FINAL A/D CAIIBRATION |
| A8E5:AD | 01 | C2 | 523 | PGAIN. CALO | LDA | U1.DRA |  |
| A8E8:C9 | 81 |  | 524 |  | CMP | \#\%10000001 | ; CHECK IF AT MAX. GAIN |
| A8EA:F0 | 28 | A914 | 525 |  | BEQ | EXIT.CALO | ; YES |
| A8EC:A9 | 31 |  | 526 |  | IDA | \#\%00110001 | ;PLACE S/H IN "TRACK" |
| MODE |  |  |  |  |  |  |  |
| A8EE:8D | 81 | C2 | 527 |  | STA | U2.DRA |  |
| A8F1: AD | 01 | C2 | 528 |  | LDA | U1.DRA | ; INCREASE PROGRAMMABIE |
| GAIN |  |  |  |  |  |  |  |
| A8F4:29 | OF |  | 529 |  | AND | \#\%00001111 | ; SINCE OVERFLOW NOT YET |
| A8F6: 4A |  |  | 530 |  | LSR |  | ;DETECTED |
| A8F7:18 |  |  | 531 |  | CLC |  |  |
| A8F8:69 | 80 |  | 532 |  | ADC | \#\%10000000 | ;RESTORE CHI MINIMUM GAIN |
| A8FA: 8D | 01 | C2 | 533 |  | STA | U1.DRA |  |
| A8FD:20 | 60 | AE | 534 |  | JSR | POST.MEAS | ; POST-MEASUREMENT |
| A900:20 | 58 | B4 | 535 |  | JSR | OUT. SETTLE | ;OUTPUTS MUST SETIIE |
| A903:20 | 2D | AE | 536 |  | JSR | PRE.MEAS | ; PRE-MEASUREMENT |
| A906:AD | OA | A5 | 537 |  | LDA | INT.NUM | ;RESET MULTIPLE |
| A909: 8D | OB | A5 | 538 |  | STA | INT. CNT | ; INTEGRATION COUNT |
| A90C:A9 | 00 |  | 539 |  | IDA | \#00 | ;RESET SAMPLE AVAILABLE |
| A90E: 8D | 35 | A5 | 540 |  | STA | SMPL. AVAII | ;FLAG |
| A911:4C | 92 | A8 | 541 |  | JMP | RANGE. CALO |  |
| A914:AD | 01 | C2 | 542 | EXIT.CALO | IDA | U1.DRA | ; SAVE PROGRAMMABLE GAIN |
| A917:29 | OF |  | 543 |  | AND | \#\%00001111 | ;FOR INSEECTION WITHIN |
| A919:8D | OF | A5 | 544 |  | STA | CHO.PGAIN | ; BASIC ROUTINE |
| A91C:AD | 13 | A5 | 545 |  | LDA | AD.CHO.CTRL |  |


| A91F: 4A |  | 546 |  | LSR |  | ; SAVE A/D RANGE |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| A920: 4A |  | 547 |  | LSR |  | ;FOR INSPECTION WITAIN |
| A921: 4A |  | 548 |  | LSR |  | ;BASIC ROUTINE |
| A922:4A |  | 549 |  | LSR |  |  |
| A923: 8D | 10 A5 | 550 |  | STA | CEO.AD.GAIN | ; SAVE A/D RANGE |
| A926:20 | 60 AE | 551 |  | JSR | POST.MEAS | ; POST-MEASUREMENT |
| A929:60 |  | 552 |  | RTS |  |  |
| A92A:AD | 13 A5 | 553 | FINAL. CALO | 0 IDA | AD.CHO.CTRL | ; INCREASE A/D RANGE TO |
| A92D: 38 |  | 554 |  | SEC |  | ; NEXT HIGHEST RANGE |
| A92E:E9 | 10 | 555 |  | SBC | \#\%00010000 | ;SINCE OVERFLOW DETECTIED |
| A930:8D | 13 A5 | 556 |  | STA | AD. CHO.CTRL |  |
| A933: 4A |  | 557 |  | ISR |  | ;SAVE A/D RANGE |
| A934:4A |  | 558 |  | ISR |  | ;FOR INSPECTION WITHIN |
| A935: 4A |  | 559 |  | LSR |  | ;BASIC ROUTINE |
| A936:4A |  | 560 |  | LSR |  |  |
| A937:8D | 10 A5 | 561 |  | STA | CHO.AD.GAIN |  |
| A93A:20 | 60 AE | 562 |  | JSR | POST.MEAS | ;POST-MEASUREMENT |
| A93D:60 |  | 563 |  | RTS |  |  |
| A93E: |  | 564 | ; |  |  |  |
| A93E: |  | 565 | ; **** $\quad$ A | AUTOMAT | IC GAIN CALIB | RATION CH1 ROUTINE **** |
| A93E: |  | 566 | ; |  |  |  |
| A93E:AD | OA A5 | 567 | AUTO.CAL1 | IDA | INT. NUM | ;MULT. INTEGRRATION COUNT |
| A941:8D | OB A5 | 568 |  | STA | INT.CNT |  |
| A944:A9 | 00 | 569 |  | LDA | \#00 | ;S/H DATA AVAILABLE |
| A946:8D | 35 A5 | 570 |  | STA | SMPL.AVAIL | ;FLAG INITIALLY ZERO |
| A949:A9 | 88 | 571 |  | LDA | \#\%10001000 | ;CH1 INITIAL GAIN $=2$ |
| A94B:8D | 01 C 2 | 572 |  | STA | U1.DRA |  |
| A94E: 20 | 58 B4 | 573 |  | JSR | OUT.SETTLE | :OUTPUT MUST SETTILE |
| A951:20 | 2D AE | 574 |  | JSR | PRE.MEAS | ; PRE-MEASUREMENT |
| A954:A9 | 11 | 575 | RANGE.CALI | 1 LDA | \#\%00010001 | ;CHI INIT A/D RANGE $=5 \mathrm{~V}$ |
| A956:8D | 14 A5 | 576 |  | STA | AD. CHI.CTRI |  |
| A959:8D | D3 C0 | 577 |  | STA | AD.CTRI | ;A/D CONTROL LOCATION |
| A95C:A9 | 31 | 578 |  | LDA | \#\%00110001 | ;PLACE S/H IN "TRACK" |
| MODE |  |  |  |  |  |  |
| A95E: 8D | 81 C2 | 579 |  | STA | U2.DRA |  |
| A961: AD | 35 A5 | 580 | WAIT1.CALI | 1 LDA | SMPL.AVAII | ;WAIT FOR SAMPLE |
| AVAIIABLE |  |  |  |  |  |  |
| A964:F0 | FB A961 | 581 |  | BEQ | WAIT1. CALI |  |
| A966:20 | 6F B4 | 582 |  | JSR | CONVERT | ; PERFORM A/D CONVERSION |
| A969:AD | D1 C0 | 583 |  | IDA | AD. HIGH |  |
| A96C:29 | OF | 584 |  | AND | \#\%00001111 | ; KEEP ONLY DATA BITS |
| A96E:C9 | OE | 585 |  | CMP | \#\%00001111 | ;CHECK FOR A/D "OVERELOW" |
| A970:90 | 07 A979 | 586 |  | BCC | CONT.CAL1 | ; NO |
| A972: AD | DO CO | 587 |  | LDA | AD.LOW |  |
| A975:C9 | E0 | 588 |  | CMP | \#\%11100000 |  |
| A977: B0 | 23 A99C | 589 |  | BCS | PGAIN.FIXI | ;YES |
| A979: AD | 14 A5 | 590 | CONT. CALI | LDA | AD.CHI.CTRU |  |
| A97C:C9 | 71 | 591 |  | CMP | \#\%01110001 | ; CHECK IF AT MIN. RANGE |
| A97E:F0 | 27 A9A7 | 592 |  | BEQ | PGAIN.CALI | ;YES |
| A980:18 |  | 593 |  | CLC |  | ; NO, DECREASE A/D RANGE |
| A981:69 | 10 | 594 |  | ADC | \#\%00010000 |  |
| A983:8D | 14 A5 | 595 |  | STA | AD.CH1.CTRL |  |
| A986:8D | D3 C0 | 596 |  | STA | AD. CTRL |  |
| A989:A9 | 31 | 597 |  | LDA | \#\%00110001 | ;PLACE S/H IN "TRACK" |
| MODE |  |  |  |  |  |  |
| A98B: 8D | 81 C2 | 598 |  | STA | U2.DRA |  |
| A98E:AD | OA A5 | 599 |  | LDA | INT. NUM | ;RESET MULTIPLE |
| A991:8D | OB A5 | 600 |  | STA | INT.CNT | ;INTEGRATION COUNT |







| ACOE:20 | AB AE | 70 | JSR | UF.CHK.SS | :CEECK FOR UNDERFIOW |
| :---: | :---: | :---: | :---: | :---: | :---: |
| AC11:AD | 1A A5 | 71 | LDA | OFFSET.ADJ |  |
| AC14:D0 | 03 AC19 | 72 | BNE | SKIP.Z.CE1 | ;SKIP FOR OFFSET ROUTINE |
| AC16:20 | ED AE | 73 | JSR | ZERO.SS | ;CHECK FOR "ZERO" DATA |
| AC19:C8 |  | 74 SKIP.Z.CHI | INY |  | ;UPDATE ADDRESS INDEX |
| AC1A:C8 |  | 75 | INY |  |  |
| AC1B:AD | 1A A5 | 76 | IDA | OFESET.ADJ |  |
| ACIE:D0 | 03 AC23 | 77 | BNE | SKIP. RATIO | ;SKIP FOR OFFSET ROUTINE |
| AC20:20 | 33 в | 78 | JSR | RATIO.SS | ;COMPUTE RATIO |
| AC23:C8 |  | 79 SKIP.RAIIO | INY |  |  |
| AC24:C8 |  | 80 | INY |  | ;UPDATE ADDRESS INDEX |
| AC25:C8 |  | 81 | INY |  |  |
| AC26:C8 |  | 82 | INY |  |  |
| AC27:EE | 34 A5 | 83 | INC | SWEEP.CNT | ;SWEEP COMPLETED |
| AC2A:AD | 34 A 5 | 84 | IDA | SWEEP.CNT |  |
| AC2D:C9 | OB | 85 | CMP | *11 | ; CHECK FOR ALL 10 SWEEPS |
| AC2F:F0 | 19 AC4A | 86 | BEQ | FINISH.SS | ; YES |
| AC31:A9 | 31 | 87 | IDA | *\%00110001 | ;PLACE S/H IN "TRACK" |
| MODE |  |  |  |  |  |
| AC33:8D | 81 C2 | 88 | STA | U2.DRA |  |
| AC36:AD | 13 A5 | 89 | IDA | AD. CHO.CTRU | ; SET UP A/D FOR CHO |
| AC39:8D | D3 C0 | 90 | STA | AD. CTRL |  |
| AC3C:AD | 0A A5 | 91 | IDA | INT. NUM | ; RESET MULTIPLE |
| AC3F:8D | OB A5 | 92 | STA | INT. CNT | ; INTEGRATION COUNT |
| AC42:A9 | 00 | 93 | LDA | \# 00 | ; RESET SAMPLE AVAIIABLE |
| AC44:8D | 35 A5 | 94 | STA | SMPL. AVAIL | ; FLAG |
| AC47:4C | A5 AB | 95 | JMP | WAIT.SS | ; CONTINUE MEASUREMENTS |
| AC4A:20 | 60 AE | 96 FINISH.SS | JSR | POST.MEAS | ;POST-MEASUREMENT |
| AC4D: 20 | 97 B1 | 97 | JSR | SWEEP .AVG | ; AVERAGE SWEEP VALUES |
| AC50:A9 | 31 | 98 | LDA | \#\% ${ }^{\text {\% }} 000110001$ | ;PLACE S/K IN "TRACK" |
| MODE |  |  |  |  |  |
| AC52:8D | 81 C2 | 99 | STA | U2.DRA |  |
| AC55:A9 | 00 | 100 | LDA | \#8800000000 | ;TRIGGER LEVEL LOW |
| AC57: 8D | 80 C 2 | 101 | STA | U2.DR8 |  |
| AC5A: 60 |  | 102 | RTS |  | ; RETURN FROM BASIC CAIL |
| AC5B: |  | 103 |  |  |  |
| AC5B:20 | EE AD | 104 M. SWEEP | JSR | INIT. SWEEP | ; INITIAIIZE ERROR FIAGS |
| AC5E:20 | OF BO | 105 | JSR | INIT.SUMS | ;INITIALIZE SUM LOCATIONS |
| AC61:20 | 2D AE | 106 | JSR | PRE.MEAS | ; PRE-MEASUREMENT |
| AC64:A9 | 01 | 107 | LDA | \#01 | ;FIRST SWEEP |
| AC66:8D | 34 A5 | 108 | STA | SWEEP. CNT |  |
| AC69:AD | OC A5 | 109 | LDA | SAMPLE.NUM | ; SAMPLE COUNT PER SWEEP |
| AC6C: 8D | OD A5 | 110 | STA | SAMPLE.CNT |  |
| AC6F:A2 | 00 | 111 | LDX | \#\$00 | ; ADDRESS INDEX |
| AC71:A0 | 00 | 112 | LDY | \#\$00 | ;ADDRESS INDEX |
| AC73:A9 | 31 | 113 | LDA | \#\%00110001 | ;PLACE S/H IN "TRACK" |
| MODE |  |  |  |  |  |
| AC75: 8D | 81 C2 | 114 | STA | U2.DRA |  |
| AC78:AD | 13 A5 | 115 | LDA | AD. CHO.CTRL | ; SET UR A/D FOR CHO |
| AC7B: 8D | D3 C0 | 116 | STA | AD.CTRL |  |
| AC7E:AD | OA A5 | 117 | LDA | INT. NUM | ;MULT. INTEGRATION COUNT |
| AC81:8D | OB A5 | 118 | STA | INT.CNT |  |
| AC84:A9 | 00 | 119 | LDA | \#00 | ; S/H DATA AVAILABLE |
| AC86:8D | 35 A5 | 120 | STA | SMPL.AVAIL | ;FLAG INITIALLY ZERO |
| AC89:AD | 35 A5 | 121 WAIT.MS | LDA | SMPL.AVAIL | ;WAIT FOR SAMPLE |
| AVAILABLE |  |  |  |  |  |
| AC8C: F0 | FB AC89 | 122 | BEQ | WAIT.MS |  |
| AC8E:20 | 6F B4 | 123 | JSR | CONVERT | ;RERFORM A/D CONVERSION |


| AC91:AD | D1 CO | 124 | LDA | AD. AIGE | ; RETRIEVE H BYTE RAW DATA |
| :---: | :---: | :---: | :---: | :---: | :---: |
| AC94:29 | OF | 125 | AND | *\%00001111 | ; KEEP ONLY DATA BITS |
| AC96:8D | 3F A5 | 126 | STA | MEAS.TEMP. H |  |
| AC99: AD | DO CO | 127 | LDA | AD.ION | ;RETRIEVE L BYTE RAW DATA |
| AC9C: 8D | 40 A5 | 128 | STA | MEAS.TEMP.L |  |
| AC9F:20 | 83 AE | 129 | JSR | OF.CEK.CHO | ; CHECK FOR A/D OVERELOW |
| ACA2:AD | 40 A5 | 130 | LDA | MEAS.TEMP.L |  |
| ACA5:38 |  | 131 | SEC |  | ; SUBTRACT LOW ORDER BYTE |
| ACA6: ED | 16 A5 | 132 | SBC | CHO.INJ.L | ; INJECTION CURRENT |
| ACA9:99 | 01 B9 | 133 | STA | SCRATCR0+1, 1 | ; $L$ BYTE CHO DATA |
| ACAC:AD | 3F A5 | 134 | LDA | MEAS.TEMP. H $^{\text {P }}$ | ;SUBTRACT HIGH ORDER BYTE |
| ACAF:ED | 15 A5 | 135 | SBC | CEO.INJ.H | ; INJECTION CURRENT |
| ACB2:99 | 00 B9 | 136 | STA | SCRATCHO,Y | ; B BYTE CHO DATA |
| ACB5:20 | C1 AE | 137 | JSR | UF.CHK.CHO | ; CHECK FOR UNDERFLOW |
| ACB8:20 | 03 AF | 138 | JSR | 2ERO.CHO | ;CEECK FOR "ZERO" DATA |
| ACBB:20 | $C B A A$ | 139 | JSR | SUM.CHO | ;ADD TO PREVIOUS CHO DATA |
| ACBE:AD | 14 A5 | 140 | LDA | AD.CHI.CTRL | ; SET UP A/D FOR CH1 |
| ACC1:8D | D3 C0 | 141 | STA | AD.CTRL |  |
| ACC4:20 | 66 B4 | 142 | JSR | MUX.SETTLE | ;A/D MUX MUST SETTIE |
| ACC7:20 | 6F B4 | 143 | JSR | CONVERT | ; PERFORM A/D CONVERSION |
| ACCA:AD | D1 C0 | 144 | IDA | AD. HIGH | ; RETRIEVE H BYTE RAW DATA |
| ACCD:29 | OF | 145 | AND | \#\%00001111 | ;KEEP ONLY DATA BITS |
| ACCF: 8 D | 3F A5 | 146 | STA | MEAS.TEMP. H |  |
| ACD2:AD | DO C0 | 147 | IDA | AD.LOW | ;RETRIEVE L BYTE RAW DATA |
| ACD5:8D | 40 A5 | 148 | STA | MEAS.TEMP.工 |  |
| ACD8:20 | 97 AE | 149 | JSR | OF.CHK.CH1 | :CHECK FOR A/D OVERFLOW |
| ACDB:AD | 40 A5 | 150 | LDA | MEAS.TEMP.L |  |
| ACDE: 38 |  | 151 | SEC |  | ;SUBTRACT LOW ORDER BYTE |
| ACDF:ED | 18 A5 | 152 | SBC | CH1.INJ.L | ; INJECTION CURRENT |
| ACE2:99 | 01 BB | 153 | STA | SCRATCH1+1, Y | ; 5 BYTE CH1 DATA |
| ACE5:AD | 3F A5 | 154 | LDA | MEAS.TEMP. H | : SUBTRACT HIGH ORDER BYTE |
| ACE8:ED | 17 A5 | 155 | SBC | CHI.INJ. ${ }^{\text {d }}$ | :INJECTION CURRENT |
| ACEB:99 | 00 BB | 156 | STA | SCRATCH1, Y | ; H BYTE CH1 DATA |
| ACEE:20 | D7 AE | 157 | JSR | UF.CHK.CH1 | ; CHECK FOR UNDERFLOW |
| ACF1:20 | 19 AF | 158 | JSR | 2ERO.CH1 | ; CHECK FOR "ZERO" DATA |
| ACF4:20 | E7 AA | 159 | JSR | SUM. CH1 | :ADD TO PREVIOUS CH1 DATA |
| ACE7:20 | 77 BO | 160 | JSR | RATIO.MS | ; COMPUTE RATIO |
| ACFA: 20 | BB BO | 161 | JSR | SUM. RATIO | ;ADD TO PREVIOUS RATIOS |
| ACFD: $\mathrm{C8}$ |  | 162 | INY |  | ;UPDATE ADDRESS INDEX |
| ACFE:C8 |  | 163 | INY |  |  |
| ACFF:AD | 13 A5 | 164 | LDA | AD. CHO.CTRI | ; SET UP FOR A/D CHO.CTRL |
| AD02:8D | D3 C0 | 165 | STA | AD.CTRL |  |
| AD05:CE | OD A5 | 166 | DEC | SAMPLE.CNT | ; DECREMENT SAMPLE COUNT |
| AD08:F0 | 13 AD1D | 167 | BEQ | COMP .MS | ; CHECK FOR END OF SWEEP |
| AD0A:A9 | 31 | 168 | LDA | \#800110001 | ;NO, PLACE S/H IN "TRACK" |
| ADOC: 8D | 81 C 2 | 169 | STA | U2.DRA | ;MODE |
| ADOF:AD | OA A5 | 170 | LDA | INI. NUM | ; RESET MULTIPLE |
| AD12:8D | OB A5 | 171 | STA | INT. CNT | ; INTEGRATION COUNT |
| AD15:A9 | 00 | 172 | LDA | \#00 | ;RESET SAMPLE AVAILABLE |
| AD17:8D | 35 A5 | 173 | STA | SMPL. AVAIL | ; FLAG |
| AD1A: 4 C | 89 AC | 174 | JMP | WAIT.MS | ;CONTINUE SAME SWEEP |
| AD1D:A9 | 00 | 175 COMP.MS | LDA | \#00 | ; DISABLE ACTIVE |
| AD1F:8D | 39 A5 | 176 | STA | DATA.ACTIVE | ; DATA MEASUREMENTS |
| AD22:20 | E9 B0 | 177 | JSR | AVG.CHO | ; COMPUTE CHO DATA AVG |
| AD25: E8 |  | 178 | INX |  | ; UPDATE ADDRESS INDEX |
| AD26:E8 |  | 179 | INX |  |  |
| AD27:20 | 19 Bl | 180 | JSR | AVG.CH1 | :COMPUTE CHI DATA AVG |
| AD2A: E8 |  | 181 | INX |  | : UPDATE ADDRESS INDEX |


| AD2B: E8 |  | 182 |  | INX |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| AD2C:20 | 49 B1 | 183 |  | JSR | AVG. RATIO | :COMPUIE SWEEP AVG RATIO |
| AD2F:E8 |  | 184 |  | InX |  |  |
| AD30:E8 |  | 185 |  | INX |  | ; UPDATE ADDRESS INDEX |
| AD31:E8 |  | 186 |  | INX |  |  |
| AD32:E8 |  | 187 |  | INX |  |  |
| AD33:EE | 34 A5 | 188 |  | INC | SWEEP.CNT | ;SWEEP COMPLETED |
| AD36:AD | 34 A5 | 189 |  | LDA | SWEEP. CNT |  |
| AD39:C9 | OB | 190 |  | CNP | \#11 | :CHECK FOR ALC 10 SNEEPS |
| AD3B:F0 | 23 AD60 | 191 |  | BEQ | FINISE.MS | ; YES |
| AD3D:20 | OF BO | 192 |  | JSR | INIT.SUMS | - INITIALIZE SUM LOCATIONS |
| AD40:AD | OC A5 | 193 |  | LDA | SAMPLE.NUM | ;RESET SAMPIE COUNT |
| AD43: 8D | OD A5 | 194 |  | STA | SAMPLE.CNT |  |
| AD46:A0 | 00 | 195 |  | IDY | *\$00 | ;RESET ADDRESS INDEX |
| AD48:A9 | 31 | 196 |  | IDA | \#\%00110001 | ;PLACE S/H IN "TRACK" |
| MODE |  |  |  |  |  |  |
| AD4A: 8D | 81 C2 | 197 |  | STA | U2.DRA |  |
| AD4D:AD | OA A5 | 198 |  | IDA | INT. NUM | ; RESET MULTIPLE |
| AD50:8D | OB A5 | 199 |  | STA | INT.CNT | ; INIEGRATIION COUNT |
| AD53:A9 | 00 | 200 |  | LDA | \#00 | ;RESET SAMPLE AVAILABLE |
| AD55:8D | 35 A5 | 201 |  | STA | SMPL.AVAIL | ; FLAG |
| AD58:A9 | 01 | 202 |  | IDA | \#01 | ; ENABLE ACTIVE |
| AD5A:8D | 39 A. 5 | 203 |  | STA | DATA.ACTIVE | ; DATA MEASUREMENTS |
| AD5D:4C | 89 AC | 204 |  | JMP | WAIT.MS | ; CONTINUE MEASUREMENTS |
| AD60:20 | 60 AE | 205 | FINISH.MS | JSR | POST.MEAS | ; POST-MEASUREMENT |
| AD63:20 | 97 Bl | 206 |  | JSR | SWEEP.AVG | ;AVERAGE SNEEP VALUES |
| AD66:A9 | 31 | 207 |  | LDA | \#\%00110001 | ;PLACE S/H IN "TRACK" |
| MODE |  |  |  |  |  |  |
| AD68:8D | 81 C 2 | 208 |  | STA | U2.DRA |  |
| AD6B:A9 | 00 | 209 |  | LDA | \#\%00000000 | ;TRIGGER LEVEL LOW |
| AD6D:8D | 80 C 2 | 210 |  | STA | U2.DRB |  |
| AD70:60 |  | 211 |  | RTS |  | ;RETURN FROM BASIC CAL工 |
| AD71: |  | 212 | ; |  |  |  |
| AD71:20 | 2D AE | 213 | DATA.FILE | JSR | PRE.MEAS | ; PRE-MEASUREMENT |
| AD74:A9 | C8 | 214 |  | LDA | ${ }_{*}^{*} 200$ | ;TOTAL SAMPLE COUNT |
| AD76:8D | OD A5 | - 215 |  | STA | SAMPLE.CNT |  |
| AD79:A0 | 00 | 216 |  | LDY | \#\$00 | ;ADDRESS INDEX |
| AD7B:A9 | 31 | 217 |  | LDA | \#\%00110001 | ;PLACE S/H IN "TRACK" |
| MODE |  |  |  |  |  |  |
| AD7D: 8D | 81 C2 | 218 |  | STA | U2.DRA |  |
| AD80:AD | 13 A5 | 219 |  | LDA | AD. CHO.CTRL | ; SET UP A/D FOR CHO |
| AD83:8D | D3 C0 | 220 |  | STA | AD. CTRL |  |
| AD86:AD | OA A5 | 221 |  | LDA | INT. NUM | ;MULT. INTEGRATION COUNT |
| AD89: 8D | OB A5 | 222 |  | STA | INT.CNT |  |
| AD8C:A9 | 00 | 223 |  | LDA | \#00 | ; S/h DATA AVAIIABLE |
| AD8E:8D | 35 A5 | 224 |  | STA | SMPL.AVAIL | ; FLAG INTTIALLY ZERO |
| AD91:AD | 35 A5 | 225 | WAIT.DF | LDA | SMPL.AVAIL | ;WAIT FOR SAMPLE |
| AVAILABLE |  |  |  |  |  |  |
| AD94:F0 | FB AD91 | 226 |  | BEQ | WAIT.DF |  |
| AD96:20 | 6F B4 | 227 |  | JSR | CONVERT | ; PERFORM A/D CONVERSION |
| AD99:AD | D1 C0 | 228 |  | LDA | AD. HIGE |  |
| AD9C:29 | OF | 229 |  | AND | \#\%00001111 | ; KEEP ONLY DATA BITS |
| AD9E:99 | 00 B9 | 230 |  | STA | SCRATCHO,Y | ; HIGH BYTE CHO |
| ADA1:AD | DO C0 | 231 |  | IDA | AD.LOW |  |
| ADA4:99 | 00 BA | 232 |  | STA | SCRATCHO.A, Y | ;LOW BYTE CHO |
| ADA7:AD | 14 A5 | 233 |  | IDA | AD.CH1.CTR | ; SET UR A/D FOR CH1 |
| ADAA:8D | D3 C0 | 234 |  | STA | AD.CTRL |  |
| ADAD:20 | 66 B4 | 235 |  | JSR | MUX.SETTLE | ;A/D MUX MUST SETTLE |




| AE97: |  | 347 | ; |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| AE97:AD | 3F A5 | 348 | OF.CHK.CAI | LDA | MEAS.TEMP.H |  |  |
| AE9A:C9 | OF | 349 |  | CMP | *SOE | ; CHECK FOR MAX VALUE |  |
| AE9C:F0 | 01 AE9F | 350 |  | BEQ | CONT. CEKK. 1 |  |  |
| AE9E: 60 |  | 351 |  | RTS |  | :NO |  |
| AE9F:AD | 40 A5 | 352 | CONT. CHK. 1 | IDA | MEAS.TEMP.L | ;YES |  |
| AEA2:C9 | FF | 353 |  | CMP | \# ${ }^{\text {PFF }}$ | ;CAECK FOR MAX VALUE |  |
| AEA4:E0 | 01 AEA7 | 354 |  | BEQ | OVER.CHI |  |  |
| AEA6: 60 |  | 355 |  | RTS |  | ;NO |  |
| AEA7:20 | 2 FaF | 356 | OVER.CHI | JSR | OVERELOW | ;YES, SET ERROR FLAG |  |
| AEAA: 60 |  | 357 |  | RTS |  |  |  |
| AEAB: |  | 358 | ; |  |  |  | ***** |
| AEAB: |  | 359 | ; ***** | SS DATA | ATA UNDERFLOW | CHECRING ROUTINE |  |
| AEAB: |  | 360 | ; |  |  |  |  |
| AEAB: $\mathrm{B9}^{\text {9 }}$ | $00 \mathrm{B8}$ | 361 | UF.CHK.SS | LDA | SCRATCH, Y | :IOAD HIGE BYTE DATA |  |
| AEAE:C9 | 10 | 362 |  | CMP | \# ${ }^{\text {P10 }}$ |  |  |
| AEB0:B0 | AEB3 | 363 |  | BCS | UNDER.SS | ;CHECK FOR UNDERFLOW; NO |  |
| AEB2:60 |  | 364 |  | RTS |  |  |  |  |
| AEB3:A9 |  | 365 | UNDER.SS | LDA | *\$00 | ;YES, SUBSTITUTE |  |
| UNDERFIOW |  |  |  |  |  |  |  |
| AEB5:99 | 00 B8 | 366 |  | STA | SCRATCH, Y | : DATA WIth a value of 1 |  |
| AEB8: A9 | 01 | 367 |  | IDA | \#\$01 |  |  |  |
| AEBA: 99 | 01 B8 | 368 |  | STA | SCRATCH+1, Y |  |  |
| AEBD: 20 | $9 F A F$ | 369 |  | JSR | UNDERFLOW | ;SET ERROR FLAG |  |
| AEC0: 60 |  | 370 |  | RIS |  |  |  |  |
| AECI: |  | 371 | , |  |  | W Ceiecking Routine | **** |
| AEC1: |  | 372 | ;**** MS | CHO | DATA UNDERELOW |  |  |
| AEC1: |  | 373 | ; |  |  |  |  |
| AEC1: $\mathrm{B9}^{\text {9 }}$ | $00 \mathrm{B9}$ | 374 | UE.CHK.CHO | IDA | SCRATCHO, Y | :LOAD HIGH BYTE DATA |  |
| AEC4:C9 | 10 | 375 |  | CMP | \#\$10 |  |  |  |
| AEC6: ${ }^{\text {O }}$ | AEC9 | 376 |  | BCS | UNDER.CHO | ; CHECK FOR UNDERFIOW |  |
| AEC8: 60 |  | 377 |  | RTS |  | ; NO |  |
| AEC9:A9 | 00 | 378 | UNDER.CHO | LDA | \#\$00 | ; YES, SUBSTITUTE |  |
| UNDERF'LOW |  |  |  |  |  |  |  |
| AECB: 99 | $00 \mathrm{B9}$ | 379 |  | STA | SCRATCHO, Y | ; DATA WITA A VALUE OF 1 |  |
| AECE: A9 | 01 | 380 |  | LDA | \#\$01 |  |  |  |
| AEDO: 99 | 01 B9 | 381 |  | STA | SCRATCHO+1, Y |  |  |
| AED3: 20 | 9F AF | 382 |  | JSR | UNDERFLOW | ; SET ERROR FIAG |  |
| AED6: 60 |  | 383 |  | RTS |  |  |  |
| AED7: |  | 384 |  |  |  | N CHECKING ROUTINE | **** |
| AED7: |  | 385 | ;**** MS | CHI | DATA UNDERFLOW |  |  |
| AED7: |  | 386 | ; |  |  |  |  |
| AED7: $\mathrm{B9}$ | BB | 387 | UF.CHK.CH1 | LDA | SCRATCH1, Y | :IOAD HIGH BYTE DATA |  |
| AEDA: C9 |  | 388 |  | CMP | \#\$10 |  |  |
| AEDC: $\mathrm{BO}^{0}$ | AEDF | 389 |  | BCS | UNDER.CH1 | ; CHECK FOR UNDERFLOW |  |
| AEDE: 60 |  | 390 |  | RTS |  | ; i YES, SUBSTITUTE |  |
| AEDF:A9 | 00 | 391 | UNDER.CHI | LDA | \#\$00 |  |  |  |
| UNDERFIOW |  |  |  |  |  |  |  |
| AEE1:99 | BB | 392 |  | STA | SCRATCH1, Y | : DATA WITH A VALUE OE 1 |  |
| AEE4:A9 | 01 | 393 |  | IDA | \#\$01 |  |  |
| AEE6:99 | 01 BB | 394 |  | STA | SCRATCH1+1, Y |  |  |
| AEE9:20 | 9F AF | 395 |  | JSR | UNDERFLOW | ;SET ERROR FLAG |  |
| AEEC: 60 |  | 396 |  | RTS |  |  |  |
| AEED: |  | 397 | ; |  |  |  |  |
| AEED: |  | 398 | ; ****** | SS ZERO CHECKING ROUTINE ***** |  |  |  |
| AEED: |  | 399 | ' ZERO .SS | LDACMP |  |  |  |
| AEED: $\mathrm{B9}$ | $00 \mathrm{B8}$ | 400 |  |  | SCRATCH, |  |  |
| AEE0:C9 00 | 00 | 401 |  |  | \#\$00 | ; CHECK FOR ZERO VALUE |  |


| AEF2:FO | 01 | AEF5 | 402 |  | BEQ | CONT.z.SS |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| AEF4:60 |  |  | 403 |  | RTS |  | :NO |
| AEF5: 89 |  | B8 | 404 | CONT.Z.SS | IDA | SCRATCE+1, Y | ; YES |
| AEF8:C9 | 00 |  | 405 |  | CMP | \#\$00 | ;CHECK FOR ZERO VALUE |
| AEFA:FO | 01 | AEFD | 406 |  | BEQ | SUBST.SS |  |
| AEFC: 60 |  |  | 407 |  | RTS |  | ; NO |
| AEFD:A9 | 01 |  | 408 | SUBST.SS | IDA | ${ }^{*} 01$ | ;YES SUBSTITUTE A |
| AEFF: 99 | 01 | B8 | 409 |  | STA | SCRATCE $+1, Y$ | ;VALUE OE ONE |
| AF02: 60 |  |  | 410 |  | RTS |  |  |
| AF03: |  |  | 411 | ; |  |  |  |
| AF03: |  |  | 412 | ; ****** | MS CHO | 0 ZERO CHECKING | G ROUTINE ***** |
| AF03: |  |  | 413 | ; |  |  |  |
| AF03:B9 |  | B9 | 414 | ZERO.CHO | LDA | SCRATCHO,Y |  |
| AF06:C9 | 00 |  | 415 |  | CMP | \#\$00 | ; CHECK FOR ZERO VALUE |
| AF08:F0 | 01 | AFOB | 416 |  | BEQ | CONT.Z.CHO |  |
| AFOA: 60 |  |  | 417 |  | RTS |  | ; NO |
| AF0B: 89 |  | B9 | 418 | CONT.Z.CEO | LDA | SCRATCHO+1, Y | ; YES |
| AFOE:C9 | 00 |  | 419 |  | CMP | *\$00 | :CHECK FOR ZERO VALUE |
| AF10:F0 | 01 | AF13 | 420 |  | BEQ | SUBST. CHO |  |
| AF12:60 |  |  | 421 |  | RTS |  | ; NO |
| AF13:A9 | 01 |  | 422 | SUBST.CHO | IDA | \#01 | ;YES SUBSTITUTE A |
| AF15:99 | 01 | B9 | 423 |  | STA | SCRATCRO+1, $Y$ | :VALUE OF ONE |
| AF18:60 |  |  | 424 |  | RTS |  |  |
| AF19: |  |  | 425 | ; |  |  |  |
| AF19: |  |  | 426 | ; ****** | MS CH1 | 1 ZERO CHECKING | G ROUTINE ***** |
| AF19: |  |  | 427 | ; |  |  |  |
| AF19:B9 | 00 | BB | 428 | ZERO.CH1 | IDA | SCRATCH1, Y |  |
| AF1C:C9 | 00 |  | 429 |  | CMP | \#\$00 | :CHECK FOR ZERO VALUE |
| AFIE:F0 | 01 | AF21 | 430 |  | BEQ | CONT.Z.CH1 |  |
| AF20:60 |  |  | 431 |  | RTS |  | ; NO |
| AF21:B9 |  | BB | 432 | CONT.Z.CH1 | IDA | SCRATCHI+1,Y; | ; YES |
| AF24:C9 | 00 |  | 433 |  | CMP | \#\$00 ; | :CHECK FOR ZERO VALUE |
| AF26:F0 | 01 | AF29 | 434 |  | BEQ | SUBST.CH1 |  |
| AF28:60 |  |  | 435 |  | RTS |  | ; NO |
| AF29:A9 | 01 |  | 436 | SUBST.CE1 | IDA | \#01 ; | :YES SUBSTITUTE A |
| AF2B:99 | 01 | BB | 437 |  | STA | SCRATCH1+1,Y ; | ; VALUE OF ONE |
| AF2E: 60 |  |  | 438 |  | RIS |  |  |
| AF2F: |  |  | 439 | ; |  |  |  |
| AF2F: |  |  | 440 | ;******* | OVER | RFLOW ERROR FLA | AG ROUTINE ****** |
| AF2F: |  |  | 441 | ; |  |  |  |
| AF2F:AD |  | A5 | 442 | OVERFLOW | IDA | SWEEP.CNT ; | ; DETERMINE SWEEP |
| AF32:C9 | 01 |  | 443 |  | CMP | \#01 ; | ; IN WHICH OVERFLOW |
| AF34:F0 | 2 D | AF63 | 444 |  | BEQ | OF.ERR.SW1 ; | ; ERROR OCCURED |
| AF36:C9 | 02 |  | 445 |  | CMP | \#02 |  |
| AF38:F0 | 2 F | AF69 | 446 |  | BEQ | OF.ERR.SW2 |  |
| AF3A:C9 | 03 |  | 447 |  | CMP | \#03 |  |
| AF3C:E0 | 31 | AF6F | 448 |  | BEQ | OF.ERR.SW3 |  |
| AF3E:C9 | 04 |  | 449 |  | CMP | 404 |  |
| AF40:F0 | 33 | AF75 | 450 |  | BEQ | OF.ERR.SW4 |  |
| AF42:C9 | 05 |  | 451 |  | CMP | \#05 |  |
| AF44:F0 | 35 | AF7B | 452 |  | BEQ | OF.ERR.SW5 |  |
| AF46:C9 | 06 |  | 453 |  | CMP | \#06 |  |
| AF48:F0 | 37 | AF81 | 454 |  | BEQ | OF.ERR.SW6 |  |
| AF4A:C9 | 07 |  | 455 |  | CMP | \#07 |  |
| AF4C:F0 | 39 | AF87 | 456 |  | BEQ | OF.ERR.SW7 |  |
| AF4E:C9 | 08 |  | 457 |  | CMP | \#08 |  |
| AF50:F0 | 3B | AF8D | 458 |  | BEQ | OF.ERR.SW8 |  |
| AF52:C9 | 09 |  | 459 |  | CMP | \#09 |  |


| AF54:F0 | 3D | AF93 | 460 |  | BEQ | OF.ERR.SW9 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| AF56:C9 | OA |  | 461 |  | CMP | ${ }^{*} 10$ |  |
| AF58:F0 | 3F | AF99 | 462 |  | BEQ | OF.ERR.SW10 |  |
| AF5A:A9 | 20 |  | 463 |  | LDA | \#\%00100000 | ; INVALID SWEEP NUMBER |
| AF5C:8D | 60 | AS | 464 |  | STA | INVALID | : SYSTEM ERROR BYE |
| AFSF:20 |  | B4 | 465 |  | JSR | ERROR | ;SYSTEM ERROR ROUTINE |
| AF62:60 |  |  | 466 |  | RTS |  |  |
| AF63:A9 | 01 |  | 467 | OF.ERR.SWI | LDA | *01 |  |
| AF65:8D | 20 | A5 | 468 |  | STA | OF.SWI |  |
| AF68:60 |  |  | 469 |  | RTS |  |  |
| AF69:A9 | 01 |  | 470 | OF.ERR.SW2 | LDA | \#01 | ; SET ERROR FLAG |
| AF6B:8D |  | A5 | 471 |  | STA | OF.SW2 | ;ASSOCIATED WITH SWEEP |
| AF6E: 60 |  |  | 472 |  | RTS |  | ; NUMBER |
| AF6F:A9 | 01 |  | 473 | OF.ERR.SW3 | LDA | \# 01 |  |
| AF71:8D |  | A. 5 | 474 |  | STA | OF.SW3 |  |
| AF74:60 |  |  | 475 |  | RTS |  |  |
| AF75:A9 | 01 |  | 476 | OF.ERR.SW4 | LDA | *01 |  |
| AF77:8D |  | A5 | 477 |  | STA | OF.SW4 |  |
| AF7A: 60 |  |  | 478 |  | RTS |  |  |
| AF7B:A9 | 01 |  | 479 | OF.ERR.SW5 | LDA | ${ }^{3} 01$ |  |
| AF7D:8D | 24 | A5 | 480 |  | STA | OF.SW5 |  |
| AF80:60 |  |  | 481 |  | ส̄กร |  |  |
| AF81:A9 | 01 |  | 482 | OF.ERR.SW6 | LDA | \#01 |  |
| AF83:8D |  | A5 | 483 |  | STA | OF.SW6 |  |
| AF86:60 |  |  | 484 |  | RTS |  |  |
| AF87:A9 | 01 |  | 485 | OF.ERR.SW7 | LDA | \#01 |  |
| AF89:8D |  | A5 | 486 |  | STA | OF.SW7 |  |
| AF8C: 60 |  |  | 487 |  | RTS |  |  |
| AF8D:A9 | 01 |  | 488 | OF.ERR.SW8 | IDA | \#01 |  |
| AF8F:8D |  | A5 | 489 |  | STA | OF.SW8 |  |
| AF92:60 |  |  | 490 |  | RTS |  |  |
| AF93:A9 | 01 |  | 491 | OF.ERR.SW9 | IDA | \#01 |  |
| AF95:8D |  | A5 | 492 |  | STA | OF.SW9 |  |
| AF98:60 |  |  | 493 |  | RTS |  |  |
| AF99:A9 | 01 |  | 494 | OF.ERR.SW10 | IDA | \#01 |  |
| AF9B:8D | 29 | A5 | 495 |  | STA | OF.SW10 |  |
| AF9E:60 |  |  | 496 |  | RTS |  |  |
| AF9F: |  |  | 497 | ; |  |  |  |
| AF9F: |  |  | 498 | ; ******* | UND | ERFLOW ERROR | FLAG ROUTINE ****** |
| AF9F: |  |  | 499 | ; |  |  |  |
| AF9F:AD | 34 | AS | 500 | UNDERELOW | IDA | SWEEP.CNT | ;DETERMINE SWEEP |
| AFA2:C9 | 01 |  | 501 |  | CMP | \#01 | ; IN WHICH UNDERFLOW |
| AFA4:F0 | 2D | AFD3 | 502 |  | BEQ | UF.ERR.SW1 | ;ERROR OCCURED |
| AFA6:C9 | 02 |  | 503 |  | CMP | \#02 |  |
| AFA8:F0 | 2F | AFD 3 | 504 |  | BEQ | UF.ERR.SW2 |  |
| AFAA:C9 | 03 |  | 505 |  | CMP | \#03 |  |
| AFAC:FO | 31 | AFDE | 506 |  | BEQ | UF.ERR.SW3 |  |
| AFAE:C9 | 04 |  | 507 |  | CMP | \#04 |  |
| AFB0:F0 | 33 | AFES | 508 |  | BEQ | UF.ERR.SW4 |  |
| AFB2:C9 | 05 |  | 509 |  | CMP | \#05 |  |
| AFB4:F0 | 35 | AFEB | 510 |  | BEQ | UF.ERR.SW5 |  |
| AFB6:C9 | 06 |  | 511 |  | CMP | \#06 |  |
| AFB8:F0 | 37 | AFF1 | 512 |  | BEQ | UF.ERR.SW6 |  |
| AFBA:C9 | 07 |  | 513 |  | CMP | \#07 |  |
| AFBC:F0 | 39 | AFF7 | 514 |  | BEQ | UF.ERR.SW7 |  |
| AFBE:C9 | 08 |  | 515 |  | CMP | \#08 |  |
| AFCO:F0 | 3B | AFFD | 516 |  | BEQ | UF.ERR.SW8 |  |
| AFC2:C9 | 09 |  | 517 |  | CMP | \#09 |  |


| AFC4:F0 | 3D B003 | 518 |  | BEQ | UE.ERR.SW9 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| AFC6:C9 | OA | 519 |  | CMP | +10 |  |
| AFC8:E0 | 3F B009 | 520 |  | BEQ | UF.ERR.SW10 | - |
| AFCA:A9 | 20 | 521 |  | IDA | *\%00100000 | ; INVALID SWEEE NUMBER |
| AFCC: 8D | 60 A5 | 522 |  | STA | INVAIID | : SYSTEM EREOR BYTE |
| AFCE: 20 | 7A B4 | 523 |  | JSR | ERROR | ;SYSTEM ERROR ROUTINE |
| AFD2: 60 |  | 524 |  | RTS |  |  |
| AFD3:A9 | 01 | 525 | UF.ERR.SWI | IDA | \#01 | ; SET ERROR FLAG |
| AFD5: 8D | 2A A5 | 526 |  | STA | UF.SW1 | ;ASSOCIATED WITH SWEEP |
| AFD8: 60 |  | 527 |  | RTS |  | ; NUMBER |
| AFD9:A9 | 01 | 528 | UF.ERR.SW2 | IDA | \% 01 |  |
| AFDB: 8D | 2B A5 | 529 |  | STA | UF.SW2 |  |
| AFDE: 60 |  | 530 |  | RTS |  |  |
| AFDF:A9 | 01 | 531 | UF.ERR.SW3 | IDA | \#01 |  |
| AFE1:8D | 2C A5 | 532 |  | STA | UF.SW3 |  |
| AFE4:60 |  | 533 |  | RIS |  |  |
| AFE5:A9 | 01 | 534 | UF.ERR.SW4 | LDA | \#01 |  |
| AFE7:8D | 2D A5 | 535 |  | STA | UF. SW4 |  |
| AFEA: 60 |  | 536 |  | RTS |  |  |
| AFEB:A9 | 01 | 537 | UF.ERR.SW5 | LDA | \#01 |  |
| AFED: 8 D | 2E A. 5 | 538 |  | STA | UF.SW5 |  |
| AFEO: 60 |  | 539 |  | RTS |  |  |
| AFF1:A9 | 01 | 540 | UF.ERR.SW6 | LDA | \#01 |  |
| AFF3:8D | 2F A5 | 541 |  | STA | UF.SW6 |  |
| AFF6:60 |  | 542 |  | RTS |  |  |
| AFF7:A9 | 01 | 543 | UF.ERR.SW7 | LDA | \#01 |  |
| AFF9:8D | 30 A5 | 544 |  | STA | UE. SW7 |  |
| AFFC: 60 |  | 545 |  | RTS |  |  |
| AFFD:A9 | 01 | 546 | UF.ERR.SW8 | LDA | \#01 |  |
| AFFF: 8D | 31 A5 | 547 |  | STA | UF.SW8 |  |
| B002:60 |  | 548 |  | RTS |  |  |
| B003:A9 | 01 | 549 | UF.ERR.SW9 | LDA | \#01 |  |
| B005:8D | 32 A5 | 550 |  | STA | UF.SW9 |  |
| B008: 60 |  | 551 |  | RTS |  |  |
| B009:A9 | 01 | 552 | UF.ERR.SW10 | 0 LDA | \#01 |  |
| B00B:8D | 33 A5 | 553 |  | STA | UF.SW10 |  |
| B00E: 60 |  | 554 |  | RIS |  |  |
| B00F: |  | 555 | ; |  |  |  |
| B00F: |  | 556 |  | CHN | JMC1. 2 | ; CHAIN IN NEXT SOURCE |
| FILE |  |  |  |  |  |  |
| B00F: |  | 1 |  |  |  |  |
| B00F: |  | 2 | ;********* |  | YSTEM SUM INIT | TIALIZATIONS ********* |
| B00F: |  | 3 |  |  |  |  |
| B00F:A9 | 00 | 4 | INIT.SUMS | IDA | \#\$00 | ; INITIALIZE ALC SUM |
| B011:8D | 55 A5 | 5 |  | STA | CHO.SUM. H | ; LOCATIONS TO ZERO |
| B014:8D | 56 A5 | 6 |  | STA | CHO.SUM.M | ; VALUES |
| B017:8D | 57 A5 | 7 |  | STA | CHO.SUM. I |  |
| B01A:8D | 58 A5 | 8 |  | STA | CHI.SUM. H |  |
| B01D:8D | 59 A5 | 9 |  | STA | CH1.SUM.M |  |
| B020:8D | 5A A5 | 10 |  | STA | CH1.SUM.L |  |
| B023:8D | 5B A5 | 11 |  | STA | RAT.SUM. H |  |
| B026:8D | 5C A5 | 12 |  | STA | RAT.SUM.MH |  |
| B029:8D | 5D A5 | 13 |  | STA | RAT.SUM.M |  |
| B02C:8D | 5E A5 | 14 |  | STA | RAT.SUM.ML |  |
| B02F:8D | 5F A5 | 15 |  | STA | RAT.SUM.L |  |
| B032:60 |  | 16 |  | RTS |  |  |
| B033: |  | 17 |  |  |  |  |
| B033: |  | 18 | ;***** | SINGLE | SWEEP RATIO | COMPUTATION ***** |


| B033: | 19 | ; |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| B033:B9 FC B7 | 20 | RATIO.SS | IDA | SCRATCH-4,Y | ;LOAD DIVIDEND LOCATIONS |
| B036:8D 47 A5 | 21 |  | STA | DIV.QUOT.ME | ;WITH CHO CORRECTED DATA |
| B039:B9 FD B7 | 22 |  | IDA | SCRATCH-3, Y |  |
| B03C:8D 48 A5 | 23 |  | STA | DIV.QUOT.M |  |
| B03F:A9 00 | 24 |  | IDA | \#\$00 |  |
| B041:8D 49 A5 | 25 |  | STA | DIV.QUOT.ML |  |
| B044:8D 4A A5 | 26 |  | STA | DIV.QUOT.L |  |
| B047:A9 00 | 27 |  | IDA | \#\$00 | ;LOAD DIVISOR LOCATIONS |
| B049:8D 4C A5 | 28 |  | STA | DIVISOR.MH | ;WITH CHI CORRECTED DATA |
| B04C:8D 4D A5 | 29 |  | STA | DIVISOR.M |  |
| B04F:B9 EE B7 | 30 |  | LDA | SCRATCH-2,Y |  |
| B052:8D 4E A5 | 31 |  | STA | DIVISOR.ML |  |
| B055:B9 FF B7 | 32 |  | IDA | SCRATCH-1, Y |  |
| B058:8D 4F A5 | 33 |  | STA | DIVISOR.L |  |
| B05B:20 30 в3 | 34 |  | JSR | DIVIDE. 32 |  |
| B05E:AD 47 A5 | 35 |  | LDA | DIV.QUOT.MH | :RETRIEVE QUOTIENT AS |
| B061:99 00 B8 | 36 |  | STA | SCRATCH, Y | ; INTEGER PART OF RATIO |
| B064:AD 48 A5 | 37 |  | LDA | DIV.QUOT.M |  |
| B067:99 01 B8 | 38 |  | STA | SCRATCH+1, Y |  |
| B06A:AD 49 A5 | 39 |  | IDA | DIV.QUOT.ML | ; RETRIEVE QUOTIENT AS |
| B06D:99 02 B8 | 40 |  | STA | SCRATCH+2,Y | ; FRACTIONAL PART OF RATIO |
| B070:AD 4A A5 | 41 |  | LDA | DIV.QUOT.L |  |
| B073:99 03 B8 | 42 |  | STA | SCRATCH $+3, Y$ |  |
| B076:60 | 43 |  | RTS |  |  |
| B077: | 44 | ; |  |  |  |
| B077: | 45 | ; ***** | MULTIP | PLE SWEEP RATIO | O COMPUTATION ***** |
| B077: | 46 |  |  |  |  |
| B077:B9 00 B9 | 47 | RATIO.MS | IDA | SCRATCHO, Y | ;IOAD DIVIDEND LOCATIONS |
| B07A:8D 47 A5 | 48 |  | STA | DIV. QUOT.ME | ; WITH CHO CORRECTED DATA |
| B07D:B9 01 B9 | 49 |  | IDA | SCRATCH0 $+1, Y$ |  |
| B080:8D 48 A5 | 50 |  | STA | DIV.QUOT.M |  |
| B083:A9 00 | 51 |  | IDA | *\$00 |  |
| B085:8D 49 A5 | 52 |  | STA | DIV.QUOT.ML |  |
| B088:8D 4A A5 | 53 |  | STA | DIV.QUOT.L |  |
| B08B:A9 00 | 54 |  | IDA | \#\$00 | ; IOAD DIVISOR LOCATIONS |
| B08D:8D 4C A5 | 55 |  | STA | DIVISOR.MH | ;WITH CHI CORRECTED DATA |
| B090:8D 4D A5 | 56 |  | STA | DIVISOR.M |  |
| B093:B9 00 BB | 57 |  | LDA | SCRATCH1, Y |  |
| B096:8D 4E A5 | 58 |  | STA | DIVISOR.ML |  |
| B099:B9 01 BB | 59 |  | LDA | SCRATCH1+1, $Y$ |  |
| B09C:8D 4F A5 | 60 |  | STA | DIVISOR.L |  |
| B09F:20 30 B3 | 61 |  | JSR | DIVIDE. 32 |  |
| B0A2:AD 47 A5 | 62 |  | LDA | DIV.QUOT.MH | ; RETRIEVE QUOTIENT AS |
| B0A5:99 00 BD | 63 |  | STA | RAT.INT, Y | ;INTEGER PART OF RATIO |
| B0A8:AD 48 A5 | 64 |  | LDA | DIV.QUOT.M |  |
| B0AB:99 01 BD | 65 |  | STA | RAT. ${ }^{\text {INT }}+1, Y$ |  |
| BOAE:AD 49 A5 | 66 |  | LDA | DIV.QUOT.ML | ; RETRIEVE QUOTIENT AS |
| BOBI:99 O0 BE | 67 |  | STA | RAT. FRAC, Y | ; FRACTIONAL PART OF RATIO |
| B0B4:AD 4A A5 | 68 |  | IDA | DIV.QUOT.L |  |
| B0B7:99 01 BE | 69 |  | STA | RAT.FRAC+1, Y |  |
| BOBA: 60 | 70 |  | RTS |  |  |
| B0BB: | 71 | ; |  |  |  |
| B0BB: | 72 | ; ********** |  | RATIO SUMMATION | N ROUTINE ******** |
| BOBB: | 73 | ; |  |  |  |
| B0BB:18 | 74 | SUM.RATIO | CLC |  | ;SUM TOGETHER WITH |
| BOBC:AD 5E A5 | 75 |  | LDA | RAT.SUM.L | ;PREVIOUS RATIOS |
| BOBF:79 01 BE | 76 |  | ADC | RAT.FRAC+1, Y | ; L BYTE FRACTIONAL PART |


| B0C2:8D | 5F A5 | 77 | STA | RAT.SUM.L |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| B0C5:AD | 5E A5 | 78 | LDA | RAT.SUM.ML |  |
| B0C8:79 | 00 BE | 79 | ADC | RAT.FRAC,Y - | ; H BYTE FRACTIONAL PART |
| B0CB:8D | 5E A5 | 80 | STA | RAT.SUM.ML |  |
| BOCE:AD | 5D A5 | 81 | IDA | RAT.SUM.M |  |
| B0D1:79 | 01 BD | 82 | ADC | RAT. $1 N T+1, Y$; | ;L BYTE INTEGER PART |
| B0D4:8D | 5D A5 | 83 | STA | RAT.SUM.M |  |
| B0D7:AD | 5C A5 | 84 | LDA | RAT.SUM.MH |  |
| B0DA: 79 | 00 BD | 85 | ADC | RAT.INT, Y ; | ; H BYTE INTEGER PART |
| B0DD: 8D | 5C A5 | 86 | STA | RAT.SUM.ME |  |
| BOEO:AD | 5B A5 | 87 | IDA | RAT.SUM.H |  |
| BOE3:69 | 00 | 88 | ADC | *00 ; | ;ALLOW FOR PROPAGATION |
| BOE5:8D | 5B A5 | 89 | STA | RAT.SUM.H |  |
| B0E8: 60 |  | 90 | RIS |  |  |
| B0E9: |  | 91 ; |  |  |  |
| B0E9: |  | 92 ; ${ }^{\text {at**** }}$ | CHANN | NEL 0 data swee | EP COMPUTATION ****** |
| B0E9: |  | 93 ; |  |  |  |
| B0E9:AD | 55 A5 | 94 AVG.CHO | LDA | CHO.SUM. H ; | :COMPUTE AVERAGE VALUE |
| BOEC: 8D | 46 A5 | 95 | STA | DIV.QUOT.H ; | ; FOR Channel 0 data |
| BOEF : AD | 56 A5 | 96 | LDA | CHO.SUM.M |  |
| B0F2:8D | 48 A5 | 97 | STA | DIV.QUOT.M ; | :LOAD DIVIDEND LOCATIONS |
| B0F5: AD | 57 A5 | 98 | LDA | CHO.SUM.L ; | ; WITH SUM OE CHO DATA |
| B0F8:8D | 4A A5 | 99 | STA | DIV.QUOT.I |  |
| B0FB:A9 | 00 | 100 | LDA | \#\$00 ; | ;IOAD DIVIDEND LOCATIONS |
| B0FD:8D | 4 B A5 | 101 | STA | DIVISOR.H ; | ;WITH TOTAL SAMPLE COUNT |
| B100:8D | 4D A5 | 102 | STA | DIVISOR.M |  |
| B103:AD | OC A5 | 103 | LDA | SAMPLE.NUM |  |
| B106:8D | 4F A5 | 104 | STA | DIVISOR.L |  |
| B109:20 | DA B2 | 105 | JSR | DIVIDE. 24 |  |
| B10C:AD | 48 A 5 | 106 | IDA | DIV.QUOT.M : | ;RETRIEVE QUOTIENT TERMS |
| B10F:9D | 00 B8 | 107 | STA | SCRATCH, X ; | CORRESPONDING TO AVG |
| B112:AD | 4A A5 | 108 | LDA | DIV.QUOT.L ; | -VALUE |
| B115:9D | 01 B8 | 109 | STA | SCRATCH +1 , X |  |
| B118:60 |  | 110 | RTS |  |  |
| B119: |  | 111 ; |  |  |  |
| B119: |  | 112 ;****** | CHANN | NEL 1 data SWEE | EP COMPUTATION ****** |
| B119: |  | 113 ; |  |  |  |
| B119:AD | 58 A5 | 114 AVG.CH1 | LDA | CH1.SUM.H ; | ; COMPUTE AVERAGE VALUE |
| B11C:8D | 46 A5 | 115 | STA | DIV.QUOT.H ; | ;FOR CHANNEL 1 DATA |
| B11F:AD | 59 A5 | 116 | LDA | CH1.SUM.M |  |
| B122:8D | 48 A5 | 117 | STA | DIV.QUOT.M : | [LOAD DIVIDEND LOCATIONS |
| B125:AD | 5A A5 | 118 | IDA | CHI.SUM.L ; | ;WITH SUM OF CHI DATA |
| B128:8D | 4A A5 | 119 | STA | DIV.QUOT.L |  |
| B12B:A9 | 00 | 120 | IDA | \#\$00 ; | [LOAD DIVISOR LOCATIONS |
| B12D: 8D | 4B A5 | 121 | STA | DIVISOR.H ; | WITH TOTAL SAMPLE COUNT |
| B130:8D | 4D A5 | 122 | STA | DIVISOR.M |  |
| B133:AD | OC A5 | 123 | LDA | SAMPLE.NUM |  |
| B136:8D | 4F A5 | 124 | STA | DIVISOR.L |  |
| B139:20 | DA B2 | 125 | JSR | DIVIDE. 24 |  |
| B13C:AD | 48 A5 | 126 | IDA | DIV.QUOT.M ; | -RETRIEVE QUOTIENT TERMS |
| B13F:9D | 00 B8 | 127 | STA | SCRATCH, X ; | CORRESPONDING TO AVG |
| B142:AD | 4A A5 | 128 | LDA | DIV.QUOT.L ; | VALUE |
| B145:9D | 01 B8 | 129 | STA | SCRATCH +1 , X |  |
| B148:60 |  | 130 | RTS |  |  |
| B149: |  | 131 ; |  |  |  |
| B149: |  | 132 ; ***** | AVERAG | EE RATIO SWEEP | COMPUTATION ***** |
| B149: |  | 133 ; |  |  |  |
| B149:AD | 58 A 5 | 134 AVG.RATIO | LDA | RAT.SUM. H ; | COMPUTE AVERAGE VALUE |


| B14C:8D | 46 A5 | 135 |  | STA | DIV.QUOT.E | ;FOR CHO/CH1 RATIO |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| B14F:AD | 5C A5 | 136 |  | LDA | RAT.SUM.MH |  |
| B152:8D | 47 A5 | 137 |  | STA | -DIV.QUOT.MH |  |
| B155:AD | 5D A5 | 138 |  | LDA | RAT.SUM.M | :LOAD DIVIDEND LOCATIONS |
| B158:8D | 48 A5 | 139 |  | STA | DIV.QUOT.M | ;WITH SUM OF RATIOS |
| B15B:AD | 5E A5 | 140 |  | LDA | RAT.SUM.ML |  |
| B15E:8D | 49 A5 | 141 |  | STA | DIV.QUOT.ML |  |
| B161:AD | 5F A5 | 142 |  | IDA | RAT.SUM.L |  |
| B164:8D | 4A A5 | 143 |  | STA | DIV.QUOT.L |  |
| B167:A9 | 00 | 144 |  | LDA | *\$00 | ;LOAD DIVISOR IOCATIONS |
| B169:8D | 4B A5 | 145 |  | STA | DIVISOR.H | ;WITH TOTAL SAMPLE COUNT |
| B16C:8D | 4C A5 | 146 |  | STA | DIVISOR.MH |  |
| B16F:8D | 4D A5 | 147 |  | STA | DIVISOR.M |  |
| B172:8D | 4E A5 | 148 |  | STA | DIVISOR.ML |  |
| B175:AD | OC A5 | 149 |  | IDA | SAMPIE.NUM |  |
| B178:8D | 4F A5 | 150 |  | STA | DIVISOR.I |  |
| B17B:20 | 9E B3 | 151 |  | JSR | DIVIDE. 40 |  |
| B17E:AD | 47 A5 | 152 |  | IDA | DIV.QUOT.MH | ; RETRIEVE QUOTIENT TERMS |
| B181:9D | 00 B8 | 153 |  | STA | SCRATCH, X | ; CORRESPONDING TO AVG |
| B184:AD | 48 A5 | 154 |  | IDA | DIV.QUOT.M | ;INTEGER PART OF RATIO |
| B187:9D | 01 B8 | 155 |  | STA | SCRATCH+1, X |  |
| B18A:AD | 49 A5 | 156 |  | IDA | DIV.QUOT.ML | ;RETRIEVE QUOTIENT TERMS |
| B18D:9D | 02 B8 | 157 |  | STA | SCRATCH $+2, \mathrm{X}$ | ; CORRESPONDING TO AVG |
| B190:AD | 4A A5 | 158 |  | IDA | DIV.QUOT.L | ; FRACTIONAL PART OF RATIO |
| B193:9D | 03 B8 | 159 |  | STA | SCRATCH $+3, \mathrm{X}$ |  |
| B196:60 |  | 160 |  | RTS |  |  |
| B197: |  | 161 | ; |  |  |  |
| $\begin{aligned} & \text { B197: } \\ & * * * * * \end{aligned}$ |  | 162 | ;******* | AVE | RAGE MEASUREM | NT COMPUTATIONS |
| B197: |  | 163 | ; |  |  |  |
| B197:20 | OF BO | 164 | SWEEP .AVG | JSR | INIT.SUMS | ; INITIALIZE SUM LOCATIONS |
| B19A:A0 | 00 | 165 |  | IDY | \#\$00 | ; ADDRESS INDEX |
| B19C:A9 | OA | 166 |  | IDA | \#10 |  |
| B19E:8D | 34 A5 | 167 |  | STA | SWEEP.CNT |  |
| BlAl:18 |  | 168 | NEXT.AVG | CLC |  |  |
| B1A2:AD | 57 A5 | 169 |  | IDA | CHO.SUM.L | ;ADD CHO SWEEP AVG TO |
| B1A5: 79 | 01 B8 | 170 |  | ADC | SCRATCH+1, | ; CURRENT CHO SWEEP SUM |
| B1A8:8D | 57 A5 | 171 |  | STA | CHO.SUM. 1 |  |
| BlAB:AD | 55 A5 | 172 |  | IDA | CHO.SUM. H |  |
| BLAE:79 | 00 B8 | 173 |  | ADC | SCRATCH, Y |  |
| B1B1:8D | 55 A5 | 174 |  | STA | CHO.SUM.H |  |
| B1B4:C8 |  | 175 |  | INY |  | ;UPDATE ADDRESS INDEX |
| B1B5:C8 |  | 176 |  | INY |  |  |
| B1B6:18 |  | 177 |  | CLC |  |  |
| B1B7:AD | 5A A5 | 178 |  | IDA | CH1.SUM.L | ;ADD CH1 SWEEP AVG TO |
| B1BA:79 | 01 B8 | 179 |  | ADC | SCRATCH $+1, Y$ | ; CURRENT CHI SWEEP SUM |
| B1BD:8D | 5A A5 | 180 |  | STA | CH1.SUM. 1 |  |
| B1C0:AD | 58 A5 | 181 |  | IDA | CHI.SUM. H |  |
| B1C3: 79 | 00 B8 | 182 |  | ADC | SCRATCH, Y |  |
| B1C6:8D | 58 A5 | 183 |  | STA | CH1.SUM.H |  |
| B1C9:C8 |  | 184 |  | INY |  | ;UPDATE ADDRESS INDEX |
| B1CA:C8 |  | 185 |  | INY |  |  |
| B1CB:18 |  | 186 |  | CLC |  |  |
| B1CC:AD | 5F A5 | 187 |  | IDA | RAT.SUM.L | ;ADD RATIO SWEEP AVG |
| B1CF: 79 | 03 B8 | 188 |  | ADC | SCRATCH+3,Y | ;TO CURRENT RATIO |
| B1D2:8D | 5F A5 | 189 |  | STA | RAT.SUM.L | : SWEEP SUM |
| B1D5:AD | 5E A5 | 190 |  | LDA | RAT.SUM.ML |  |
| B1D8:79 | 02 B8 | 191 |  | ADC | SCRATCH $+2, Y$ |  |


| B1DB:8D | 5E A5 | 192 | STA | RAT.SUM.ML |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| B1DE:AD | 5D A5 | 193 | LDA | RAT.SUM.M |  |
| B1E1:79 | 01 B8 | 194 | ADC | SCRATCE $+1, Y$ |  |
| B1E4:8D | 5D A5 | 195 | STA | RAT.SUM.M |  |
| B1E7:AD | 5C A5 | 196 | LDA | RAT.SUM.MH |  |
| B1EA: 79 | 00 B8 | 197 | ADC | SCRATCH, Y |  |
| B1ED:8D | 5C A5 | 198 | STA | RAT.SUM.MH |  |
| B1F0:AD | 5B A5 | 199 | LDA | RAT.SUM. H |  |
| B1F3:69 | 00 | 200 | ADC | \#00 | ;ALLOW FOR PRORAGATION |
| B1F5:8D | 5B A5 | 201 | STA | RAT.SUM. H |  |
| B1F8:C8 |  | 202 | INY |  |  |
| B1F9:C8 |  | 203 | INY |  | ;UPDATE ADDRESS INDEX |
| B1FA:C8 |  | 204 | INY |  |  |
| B1FB:C8 |  | 205 | INY |  |  |
| B1FC:CE | 34 A5 | 206 | DEC | SWEEP.CNT |  |
| B1FF:D0 | A0 B1A1 | 207 | BNE | NEXT.AVG | ; CHECK FOR 10 SWEEPS |
| B201:AD | 55 A5 | 208 | LDA | CHO.SUM. H | ;YES, LOAD DIVIDEND |
| B204:8D | 46 A5 | 209 | STA | DIV.QUOT.H | ;LOCATIONS WITH SUM OF |
| B207:AD | 57 A. 5 | 210 | LDA | CHO.SUM. | ;CHO AVERAGE VALUES |
| B20A:8D | 4A A5 | 211 | STA | DIV.QUOT.L |  |
| B20D:A9 | 00 | 212 | LDA | *\$00 | ;LOAD DIVISOR LOCATIONS |
| B20F:8D | 4B A5 | 213 | STA | DIVISOR. H | ;FOR TOTAL SWEEP COUNT |
| B212:A9 | OA | 214 | IDA | \#10 | ; EQUAL TO 10 |
| B214:8D | 4F A5 | 215 | STA | DIVISOR.L |  |
| B217:20 | 9C B2 | 216 | JSR | DIVIDE. 16 |  |
| B21A:AD | 46 A5 | 217 | IDA | DIV.QUOT.H | ;RETRIEVE QUOTIENT AND |
| B21D:99 | $00 \mathrm{B8}$ | 218 | STA | SCRATCH,Y | ; SAVE AS OVERALC CHO |
| B220:AD | 4A A5 | 219 | LDA | DIV.QUOT.L | ;AVERAGE VALUE |
| B223:99 | 01 B8 | 220 | STA | SCRATCH+1, Y |  |
| B226:C8 |  | 221 | INY |  | ;UPDATE ADDRESS INDEX |
| B227:C8 |  | 222 | INY |  |  |
| B228:AD | 58 A5 | 223 | LDA | CH1.SUM. H | ;IOAD DIVIDEND LOCATIONS |
| B22B:8D | 46 A5 | 224 | STA | DIV.QUOT.H | ;WITH SUM OF CHI AVERAGE |
| B22E:AD | 5A A5 | 225 | IDA | CH1.SUM. 工 | ; VALUES |
| B231:8D | 4A A. 5 | 226 | STA | DIV.QUOT.L |  |
| B234:A9 | 00 | 227 | LDA | \#\$00 | ;LOAD DIVISOR LOCATIONS |
| B236:8D | 4B A. 5 | 228 | STA | DIVISOR. H | ;FOR TOTAL SWEEP COUNT |
| B239:A9 | OA | 229 | LDA | \#10 | ; EQUAL TO 10 |
| B23B:8D | 4F A5 | 230 | STA | DIVISOR.L |  |
| B23E:20 | 9C B2 | 231 | JSR | DIVIDE. 16 |  |
| B241:AD | 46 A5 | 232 | LDA | DIV.QUOT.H | ; RETRIEVE QUOTIENT AND |
| B244:99 | $00 \mathrm{B8}$ | 233 | STA | SCRATCH, Y | ; SAVE AS OVERALI CHI |
| B247:AD | 4A A5 | 234 | LDA | DIV.QUOT.L | ; AVERAGE VALUE |
| B24A:99 | 01 B8 | 235 | STA | SCRATCH+1,Y |  |
| B24D:C8 |  | 236 | INY |  | ; UPDATE ADDRESS INDEX |
| B24E:C8 |  | 237 | INY |  |  |
| B24F:AD | 5B A5 | 238 | LDA | RAT.SUM. H | :LOAD DIVIDEND LOCATIONS |
| B252:8D | 46 A5 | 239 | STA | DIV.QUOT.H | ;WITH SUM OF AVERAGE |
| B255:AD | 5C A5 | 240 | LDA | RAT.SUM.M | : RATIOS |
| B258:8D | 47 A5 | 241 | STA | DIV.QUOT.MH |  |
| B25B:AD | 5D A5 | 242 | LDA | RAT.SUM.M |  |
| B25E:8D | 48 A5 | 243 | STA | DIV.QUOT.M |  |
| B261:AD | 5E A5 | 244 | LDA | RAT.SUM.ML |  |
| B264:8D | 49 A5 | 245 | STA | DIV.QUOT.ML |  |
| B267:AD | 5F A5 | 246 | LDA | RAT.SUM.L |  |
| B26A:8D | 4A A5 | 247 | STA | DIV.QUOT.L |  |
| B26D:A9 | 00 | 248 | LDA | \#\$00 | ;LOAD DIVISOR LOCATIONS |
| B26F:8D | 4 B A5 | 249 | STA | DIVISOR.H | ;FOR TOTAL SWEEP COUNT |


| B272:8D | 4C A5 | 250 |  | STA | DIVISOR.MA | : EQUAL TO 10 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| B275:8D | 4D A5 | 251 |  | STA | DIVISOR.M |  |
| B278:8D | 4E A5 | 252 |  | STA | DIVISOR.ML |  |
| B27B:A9 | OA | 253 |  | LDA | \#10 |  |
| B27D:8D | 4F A5 | 254 |  | STA | DIVISOR.L |  |
| B280:20 | 9E B3 | 255 |  | JSR | DIVIDE. 40 |  |
| B283:AD | 47 A5 | 256 |  | IDA | DIV.QUOT.MH | ;RETRIEVE QUOTIENT AS |
| B286:99 | $00 \mathrm{B8}$ | 257 |  | STA | SCRATCH, Y | ;OVERALC AVG INTEGER |
| B289:AD | 48 A5 | 258 |  | LDA | DIV.QUOT.M | ;PART OF RATIO |
| B28C:99 | 01 B8 | 259 |  | STA | SCRATCH+1, Y |  |
| B28F:AD | 49 A5 | 260 |  | IDA | DIV.QUOT.ML | :RETRIEVE QUOTIENT AS |
| B292:99 | 02 B8 | 261 |  | STA | SCRATCH+2, | : OVERALL AVG FRACTIONAL |
| B295:AD | 4A A5 | 262 |  | LDA | DIV.QUOT.L | ;PART OF RATIO |
| B298:99 | 03 B8 | 263 |  | STA | SCRATCH+3, $Y$ |  |
| B29B: 60 |  | 264 |  | RTS |  |  |
| B29C: |  | 265 | : |  |  |  |
| B29C: |  | 266 | ;********** |  | 16 BIT UNSIGNED | D DIVISION ********** |
| B29C: |  | 267 |  |  |  |  |
| B29C: 8E | 50 A5 | 268 | DIVIDE. 16 | STX | DIV.TEMPX | ; SAVE X REGISTER |
| B29F:A9 | 00 | 269 |  | IDA | *\$00 | ;CLEAR PARTIAL DIVIDEND |
| B2A1:8D | 41 A5 | 270 |  | STA | REMAIN. H |  |
| B2A4:8D | 45 A5 | 271 |  | STA | REMAIN.L |  |
| B2A7:A2 | 10 | 272 |  | LDX | \#16 | ;DIVIDEND BIT COUNT = 16 |
| B2A9:0E | 4A A5 | 273 | DNXTBT. 16 | ASL | DIV.QUOT.L | ;SHIFT DIVIDEND/QUOTIENT |
| B2AC:2E | 46 A5 | 274 |  | ROL | DIV.QUOT.H |  |
| B2AF:2E | 45 A5 | 275 |  | ROL | REMAIN.L | ;SHIET PARTIAL DIVIDEND |
| B2B2:2E | 41 A5 | 276 |  | ROL | REMAIN.H |  |
| B2B5:AD | 45 A5 | 277 |  | IDA | REMAIN.L |  |
| B2B8:38 |  | 278 |  | SEC |  | ;SUBTRACT LOW BYTES |
| B2B9:ED | 4F A5 | 279 |  | SBC | DIVISOR.L |  |
| B2BC:8D | 54 A5 | 280 |  | STA | DIV.TEMP.L | ; SAVE LOW BYTE RESULT |
| B2BF:AD | 41 A5 | 281 |  | LDA | REMAIN.H |  |
| B2C2:ED | 4B A5 | 282 |  | SBC | DIVISOR.H | ; SUBTRACT HIGH BYTES |
| B2C5:90 | OC B2D3 | 283 |  | BCC | CNTDN. 16 | ;DIVISOR > DIVIDEND ? |
| B2C7:EE | 4A A5 | 284 |  | INC | DIV.QUOT.L | ; NO, SET BIT IN QUOTIENT |
| B2CA:8D | 41 A5 | 285 |  | STA | REMAIN.H | ; AND ENTER SUBTRACT |
| RESULT |  |  |  |  |  |  |
| B2CD:AD | 54 A5 | 286 |  | IDA | DIV.TEMP.L | ; INTO PARTIAL DIVIDEND |
| B2D0:8D | 45 A5 | 287 |  | STA | REMAIN.L |  |
| B2D3:CA |  | 288 | CNITD .16 | DEX |  | ; DECREMENT BIT COUNT |
| B2D4:D0 | D3 B2A9 | 289 |  | BNE | DNXTBT. 16 | [LOOP UNTIL ALL 16 BITS |
| B2D6:AE | 50 A5 | 290 |  | IDX | DIV.TEMPX | ;RESTORE X REGISTER |
| B2D9:60 |  | 291 |  | RTS |  |  |
| B2DA: |  | 292 | ; |  |  |  |
| B2DA: |  | 293 | ;********** |  | 24 BIT UNSIGNED | DIVISION ********** |
| B2DA: |  | 294 | ; |  |  |  |
| B2DA: 8E | 50 A 5 | 295 | DIVIDE. 24 | STX | DIV.TEMPX | ; SAVE X REGISTER |
| B2DD:A9 | 00 | 296 |  | LDA | \#\$00 | ; CLEAR PARTIAL DIVIDEND |
| B2DF:8D | 41 A5 | 297 |  | STA | REMAIN.H |  |
| B2E2:8D | 43 A5 | 298 |  | STA | REMAIN.M |  |
| B2E5:8D | 45 A5 | 299 |  | STA | REMAIN.L |  |
| B2E8:A2 | 18 | 300 |  | LDX | \#24 | ;DIVIDEND BIT COUNT $=24$ |
| B2EA: $0 E$ | 4A A5 | 301 | DNXTET. 24 | ASL | DIV.QUOT.L | ;SHIFT DIVIDEND/QUOTIENT |
| B2ED: 2E | 48 A5 | 302 |  | ROL | DIV.QUOT.M |  |
| B2F0:2E | 46 A5 | 303 |  | ROL | DIV.QUOT.H |  |
| B2F3:2E | 45 A5 | 304 |  | ROL | REMAIN.L | ;SHIFT PARTIAL DIVIDEND |
| B2F6:2E | 43 A5 | 305 |  | ROL | REMAIN.M |  |
| B2F9:2E | 41 A5 | 306 |  | ROL | REMAIN.H |  |


| B2FC:AD | 45 A5 | 307 |  | LDA | REMAIN.L |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| B2FF: 38 |  | 308 |  | SEC |  | ;SUBTRACT LOW BYTES |
| B300: ED | 4F A5 | 309 |  | SBC | DIVISOR.I |  |
| B303:8D | 54 A5 | 310 |  | STA | DIV.TEMP.工 | ;SAVE LOW BYTE RESULT |
| B306:AD | 43 A5 | 311 |  | IDA | REMAIN.M |  |
| B309:ED | 4D A5 | 312 |  | SBC | DIVISOR.M | ;SUBTRACT MIDDIE BYTES |
| B30C:8D | 52 A5 | 313 |  | STA | DIV.TEMP.M | ;SAVE MIDDLE BYTE RESULT |
| B30F:AD | 41 A5 | 314 |  | IDA | REMAIN. H |  |
| B312:ED | 4B AS | 315 |  | SBC | DIVISOR.H | ;SUBTRACT HIGH BYTES |
| B315:90 | 12 B329 | 316 |  | BCC | CNITDN. 24 | :DIVISOR > DIVIDEND ? |
| B317:EE | 4A A5 | 317 |  | INC | DIV.QUOT.L | ;NO, SET BIT IN QUOTIENT |
| B31A:8D | 41 A5 | 318 |  | STA | REMAIN.H | ;AND ENTER SUBTRACT |
| RESUTT |  |  |  |  |  |  |
| B31D:AD | 52 A. | 319 |  | LDA | DIV.TEMP.M | :INTO PARTIAL DIVIDEND |
| B320:8D | 43 A5 | 320 |  | STA | REMAIN.M |  |
| B323:AD | 54 A5 | 321 |  | LDA | DIV.TEMP.I |  |
| B326:8D | 45 A5 | 322 |  | STA | REMAIN.L |  |
| B329:CA |  | 323 | CNIDN. 24 | DEX |  | ;DECREMENT BIT COUNT |
| B32A:D0 | BE B2EA | 324 |  | BNE | DNXTBT. 24 | ;LOOP UNTIL ALI 24 BITS |
| B32C:AE | 50 A5 | 325 |  | LDX | DIV.TEMPX | ;RESTORE X REGISTER |
| B32F:60 |  | 326 |  | RTS |  |  |
| B330: |  | 327 | ; |  |  |  |
| B330: |  | 328 | ; ********** |  | 32 BIT UNSIGNED | D DIVISION ********** |
| B330: |  | 329 | ; |  |  |  |
| B330:8E | 50 A5 | 330 | DIVIDE. 32 | STX | DIV.TEMPX | ; SAVE X REGISTER |
| B333:A9 | 00 | 331 |  | LDA | \#\$00 | ;CLEAR PARTIAL DIVIDEND * |
| B335:8D | 42 A5 | 332 |  | STA | REMAIN.ME |  |
| B338:8D | 43 A5 | 333 |  | STA | REMAIN.M |  |
| B33B:8D | 44 A5 | 334 |  | STA | REMAIN.ML |  |
| B33E:8D | 45 A5 | 335 |  | STA | REMAIN.L |  |
| B341:A2 | 20 | 336 |  | LDX | \#32 | ;DIVIDEND BIT COUNT $=32$ |
| B343:0E | 4A A5 | 337 | DNXTBT. 32 | ASL | DIV.QUOT.L | ;SHIFT DIVIDEND/QUOTIENT |
| B346:2E | 49 A5 | 338 |  | ROL | DIV.QUOT.ML |  |
| B349:2E | 48 A5 | 339 |  | ROL | DIV.QUOT.M |  |
| B34C: 2 E | 47 A5 | 340 |  | ROL | DIV.QUOT.MH |  |
| B34F:2E | 45 A5 | 341 |  | ROL | REMAIN.L | ; SHIET PARTIAL DIVIDEND |
| B352:2E | 44 A5 | 342 |  | ROL | REMAIN.ML |  |
| B355:2E | 43 A5 | 343 |  | ROL | REMAIN.M |  |
| B358:2E | 42 A5 | 344 |  | ROL | REMAIN.MH |  |
| B35B:AD | 45 A5 | 345 |  | LDA | REMAIN.L |  |
| B35E: 38 |  | 346 |  | SEC |  | ; SUBTRACT LOW BYTES |
| B35F:ED | 4F A5 | 347 |  | SBC | DIVISOR.L |  |
| B362:8D | 54 A5 | 348 |  | STA | DIV.TEMP.I | ; SAVE LOW BYTE RESULT |
| B365:AD | 44 A5 | 349 |  | LDA | REMAIN.ML |  |
| B368:ED | 4E A5 | 350 |  | SBC | DIVISOR.ML | ; SUBTRACT M/L BYTES |
| B36B:8D | 53 A5 | 351 |  | STA | DIV.TEMP.ML | ; SAVE M/L BYtE RESULT |
| B36E:AD | 43 A5 | 352 |  | LDA | REMAIN.M |  |
| B371: ED | 4D A5 | 353 |  | SBC | DIVISOR.M | ; SUBTRACT MIDDLE BYTES |
| B374:8D | 52 A5 | 354 |  | STA | DIV.TEMP.M | ; SAVE MIDDLE BYTE RESULT |
| B377:AD | 42 A5 | 355 |  | LDA | REMAIN.MH |  |
| B37A:ED | 4C A5 | 356 |  | SBC | DIVISOR.MH | ; SUBTRACT M/H BYTES |
| B37D:90 | 18 B397 | 357 |  | BCC | CNTDN. 32 | ;DIVISOR > DIVIDEND ? |
| B37F:EE | 4A A5 | 358 |  | INC | DIV.QUOT.L | ; NO, SET BIT IN QUOTIENT |
| B382:8D | 42 A5 | 359 |  | STA | REMAIN.MH | ; AND ENTER SUBTRACT |
| RESULT |  |  |  |  |  |  |
| B385:AD | 52 A5 | 360 |  | IDA | DIV.TEMP.M | ; INTO PARTIAL DIVIDEND |
| B388:8D | 43 A5 | 361 |  | STA | REMAIN.M |  |
| B38B:AD | 53 A5 | 362 |  | LDA | DIV.TEMP.ML |  |


| B38E: 8D | 44 A5 | 363 |  | STA | REMAIN.ML |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| B391:AD | 54 A5 | 364 |  | IDA | DIV.TEMP.L |  |
| B394:8D | 45 A5 | 365 |  | STA | REMAIN.L |  |
| B397:CA |  | 366 | CNTDN. 32 | DEX |  | ;DECREMENT BIT COUNT |
| B398:D0 | A9 B343 | 367 |  | BNE | DNXTBT. 32 | ;LOOR UNTIL ALL 32 BITS |
| B39A:AE | 50 A 5 | 368 |  | LDX | DIV.TEMPX | ;RESTORE X REGISTER |
| B39D: 60 |  | 369 |  | RTS |  |  |
| B39E: |  | 370 | ; |  |  |  |
| B39E: |  | 371 | ; $\begin{aligned} & \text { t******** }\end{aligned}$ |  | 40 BIT UNSIGNE | D DIVISION ********** |
| B39E: |  | 372 | ; |  |  |  |
| B39E:8E | 50 A 5 | 373 | DIVIDE. 40 | STX | DIV.TEMPX | ; SAVE X REGISTER |
| B3A1:A9 | 00 | 374 |  | LDA | \#\$00 | ;CLEAR PARTIAL DIVIDEND |
| B3A3:8D | 41 A5 | 375 |  | STA | REMAIN.H |  |
| B3A6:8D | 42 A5 | 376 |  | STA | REMAIN.MH |  |
| B3A9:8D | 43 A5 | 377 |  | STA | REMAIN.M |  |
| B3AC: 8D | 44 A5 | 378 |  | STA | REMAIN.ML |  |
| B3AF:8D | 45 A5 | 379 |  | STA | REMAIN.L |  |
| B3B2:A2 | 28 | 380 |  | LDX | \# 40 | ;DIVIDEND BIT COUNT $=40$ |
| B3B4:0E | 4A A5 | 381 | DNXTBT. 40 | ASL | DIV.QUOT.L | ;SHIFT DIVIDEND/QUOTIENT |
| B3B7:2E | 49 A5 | 382 |  | ROL | DIV.QUOT.ML |  |
| B3BA:2E | 48 A5 | 383 |  | ROL | DIV.QUOT.M |  |
| B3BD:2E | 47 A5 | 384 |  | ROL | DIV.QUOT.MH |  |
| B3C0:2E | 46 A5 | 385 |  | ROL | DIV.QUOT.H |  |
| B3C3:2E | 45 A5 | 386 |  | ROL | REMAIN.L | ;SHIFT PARTIAL DIVIDEND |
| B3C6:2E | 44 A5 | 387 |  | ROL | REMAIN.ML |  |
| B3C9:2E | 43 A5 | 388 |  | ROL | REMAIN.M |  |
| B3CC:2E | 42 A 5 | 389 |  | ROL | REMAIN.MH |  |
| B3CF:2E | 41 A5 | 390 |  | ROL | REMAIN. H |  |
| B3D2:AD | 45 A5 | 391 |  | LDA | REMAIN.L |  |
| B3D5:38 |  | 392 |  | SEC |  | ; SUBTRACT LOW BYTES |
| B3D6:ED | 4F A5 | 393 |  | SBC | DIVISOR.L |  |
| B3D9:8D | 54 A5 | 394 |  | STA | DIV.TEMP.L | :SAVE LOW BYTE RESULT |
| B3DC:AD | 44 A5 | 395 |  | LDA | REMAIN.ML |  |
| B3DF:ED | 4E A5 | 396 |  | SBC | DIVISOR.ML | ; SUBTRACT M/L BYTES |
| B3E2:8D | 53 A5 | 397 |  | STA | DIV.TEMP.ML | ; SAVE M/L BYTE RESULT |
| B3E5:AD | 43 A5 | 398 |  | LDA | REMAIN.M |  |
| B3E8:ED | 4D A5 | 399 |  | SBC | DIVISOR.M | ;SUBTRACT MIDDLE BYTES |
| B3EB:8D | 52 A5 | 400 |  | STA | DIV.TEMP.M | ;SAVE MIDDLE BYTE RESULT |
| B3EE:AD | 42 A 5 | 401 |  | LDA | REMAIN.MH |  |
| B3F1:ED | 4C A5 | 402 |  | SBC | DIVISOR.MH | ; SUBTRACT M/H BYTES |
| B3F4:8D | 51 A5 | 403 |  | STA | DIV.TEMP.MH | ;SAVE M/H BYTE RESUTT |
| B3F7:AD | 41 A5 | 404 |  | LDA | REMAIN.H |  |
| B3FA:ED | 4B A5 | 405 |  | SBC | DIVISOR.H | ; SUBTRACT HIGH BYTES |
| B3FD:90 | 1E B41D | 406 |  | BCC | CNTDN. 40 | ;DIVISOR > DIVIDEND ? |
| B3FF:EE | 4A A5 | 407 |  | INC | DIV.QUOT.L | ; NO, SET BIT IN QUOTIENT |
| B402:8D | 41 A5 | 408 |  | STA | REMAIN.H | ;AND ENTER SUBTRACT |
| RESULT |  |  |  |  |  |  |
| B405:AD | 51 A5 | 409 |  | LDA | DIV.TEMP.MH | ; INIO PARTIAL DIVIDEND |
| B408:8D | 42 A5 | 410 |  | STA | REMAIN.MH |  |
| B40B:AD | 52 A5 | 411 |  | LDA | DIV.TEMP.M |  |
| B40E:8D | 43 A5 | 412 |  | STA | REMAIN.M |  |
| B411:AD | 53 A5 | 413 |  | LDA | DIV.TEMP.ML |  |
| B414:8D | 44 A5 | 414 |  | STA | REMAIN.ML |  |
| B417:AD | 54 A5 | 415 |  | LDA | DIV.TEMP.L |  |
| B41A:8D | 45 A5 | 416 |  | STA | REMAIN.L |  |
| B41D:CA |  | 417 | CNTDN. 40 | DEX |  | ; DECREMENT BIT COUNT |
| B41E:D0 | 94 B3B4 | 418 |  | BNE | DNXTBT. 40 | ; LOOP UNTIL ALL 40 BITS |
| B420:AE | 50 A5 | 419 |  | LDX | DIV.TEMPX | ; RESTORE X REGISTER |


| B423:60 |  | 420 |  | RIS |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| B424: |  | 421 | ; |  |  |  |
| B424: |  | 422 | ;********** |  | SOFTWARE DELAY | Y ROUTINE \#1 ********** |
| B424: |  | 423 | , |  |  |  |
| B424:8E | 61 A5 | 424 | SW. DELAY1 | STX | DLY.TEMPX | ;SAVE X \& Y REGISTERS |
| B427:8C | 62 A5 | 425 |  | STY | DIY.TEMPY |  |
| B42A:AD | 63 A5 | 426 |  | LDA | DELAY | ;LOAD TIMING BYTE |
| B42D:A2 | A5 | 427 | GEN. DLY1 | LDX | *\$A5 | ; IOAD X AND Y FOR A |
| B42F:A0 | EA | 428 |  | LDY | *SEA | ; 300 MSEC TIME DELAY |
| B431:CA |  | 429 | WRIT.DIY1 | DEX |  |  |
| B432:D0 | ED B431 | 430 |  | BNE | WAIT.DLY1 | ;LOOP UNTIL $\mathrm{X}=0$ |
| B434:88 |  | 431 |  | DEY |  |  |
| B435:D0 | FA B431 | 432 |  | BNE | WAIT.DLYI | ; LOOP UNTIL $\mathrm{X}=\mathrm{Y}=0$ |
| B437:38 |  | 433 |  | SEC |  |  |
| B438:E9 | 01 | 434 |  | SBC | \#01 | ;DECREMENT TIMING BYTE |
| B43A:D0 | F1 B42D | 435 |  | BNE | GEN.DLY1 | ;LOOP UNTIL ACC=0 |
| B43C:AE | 61 A5 | 436 |  | IDX | DLY.TEMPX | ;RESTORE X \& Y REGISTERS |
| B43F:AC | 62 A5 | 437 |  | IDY | DLY.TEMPY |  |
| B442:60 |  | 438 |  | RTS |  |  |
| B443: |  | 439 | ; |  |  |  |
| B443: |  | 440 | ;********** |  | SOFTWARE DELAY | Y ROUTINE 2 ( 2 ********** |
| B443: |  | 441 | , |  |  |  |
| B443:8E | 61 A5 | 442 | SW.DELAY2 | STX | DLY.TEMPX | ; SAVE X REGISTER |
| B446:AD | 63 A5 | 443 |  | IDA | DELAY | ; LOAD TIMING BYTE |
| B449:A2 | 12 | 444 | GEN. DLY2 | IDX | \#\$12 | ;LOAD X FOR 100 USEC |
| B44B:CA |  | 445 | WAIT.DLY2 | DEX |  | ;TIME DELAY |
| B44C:D0 | FD B44B | 446 |  | BNE | WAIT.DLY2 | ;LOOP UNTII $\mathrm{X}=0$ |
| B44E:EA |  | 447 |  | NOP |  |  |
| B44F:38 |  | 448 |  | SEC |  |  |
| B450:E9 | 01 | 449 |  | SBC | \#01 | :DECREMENT TIMING BYTE |
| B452:D0 | F5 B449 | 450 |  | BNE | GEN. DLY2 | ;LOOP UNTIL ACC $=0$ |
| B454:AE | 61 A5 | 451 |  | LDX | DLY. TEMPX | ;RESTORE X REGISTER |
| B457:60 |  | 452 |  | RTS |  |  |
| B458: |  | 453 | ; |  |  |  |
| B458: |  | 454 | ; *********** |  | OUTPUT SETITI | ING ROUTINE ********** |
| B458: |  | 455 |  |  |  |  |
| B458:A9 | 05 | 456 | OUT. SETTLE | LDA | \#05 | ;SET FOR 1.5 SEC |
| B45A: 8D | 63 A5 | 457 |  | STA | DELAY |  |
| B45D: 20 | 24 B4 | 458 |  | JSR | SW.DELAY1 | ; OUTPUT SETTLING TIME |
| B460:A9 | 7 F | 459 |  | IDA | \#\%01111111 | :CLEAR ANY INTERRUPTS |
| B462:8D | 8D C2 | 460 |  | STA | U2.IFR | ;BEFORE RETURNING |
| B465:60 |  | 461 |  | RTS |  |  |
| B466: |  | 462 | ; |  |  |  |
| B466: |  | 463 | ;****** | A/D | MULTIPLEXOR SE | TTIING ROUTINE ****** |
| B466: |  | 464 | ; |  |  |  |
| B466:A9 | 01 | 465 | MUX.SETTLE | LDA | \#01 | ; SET FOR 100 USEC |
| B468:8D | 63 A5 | 466 |  | STA | DELAY |  |
| B46B: 20 | 43 B4 | 467 |  | JSR | SW.DELAY2 |  |
| B46E: 60 |  | 468 |  | RTS |  |  |
| B46F: |  | 469 | ; |  |  |  |
| B46F: |  | 470 |  |  | A/D CONVERSIO | N ROUTINE *********** |
| B46F: |  | 471 | . |  |  |  |
| B46F:AD | D2 C0 | 472 | CONVERT | LDA | START.CONV | ; START CONVERSION |
| B472:AD | D1 C0 | 473 | WAIT. CONV | IDA | AD. HIGH | :CHECK FOR CONVERSION |
| B475:2A |  | 474 |  | ROL |  | ; STATUS BIT SET |
| B476:2A |  | 475 |  | ROL |  |  |
| B477:B0 | F9 B472 | 476 |  | BCS | WAIT.CONV | ; RETURN WHEN CONVERSION |
| B479:60 |  | 477 |  | RTS |  | ; COMPLETED |



| B4E7:AD 65 A5 | 533 |  | LDA | CLK. CNTRRL. B |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| B4EA:8D C3 CO | 534 |  | STA | CRB.CLK |  |
| B4ED:AD CO CO | 535 |  | LDA | DRA.CLK | ;CLEAR PREVIOUS INTERRUPT |
| FLAGS |  |  |  |  |  |
| B4F0:AD C2 C0 | 536 |  | LDA | DRB.CLK |  |
| B4F3:A9 2F | 537 |  | IDA | \#\$2F | ; ENABLE INTERRUPTS OUT OE |
| PIA |  |  |  |  |  |
| B4F5:8D C2 C0 | 538 |  | STA | DRB.CLK |  |
| B4F8:58 | 539 |  | CLI |  | ;ENABLE PROCESSOR |
| INTERRUPTS |  |  |  |  |  |
| B4F9:60 | 540 |  | RTS |  |  |
| B4FA: | 541 | ; |  |  |  |
| B4FA: | 542 | ;********* | CLOCK | STOP ROUTINE | E ********* |
| B4FA: | 543 | : |  |  |  |
| B4FA: 78 | 544 | CLIESTOP | SEI |  | ;DISABLE PROCESSOR |
| INTERRUPTS |  |  |  |  |  |
| B4FB:A9 04 | 545 |  | LDA | \#\$04 | ;DISABLE CLOCK INTERRUPTS |
| B4FD:8D 64 A5 | 546 |  | STA | CLK. CNTRL.A |  |
| B500:8D 65 A5 | 547 |  | STA | CLK. CNTRL. B |  |
| B503:AD 64 A5 | 548 |  | LDA | CLK. CNIRL.A |  |
| B506:8D C1 C0 | 549 |  | STA | CRA. CLK |  |
| B509:AD 65 A5 | 550 |  | LDA | CLK. CNTRL. B |  |
| B50C:8D C3 C0 | 551 |  | STA | CRB.CLK |  |
| B50F:AD CO CO | 552 |  | LDA | DRA.CLK | ;CLEAR PREVIOUS CLOCK |
| INTERRUPT FLAGS |  |  |  |  |  |
| B512:AD C2 C0 | 553 |  | LDA | DRB.CLK |  |
| B515:60 | 554 |  | RTS |  |  |
| B516: | 555 | ; |  |  |  |
| B516: | 556 | ;********* | ENABLE | E PROCESSOR | INTERRUPT ROUTINE ********* |
| B516: | 557 | ; |  |  |  |
| B516:58 | 558 | ENABLE | CLI |  |  |
| B517:60 | 559 |  | RTS |  |  |
| B518: | 560 | ; |  |  |  |
| B518: | 561 | ;********* | INTERRUPT SERVICE ROUTINE ********** |  |  |
| B518: | 562 | ; |  |  |  |
| B518:A5 45 | 563 | ISR | LDA | \$45 | ; RESTORE ACCUMULATOR |
| B51A: 48 | 564 |  | PHA |  | ;PUSH ACCUMULATOR |
| B51B:8A | 565 |  | TXA |  | ;PUSH X REGISTER |
| B51C: 48 | 566 |  | PHA |  |  |
| B51D: 98 | 567 |  | TYA |  | ;PUSH Y REGISTER |
| B51E:48 | 568 |  | PHA |  |  |
| B51F:AD 8D C2 | 569 |  | LDA | U2.IFR | ;INSPECT INTERRUPTS |
| B522:2D 8E C2 | 570 |  | AND | U2.IER |  |
| B525:2A | 571 |  | ROL |  |  |
| B526:2A | 572 |  | ROL |  |  |
| B527:B0 22 B54B | 573 |  | BCS | TIMER1 | ; CHECK FOR T1 IRQ |
| B529:AD CI C0 | 574 |  | LDA | CRA.CLK | ; CHECK FOR IRQ FROM CLOCK |
| PORT A |  |  |  |  |  |
| B52C:29 40 | 575 |  | AND | \#801000000 |  |
| B52E:2A | 576 |  | ROL |  |  |
| B52F:2A | 577 |  | ROL |  |  |
| B530:B0 16 B548 | 578 |  | BCS | CLOCK | ; IF YES THEN SERVICE CLOCK |
| INTERRUPT |  |  |  |  |  |
| B532:AD C3 C0 | 579 |  | LDA | CRB.CLK | ; CHECK FOR IRQ FROM CLOCK |
| PORT B |  |  |  |  |  |
| B535:29 C0 | 580 |  | AND | \#\%11000000 |  |
| B537:2A | 581 |  | ROL |  |  |



| B5BE:DO FD | B5BD | 637 |  | BNE | INT. DATA |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| B5C0:A9 30 |  | 638 |  | LDA | \#\%00110000 | ; CEASE INTEGRATION |
| B5C2:8D 81 | C2 | 639 |  | STA | U2.DRA |  |
| B5C5:A9 40 |  | 640 |  | LDA | *\%01000000 | ; CLEAR IRQ FLAG |
| B5C7:8D 8D | C2 | 641 |  | STA | U2.IFR |  |
| B5CA:CE OB | A5 | 642 |  | DEC | INT.CNT | ; INTEGRATION COUNT |
| B5CD:DO 30 | B5FF | 643 |  | BNE | RETURN |  |
| B5CF:A9 20 |  | 644 | INT. FINISH | LDA | * $\% 00100000$ | ;PLACE S/H IN "HOLD" |
| B5D1:8D 81 | C2 | 645 |  | STA | U2.DRA |  |
| B5D4:A9 01 |  | 646 |  | LDA | \# 01 | ;S/H CIRCUITRY MUST |
| B5D6:8D 63 | A5 | 647 |  | STA | DELAY | ;SETTLE |
| B5D9:20 43 | B4 | 648 |  | JSR | SW.DELAY2 |  |
| B5DC:A9 21 |  | 649 |  | LDA | \#\%00100001 | ;DISCHARGE INTEGRATORS |
| B5DE:8D 81 | C2 | 650 |  | STA | U2.DRA |  |
| B5E1:A9 01 |  | 651 |  | LDA | ${ }^{\text {\# }} 01$ | ; SET SAMPLE AVAILABLE |
| B5E3:8D 35 | A5 | 652 |  | STA | SMPL.AVAII | ; FLAG |
| B5E6:4C FF | B5 | 653 |  | JMP | RETURN |  |
| B5E9:A9 80 |  | 654 | SKIP.DATA | IDA | \#\%10000000 | ;NO, RISING TRANSITION |
| B5EB:8D 80 | C2 | 655 |  | STA | U2.DRB | ;FOR FLASH TRIGGER |
| B5EE:A9 40 |  | 656 |  | LDA | \#\%01000000 | ;CLEAR IRQ FLAG |
| B5F0:8D 8D | C2 | 657 |  | STA | U2.IFR |  |
| B5F3:4C FF | B5 | 658 |  | JMP | RETURN |  |
| B5F6:EE 66 | A5 | 659 | CLK. SERV | INC | TIME.OUT | ; INCREMENT TIME.OUT COUNT |
| FOR BASIC |  |  |  |  |  |  |
| B5F9:AD C0 | CO | 660 |  | IDA | DRA.CLK | ; CLEAR PREVIOUS INTERRUPT |
| FLAGS |  |  |  |  |  |  |
| B5FC:AD C2 | CO | 661 |  | IDA | DRB.CLK |  |
| B5FF: 68 |  | 662 | RETURN | PLA |  | ;PULL Y REGISTER |
| B600:A8 |  | 663 |  | TAY |  |  |
| B601:68 |  | 664 |  | PLA |  | ;PULL X REGISTER |
| B602:AA |  | 665 |  | TAX |  |  |
| B603:68 |  | 666 |  | PLA |  | ;PULJ ACCUMULATOR |
| B604: 40 |  | 667 |  | RTI |  | ;RETURN FROM IRQ |


| A513 | AD.CEO.CTRL | A5 | CH1.CTRS | COD3 | TRL | 1 | G |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| CODO | AD.LOW | A843 | AUTO | A87C | AUTO. CALO | A93E | AUTO.CAL1 |
| B0E9 | AVG.CEO | B119 | AVG.CH1 | AB03 | AVG. INJO | AB32 | AVG.INJ1 |
| B149 | AVG.RATIO | A510 | CHO.AD.GAIN | A515 | CEO.INJ.H | A516 | CHO.INJ.L |
| A50F | CHO.PGAIN | A555 | CHO.SUM. B | A557 | CHO.SUM.L | A556 | CHO.SUM.M |
| A512 | CHI.AD.GAIN | A517 | CHI. INJ. H | A518 | CHI.INJ.L | A511 | CHI.PGAIN |
| A558 | CH1.SUM. ${ }^{\text {C }}$ | A55A | CH1.SUM.L | A559 | CH1.SUM.M | AB79 | CHOICE. 1 |
| AB7C | CHOICE. 2 | AB7F | CHOICE. 3 | A564 | CLIK. CNIRL.A | A565 | CLK. CNIRL. ${ }^{\text {B }}$ |
| B5F6 | CLK.SERV | A567 | CLKSTART.H | A568 | CILKSTART.L | B4E1 | CTKSTART |
| A569 | CLKSTOP. H | B4FA | CLIKSTOP | A56A | CIKSTOP.L | B548 | CLOCK |
| B2D3 | CNTDN. 16 | B329 | CNIDN. 24 | B397 | CNTDN. 32 | B41D | CNIDN. 40 |
| AD1D | COMP.MS | A658 | CONEIG | A8B7 | CONT. CALO | A979 | CONT. CALI |
| AE8B | CONT.CHK. 0 | AE9F | CONT.CHK. 1 | AE77 | CONT.CHK.SS | B564 | CONT.MEAS |
| AFOB | CONT.Z.CHO | AF21 | CONT.Z.CHI | AEF5 | CONT.Z.SS | B46F | CONVERT |
| FDED | COUT | C0C1 | CRA.CLX | COC3 | CRB.CLK | A51B | DA.COPY1 |
| A51C | DA. COPY2 | A51D | DA.COPY3 | A51E | DA.COPY4 | A51F | DA.COPY5 |
| A802 | DA.CTRL | COB8 | DA.CTRL1 | C0B9 | DA.CTRL2 | COBA | DA.CTRL3 |
| COBB | DA.CTRL4 | COBC | DA.CIRL5 | A539 | DATA.ACTIVE | B5AB | DATA.CHK |
| AD71 | DATA.FILE | A53B | DATA.SETTLE | A563 | DELAY | A546 | DIV.QUOT.H |
| A54A | DIV.QUOT.L | A548 | DIV.QUOT.M | A547 | DIV.QUOT.MH | A549 | DIV.QUOT.ML |
| A554 | DIV.TEMP.I | A551 | DIV.TEMP.MH | A552 | DIV.TEMP.M | A553 | DIV.TEMP.ML |
| A550 | DIV.TEMPX | .B29C | DIVIDE. 16 | B2DA | DIVIDE. 24 | B330 | DIVIDE. 32 |
| B39E | DIVIDE. 40 | A54B | DIVISOR. ${ }^{\text {a }}$ | A54F | DIVISOR.L | A54D | DIVISOR.M |
| A54C | DIVISOR.MH | A54E | DIVISOR.MU | A561 | DLY.TEMPX | A562 | DLY.TEMPY |
| B2A9 | DNXTBT. 16 | B2EA | DNXTBT. 24 | B343 | DNXTET. 32 | B384 | DNXTBT. 40 |
| COCO | DRA.CLK | COC2 | DRB.CLK | A56D | DUMMY.FLASH | B516 | ENABLE |
| A56B | ENABLE.H | A56C | ENABLE.L | B47A | ERROR | A914 | EXIT.CALO |
| A9D6 | EXIT.CAL1 | A92A | FINAL.CALO | A9EC | FINAL.CAII | ADEO | FINISH.DF |
| AABE | FINISH.DK | AD60 | FINISH.MS | AC4A | FINISH.SS | A508 | FLASH.F |
| A53C | FLASH. H | AA1A | FLASH. INJ | A53D | FLASH.I | B42D | GEN.DLY1 |
| B449 | GEN.DLY2 | B00F | INIT. SUMS | ADEE | INIT.SWEEP | A536 | INJ.ACTIVE |
| B577 | INJ.CHK | A537 | INJ.STATUS | A821 | INT.CAL | A50B | INT.CNT |
| B5BD | INT. DATA | B5CF | INT.FINISH | B589 | INT.INJ | A53E | INT.LOOP |
| A50A | INT.NUM | A509 | INT.PERIOD | A560 | INVALID | B518 | ISR |
| AC5B | M. SWEEP | A858 | MAN. CAL | A50E | MANUAL | A83F | MAN |
| A538 | MEAS.ACTIVE | AA4B | MEAS. INJ | A519 | MEAS.MODE | A53F | MEAS. TEMP. H |
| A540 | MEAS. TEMP.L | B466 | MUX. SETTIE | B1A1 | NEXT.AVG | B4B5 | NEXT1 |
| B4C0 | NEXT2 | B4CB | NEXT3 | B4D6 | NEXT4 | AE83 | OF.CHK.CHO |
| AE97 | OF.CHK.CHI | AE6F | OF.CHK.SS | AF99 | OF.ERR.SW10 | AF63 | OF.ERR.SWI |
| AF69 | OF.ERR.SW2 | AF6F | OF.ERR.SW3 | AF75 | OF.ERR.SW4 | AF7B | OF.ERR.SW5 |
| AF81 | OF.ERR.SW6 | AF87 | OF.ERR.SW7 | AF'8D | OF.ERR.SW8 | AF93 | OF.ERR.SW9 |
| A520 | OF.SW1 | A529 | OF.SW10 | A521 | OF.SW2 | A522 | OF.SW3 |
| A523 | OF.SW4 | A524 | OF.SW5 | A525 | OF.SW6 | A526 | OF.SW7 |
| A527 | OF.SW8 | A528 | OF.SW9 | A51A | OFFSET.ADJ | B458 | OUT.SETTLE |
| AE93 | OVER.CHO | AEA7 | OVER.CH1 | AE7F | OVER.SS | AF2F | OVERFLOW |
| A8E5 | PGAIN.CALO | A9A7 | PGAIN. CALI | A8DA | PGAIN.FIXO | A99C | PGAIN.FIXI |
| A501 | PHASE1.L | A6B4 | PHASE1 | A500 | PHASE1.H | A82D | PHASE2 |
| A502 | PHASE2.H | A503 | PHASE2.L | AA00 | Phase3 | A504 | PHASE3. H |
| A505 | PHASE3.L | A506 | PHASE4.H | A507 | PHASE4.工 | AB61 | PHASE4 |
| AABF | POST.INJ | AE60 | POST.MEAS | AA2A | PRE.INJ | AE2D | PRE.MEAS |
| A892 | RANGE.CALO | A954 | RANGE.CAL1 | BEOO | RAT.FRAC | BD00 | RAT.INT |
| A55B | RAT.SUM. H | A55F | RAT.SUM.L | A55C | RAT. SUM.MH | A55D | RAT. SUM.M |
| A55E | RAT. SUM.ML | B077 | RATIO.MS | B033 | RATIO.SS | A541 | REMAIN. H |
| A545 | REMAIN.L | A543 | REMAIN.M | A.542 | REMAIN.MH | A544 | REMAIN.ML |
| B5FF | RETURN | AB82 | S.SWEEP | A50D | SAMPLE.CNT | A50C | SAMPLE. NUM |
| B900 | SCRATCHO | B800 | SCRATCH | BA00 | SCRATCHO.A | BB00 | SCRATCH1 |
| BCOO | SCRATCH1.A | B5E9 | SKIP. DATA | B59E | SKIP.INJ | AC23 | SKIP.RATIO |
| ABDC | SKIP.Z.CHO | AC19 | SKIP.2.CHI | A535 | SMPL.AVAIL | COD2 | START.CONV |


| AF13 | SUBST. CHO | AF29 | SUBST.CH1 | AEFD | SUBST.SS | AACB | SUM. CHO |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| AAE7 | SUM.CH1 | B0BB | SUM. RATIO | B424 | SW.DELAY1 | B443 | SW. DELAY2 |
| B197 | SWEEP.AVG | A534 | SWEEP.CNI | AA40 | SYNC. INJ | AE4B | SINC. MEAS |
| A53A | SYNC.OK | A7CB | SYS.INIT | A6C4 | TABLE | A566 | TIME.OUT |
| B54B | TIMER1 | A7B5 | TIMER | A689 | TMASTER | C20B | U1.ACR |
| C203 | U1.DDRA | C202 | U1.DDRB | C201 | U1.DRA | ? 2200 | U1.DRB |
| C20E | U1.IER | C20D | U1.IFR | C20C | U1.PCR | ?C205 | U1.T1C.H |
| ? 2004 | U1.T1C.I | ?C207 | U1.T1L.H | ?C206 | U1.TIL.L | ?C209 | U1.T2C.H |
| ?C208 | U1.T2C.L | C28B | U2.ACR | C283 | U2.DDRA | C282 | U2.DDRB |
| C281 | U2.DRA | C280 | U2.DRB | C28E | U2.IER | C28D | U2.IFR |
| C28C | U2.PCR | C285 | U2.T1C.H | C284 | U2.T1C.L | ? 2287 | U2.T1L.H |
| ? 2286 | U2.T1L.L | ?C289 | U2.T2C.H | ?C288 | U2.T2C.L | AEC1 | UF.CHK.CHO |
| AED7 | UF.CHK.CH1 | AEAB | UF.CHK.SS | AFD3 | UF.ERR.SW1 | B009 | UF.ERR.SW10 |
| AFD9 | UF.ERR.SW2 | AFDF | UF.ERR.SW3 | AFES | UF.ERR.SW4 | AFEB | UF.ERR.SW5 |
| AFF1 | UF.ERR.SW6 | AFF7 | UF.ERR.SW7 | AFFD | UF.ERR.SW8 | B003 | UF.ERR.SW9 |
| A52A | UF.SW1 | A533 | UF.SW10 | A52B | UF.SW2 | A52C | UF.SW3 |
| A52D | UF.SW4 | A52E | UF.SW5 | A52F | UF.SW6 | A530 | UF.SW7 |
| A531 | UF.SW8 | A532 | UF.SW9 | AEC9 | UNDER.CHO | AEDF | UNDER.CHI |
| AEB3 | UNDER.SS | AF9F | UNDERFLOW | B472 | WAIT.CONV | AD91 | WAIT.DF |
| B431 | WAIT.DLYI | B44B | WAIT.DLY2 | AA68 | WAIT.INJ | AC89 | WAIT.MS |
| ABA5 | WAIT.SS | A89F | WAIT1. CALO | A961 | WAIT1.CAL1 | AF03 | 2ERO.CHO |
| AF19 | ZERO.CHI | AEED | 2ERO.SS |  |  |  |  |
| ** SUCCESSFUL ASSEMBLY := NO ERRORS |  |  |  |  |  |  |  |
| ** ASSEMBLER CREATED ON 21-MAY-83 REL-07 |  |  |  |  |  |  |  |
| ** TOTAL LINES ASSEMBLED 2033 |  |  |  |  |  |  |  |
| ** FREE SPACE PAGE COUNT 62 |  |  |  |  |  |  |  |
| 2 JMC1.1 |  |  |  |  |  |  |  |
| 3 JMC1. 2 |  |  |  |  |  |  |  |

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## VITA

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